

Max-Planck-Institute for Solid State Research



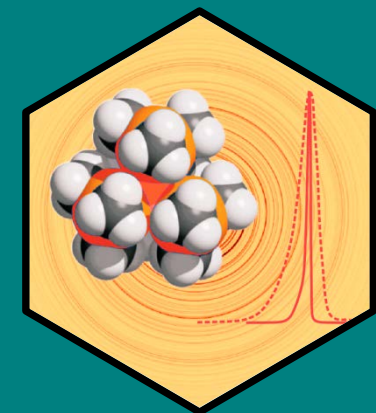
NEW METHODS FOR INCREASING ACCURACY OF *IN-SITU* POWDER DIFFRACTION

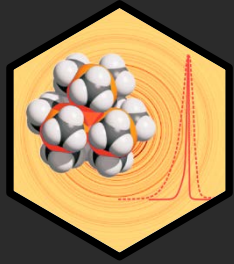
APD-IV

NIST

Gaithersburg, April 22-25, 2013

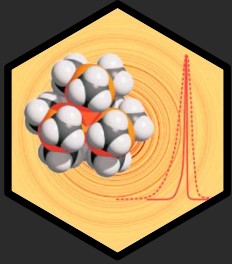
Robert E. Dinnebier
Martin Etter
Oksana Magdysyuk
Melanie Müller
Tomče Runčevski





OUTLINE

- The method of *in-situ* powder diffraction
- „bottle necks“ and possible solutions
- State of the art examples:
 - Monitoring of mechanochemical milling reactions
 - Gas adsorption in large pore MOF's
 - Advances in MEM analysis
- Conclusions



THE METHOD OF CHOICE: TIME RESOLVED '*IN SITU*' POWDER DIFFRACTION

Phase transitions

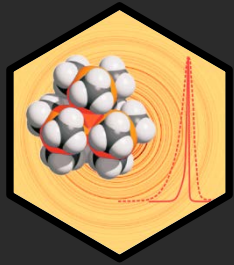
- ‚Polymorph screening’
- Phase transitions of different order
- Reversibility of phase transitions (hysteresis, strain-order parameter coupling)

Chemical reactions

- solid/solid, liquid/liquid, solid/gaseous etc.
- Influence of external conditions (pressure, temperature, stress, Electric/magnetic field, Light, mechanical power, etc.)

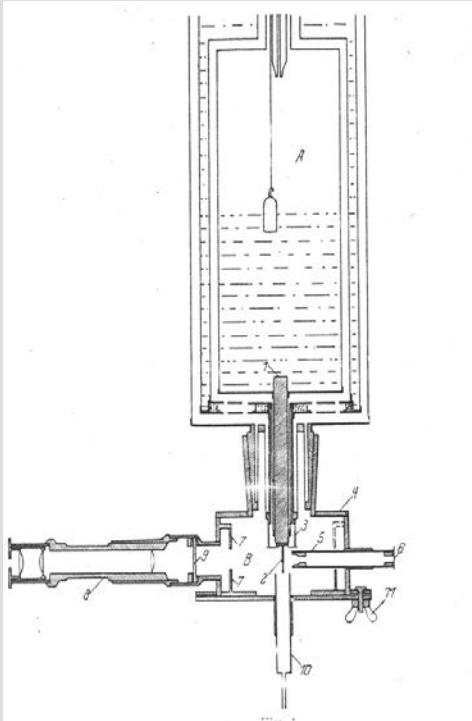
Physical effects

- thermal expansion (change of bond lengths, cell volume, cell axes etc.)
- Compressibility (bulk modulus)
- **structural changes**

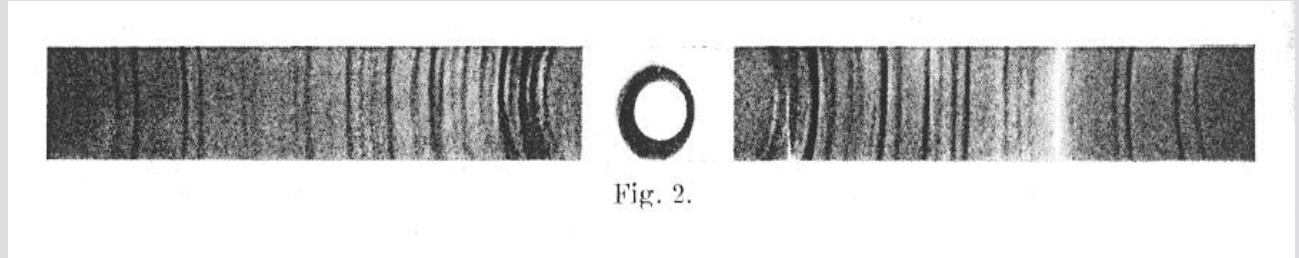


CRYSTAL STRUCTURE FROM 'IN SITU' POWDER DIFFRACTION: THE BEGINNING...

An early 'in-situ' powder xray diffraction experiment on the structure of α -N₂

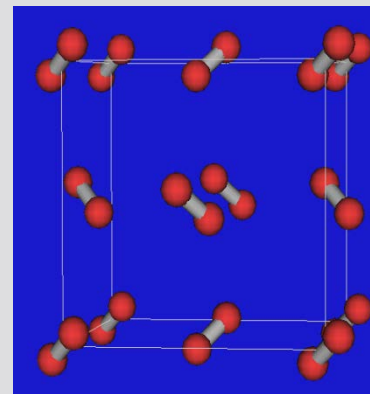


Schematic drawing
of the used cryo-
camera

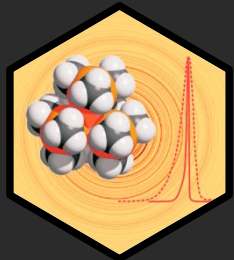


Debye-Scherrer film of α -N₂ at 34K

Exposure time: 18h

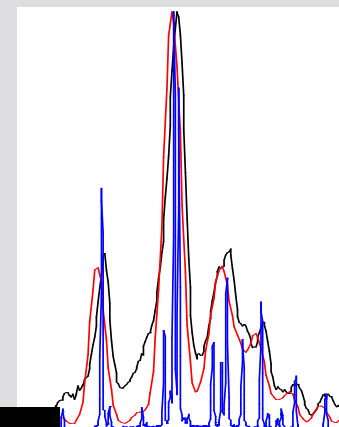


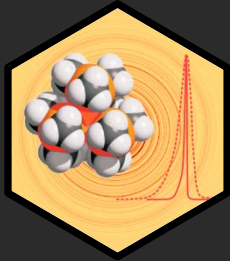
α -N₂ Pa-3



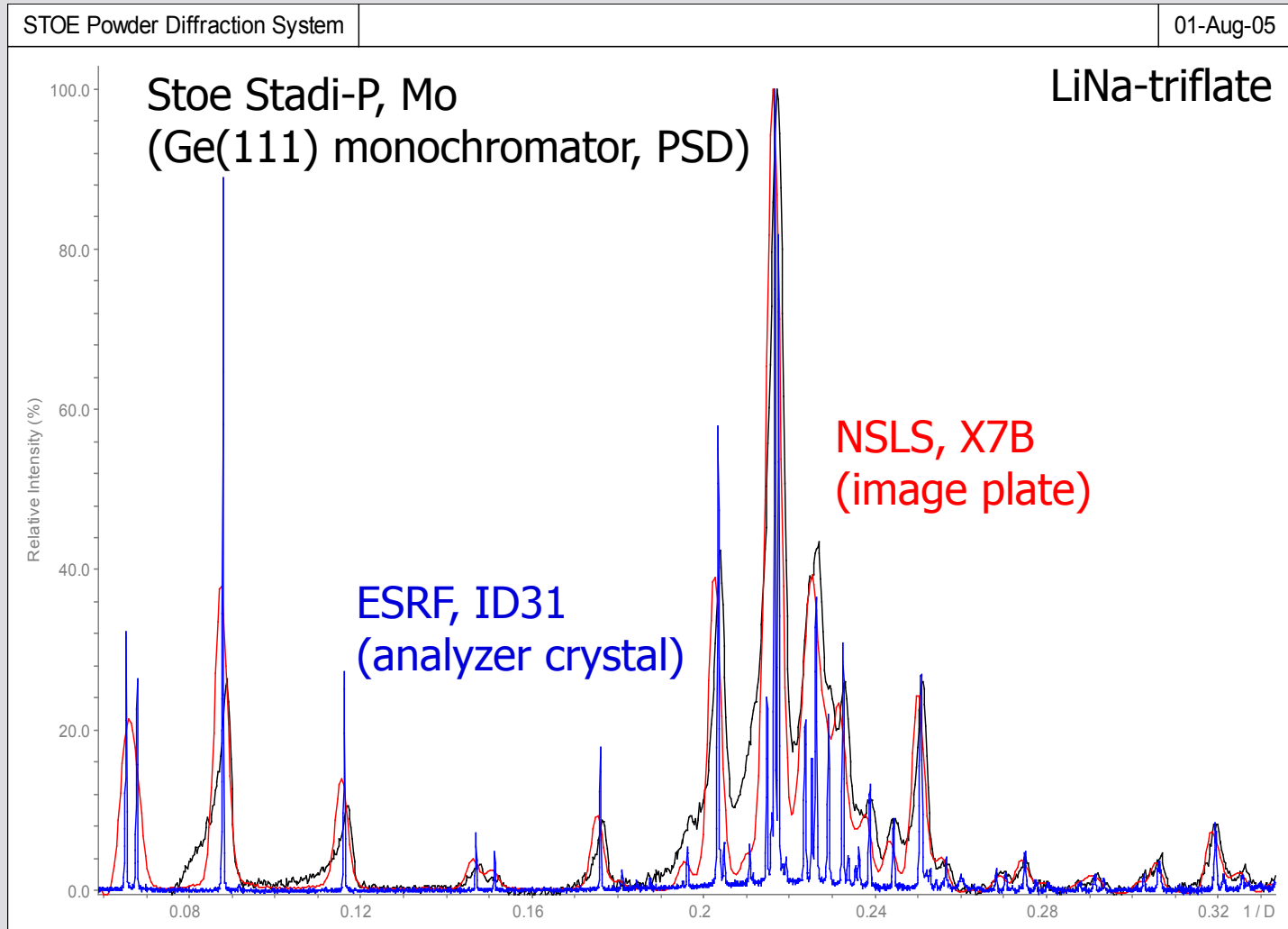
THE BOTTLE NECKS

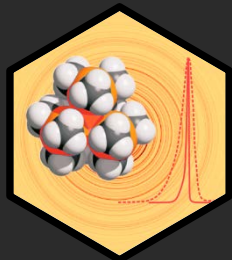
- Scanning Speed
- Instrumental Resolution
- Data Reduction Software





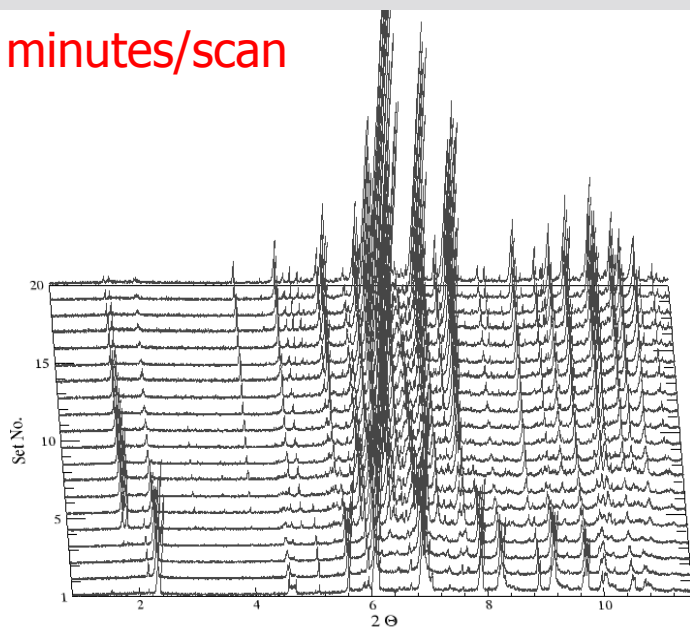
WHY BETTER RESOLUTION...



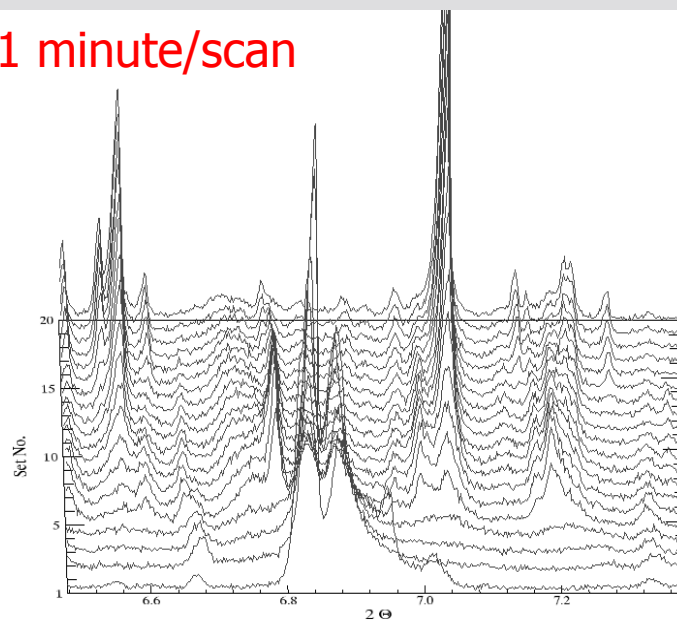


WHY HIGHER SPEED

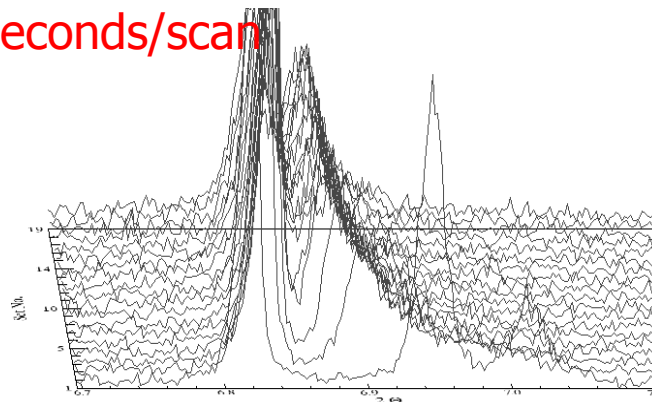
5 minutes/scan



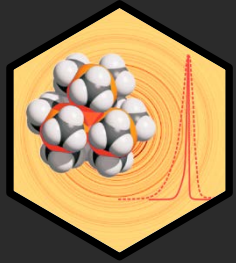
1 minute/scan



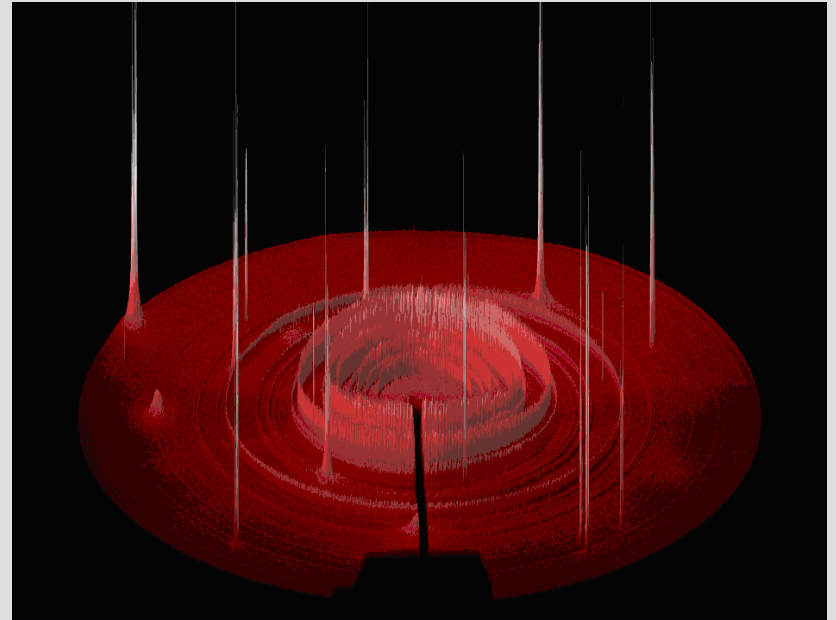
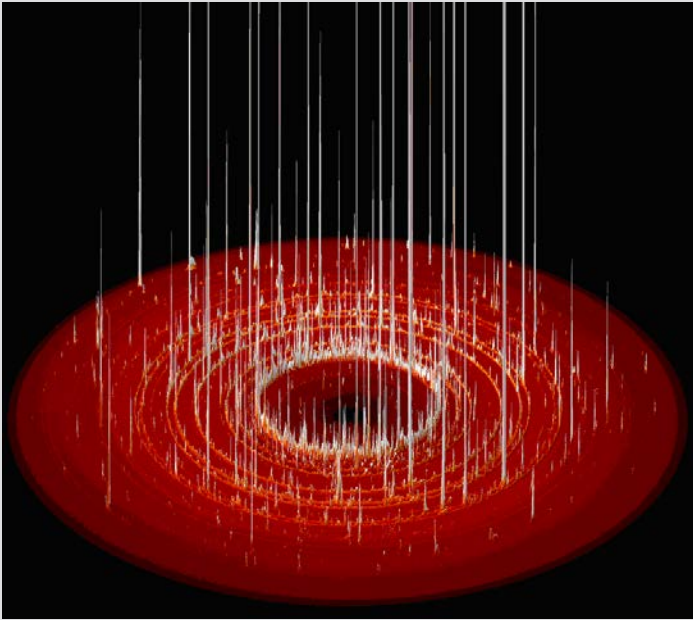
30 seconds/scan



Radiation induced phase transition
of Rubidium triflate measured at
ID31, ESRF, July, 2005

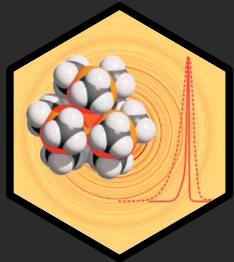


WHY BETTER SOFTWARE

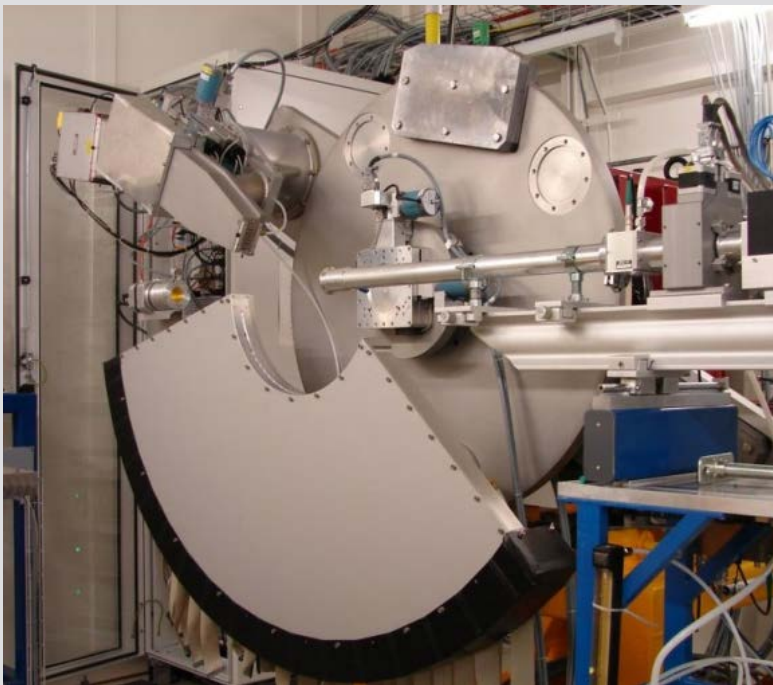


Rendered images of a high pressure powder diffraction data set (left) and a high temperature powder diffraction data set (right) collected by a 2D-image plate detector.

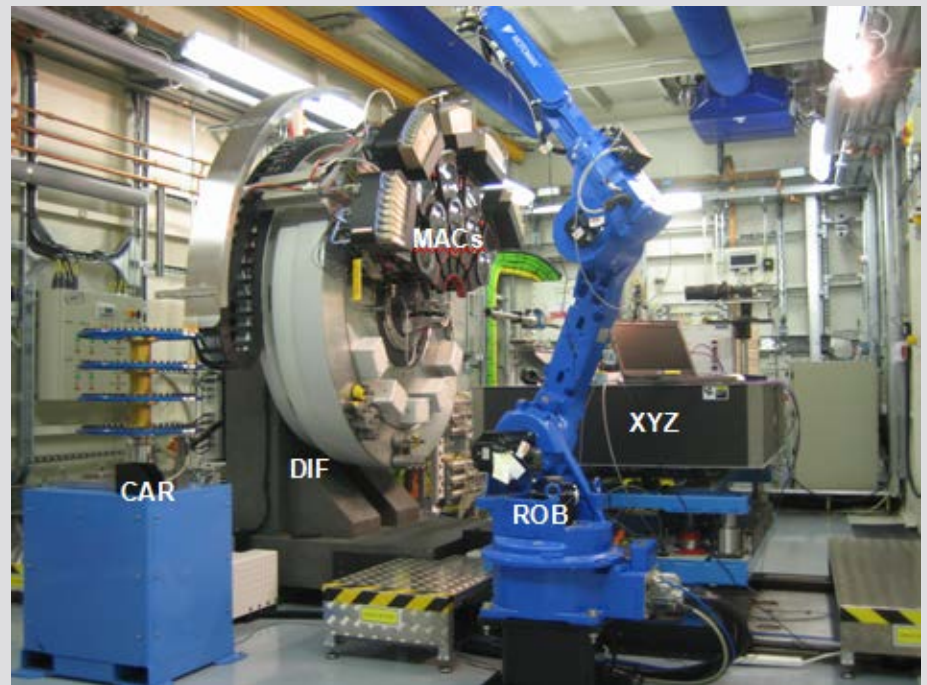
The white spikes are high intensity peaks originating from larger grains within a fine grained matrix (left) or reflections of the high temperature sapphire capillary (right).



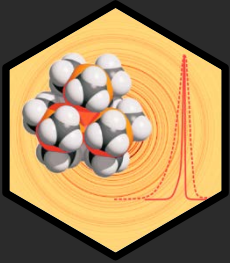
SOLUTION OF THE SPEED/RESOLUTION PROBLEM: HIGH RESOLUTION PSD'S



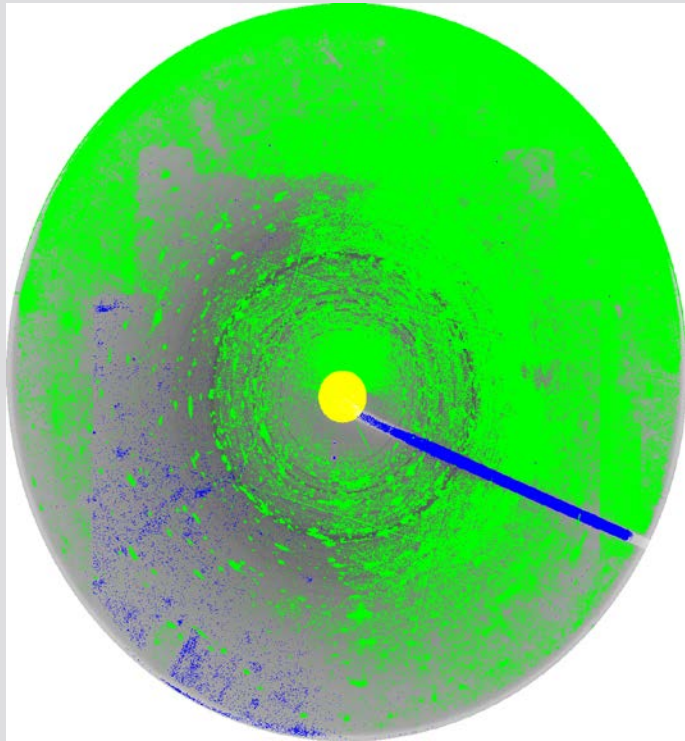
The 30720-element Mythen II strip detector, covering 120° at the SLS



5* with 9 Si(111) analyzing crystals at I11, Diamond

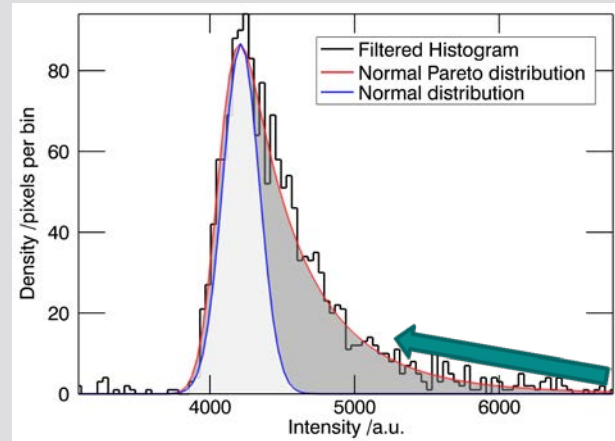


IMPROVING DATA REDUCTION BY THROWING AWAY INFORMATION

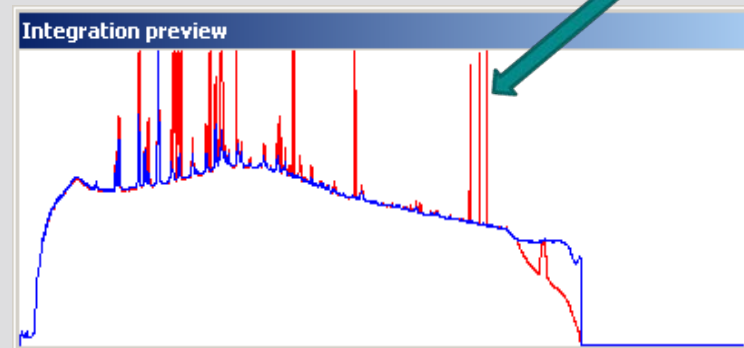


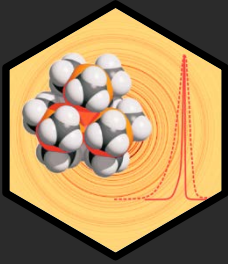
Green mask: top 48%.
Blue mask: bottom 2%
Yellow mask: beam stop
Grey region: used for integration

The effect of fractile filtering on the diffraction image is shown:

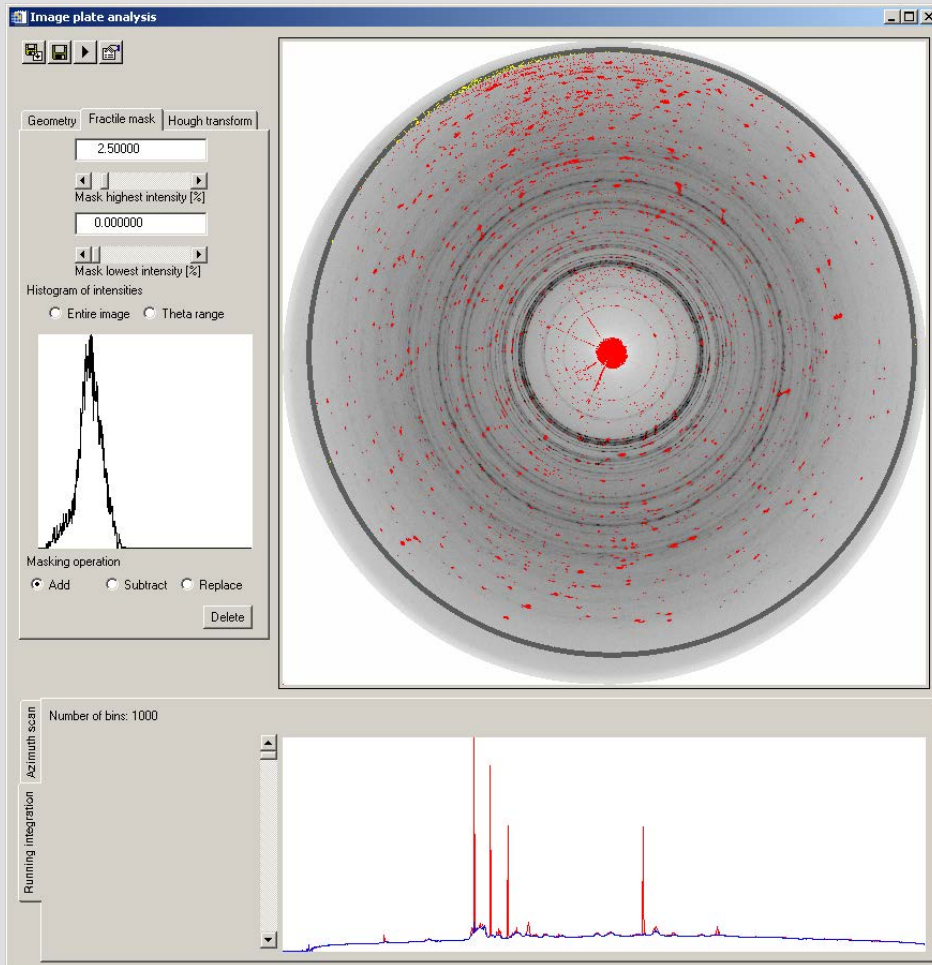


Filtered part

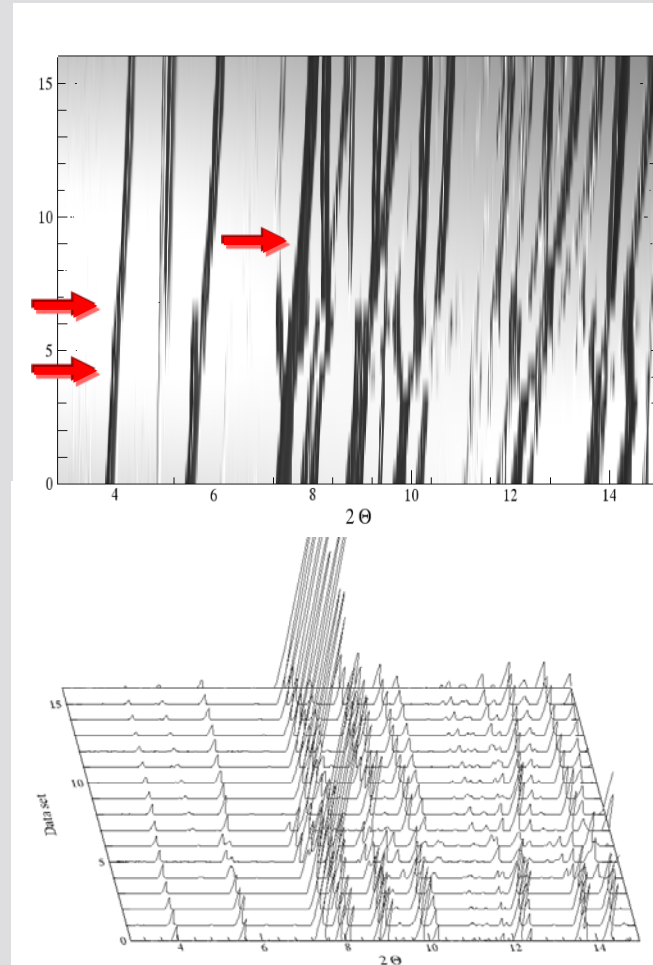




BENEFIT FOR DETECTING PHASE TRANSITIONS



FeSb_2O_4 at 53 kbar

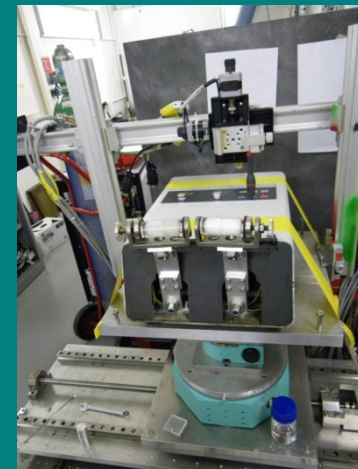


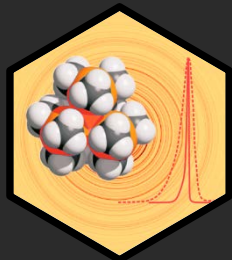
Final integrated powder patterns

REAL-TIME AND IN SITU MONITORING OF MECHANOCHEMICAL MILLING REACTIONS

**Tomislav Friščić, Ivan Halasz, Patrick J. Beldon,
Ana M. Belenguer, Frank Adams, Simon A.J. Kimber,
Veijo Honkimäki, Robert E. Dinnebier**

2013, Nature Chemistry 5, 66–73.





MECHANOCHEMISTRY: DIFFERENT WAYS OF INTRODUCING ENERGY INTO A REACTION:



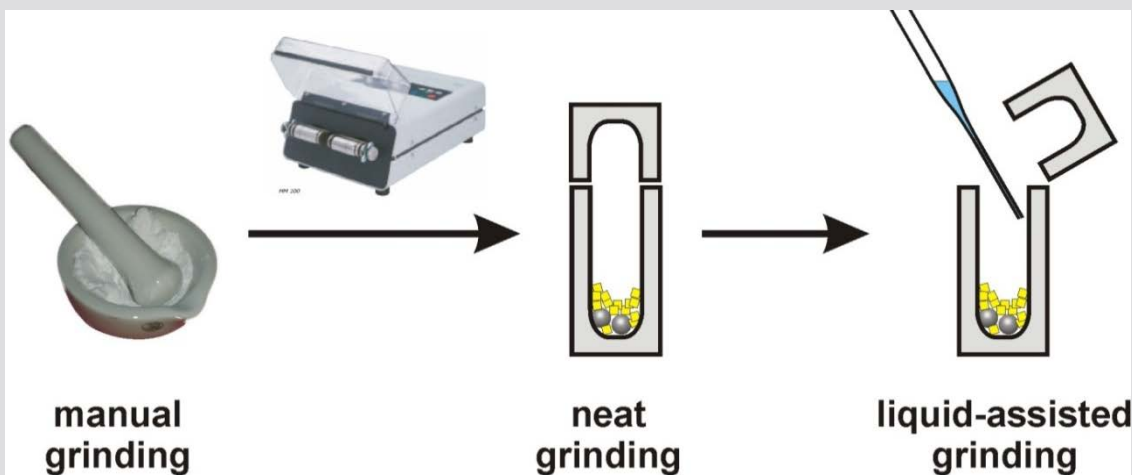
Heat



Radiation/electricity



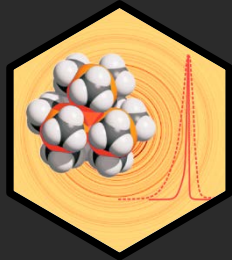
Grinding (mechanical force)



manual grinding

neat grinding

liquid-assisted grinding



METAL-ORGANIC FRAMEWORKS (MOFs) - ARCHETYPAL MODERN MATERIALS

leads to the adsorption of 26 times more
Joseph Wood

BASF develops MOF materials

CONFERENCE REPORT

Ulrich Müller described BASF's progress in commercializing metal-organic framework materials (MOFs) at the 2005 International Conference on Materials for Advanced Technologies in Singapore. BASF became interested in MOFs because of their remarkable surface areas. MOFs can also be synthesized with readily available, cheap, rhodium.

The company now of MOF from a simply mixed with terephthalate in a recycled. The MOF shaped into tablets BASF are not look materials, rather develop systems to Muller explained th maximal storage of Filling a gas canister



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New Material Could Vastly Improve Carbon Capture

"Metal organic frameworks" look like rocks, act like sponges

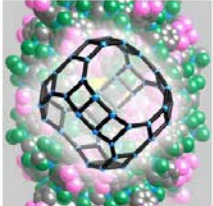
By Paul Voosen | June 30, 2009 | 5

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To capture the carbon dioxide generated by coal plants, chemical companies like Dow Chemical Co. and energy giants like Alstom SA have been betting big on liquid solvents like amine, a corrosive derivative of ammonia that has a thirst for binding with CO₂.

Problem is, once the two are bound, they never want to part.

In an attempt to circumvent the huge energy demands needed by currently available



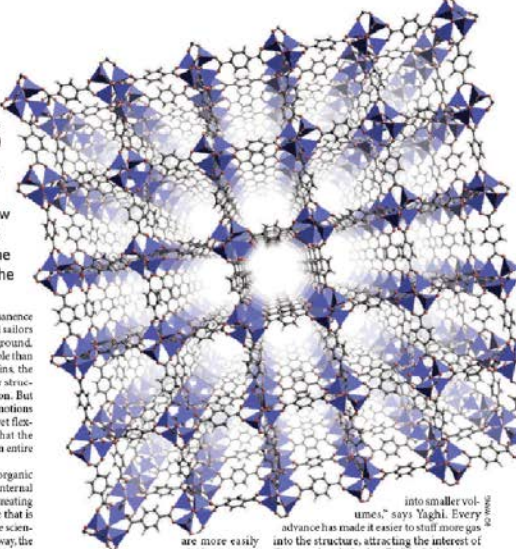
METAL ORGANIC FRAMEWORKS:

NEWS FEATURE

NATURE | 449 | 8 August 2007

SPACE INVADERS

space exploration usually means leaving Earth's orbit. But chemists are now burrowing inside solids to open new vistas. Katharine Anderson reports from the internal frontier.



Solids have a reputation for permanence and reliability. Astronauts and sailors rejoice in returning to solid ground. And few things are more durable than old gold. Unlike their flightier cousins, the seas and liquids, solids have regular structures that generally resist deformation. But now are materials that challenge our notions of a solid. They are rigid and strong, yet flexible, incredibly light yet so porous that the surface area of one gram could cover an entire football pitch.

These solids are known as metal-organic frameworks, or MOFs. They have no internal walls, just a bare molecular scaffold, creating a regular, sturdy, crystalline structure that is packed full with empty space. But if the scientists building these structures get their way, the materials will soon be packed full of useful gases, such as compressed hydrogen or methane, making it easier to store and transport these alternative fuels.

Chemists expected that the vast, open

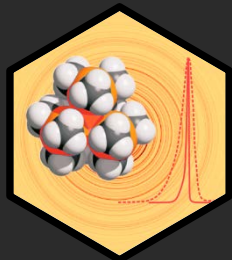
into smaller volumes," says Yaghi. Every advance has made it easier to stuff more gas into the structure, attracting the interest of German chemicals giant BASF, which is about to move its MOF research into small-scale production.

Yaghi is thrilled by BASF's investment in his work. In Yaghi's lab, MOFs can now be

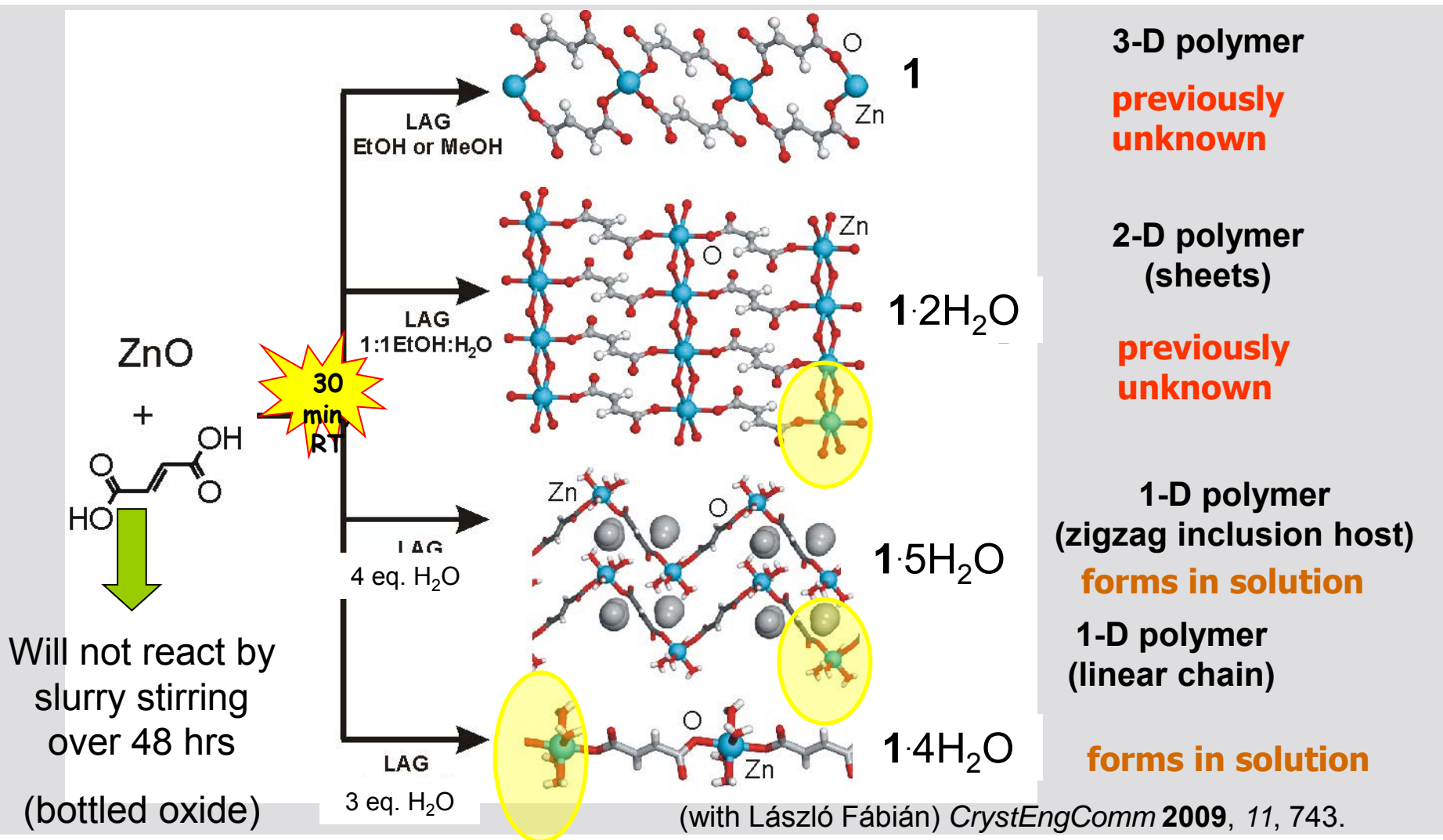
are more easily penetrated than most, including clays and zeolites, to trap and store gases for decades. But the pores and channels in naturally occurring porous materials are of

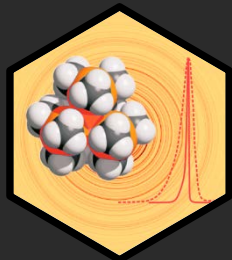
PHOTO: J. YAGHI

Applications in hydrogen storage, CO₂ sequestration and catalysis
Commercial materials: Basolite (BASF/Sigma-Aldrich)



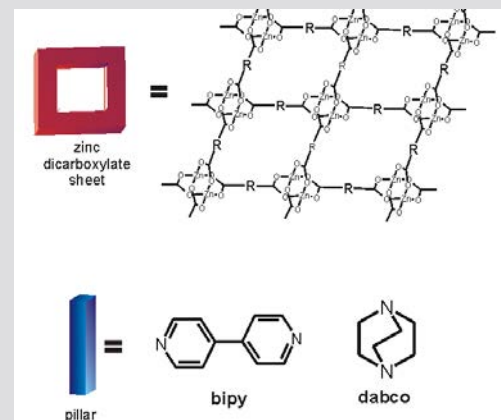
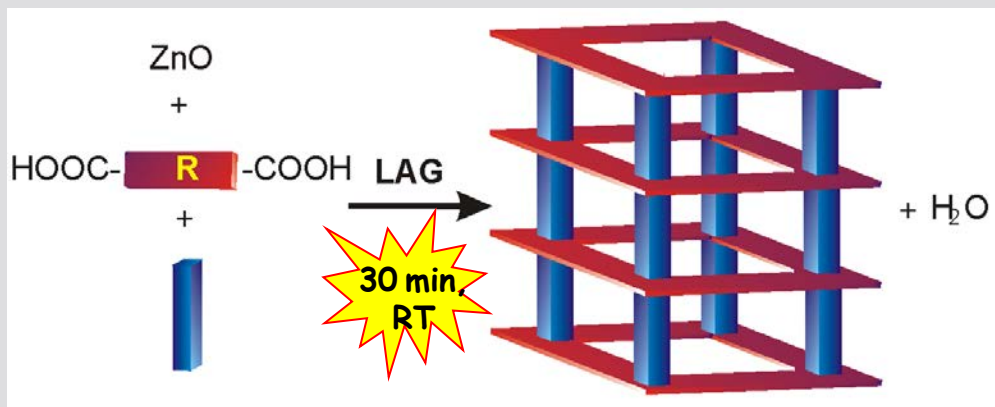
LIQUID-ASSISTED GRINDING (LAG): SCREENING FOR METAL-ORGANIC ARCHITECTURES



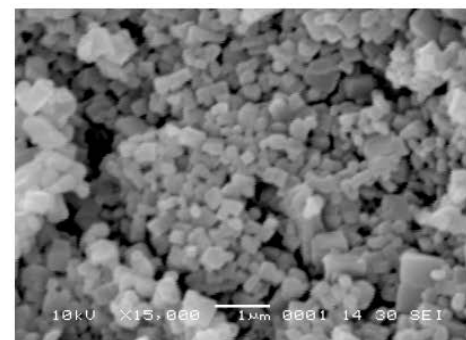
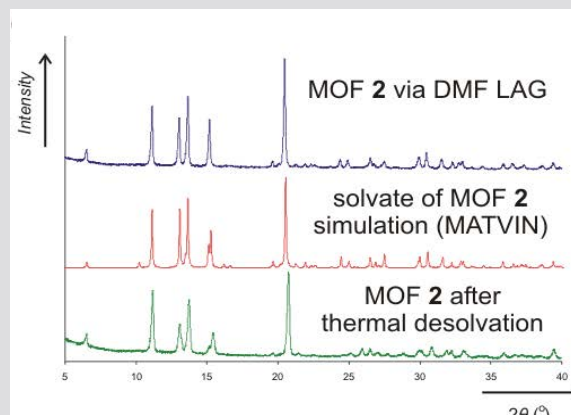
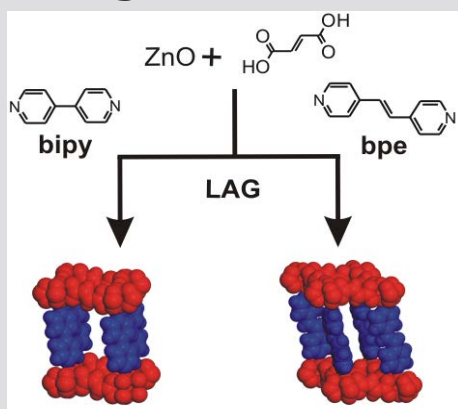


PILLARED MOFS DIRECTLY FROM THE OXIDE

We'd like:



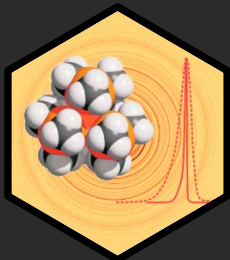
We get:



(No rxn by neat grinding)

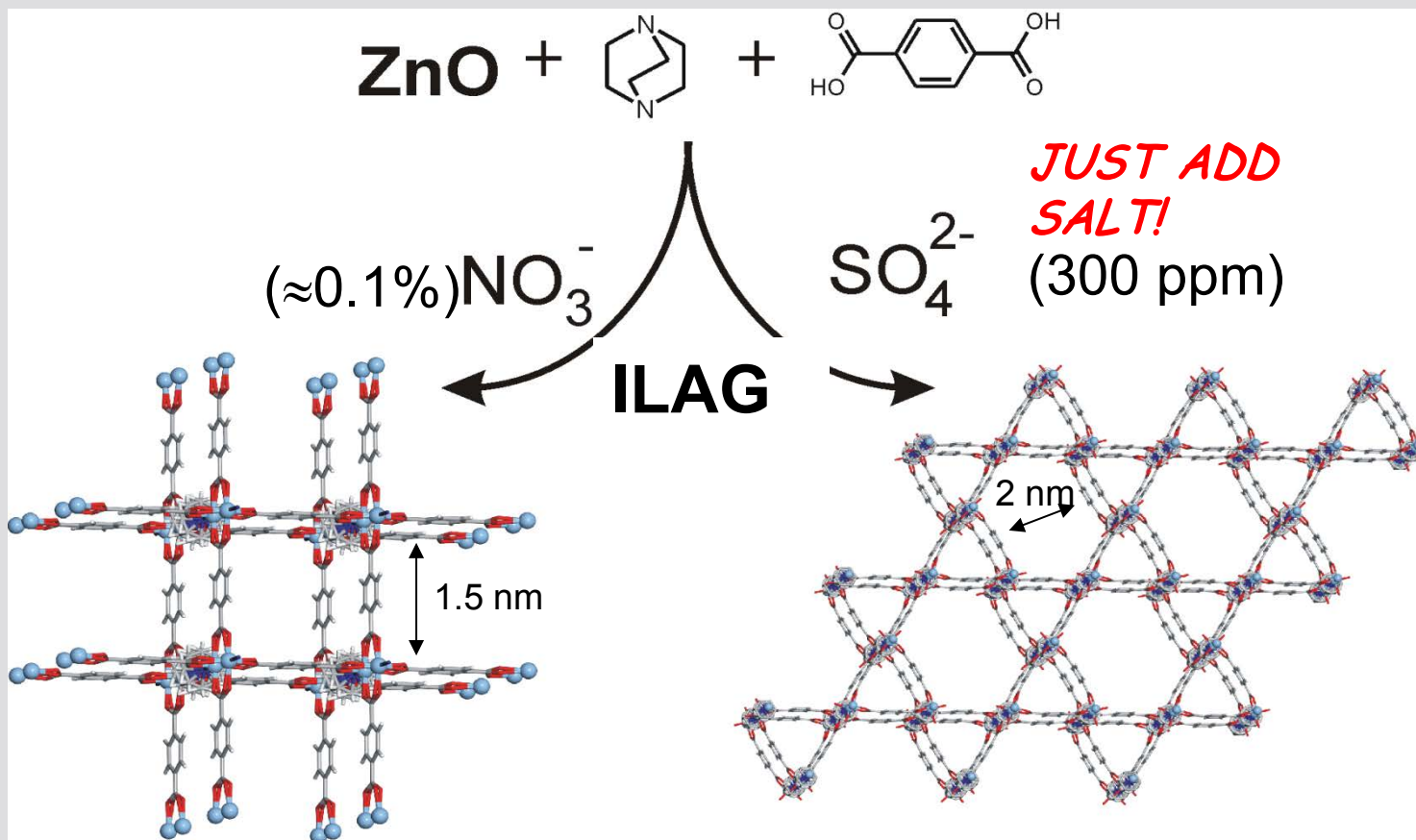
(with László Fábián *CrystEngComm* 2009, 11, 743)

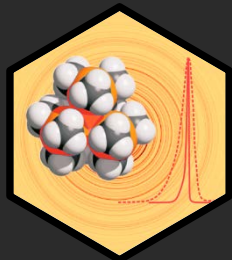
30 min grinding replaces 2 days solvothermal synthesis in DMF, 160°C



TEMPLATING AND CATALYSIS IN MOF MECHANOSYNTHESIS

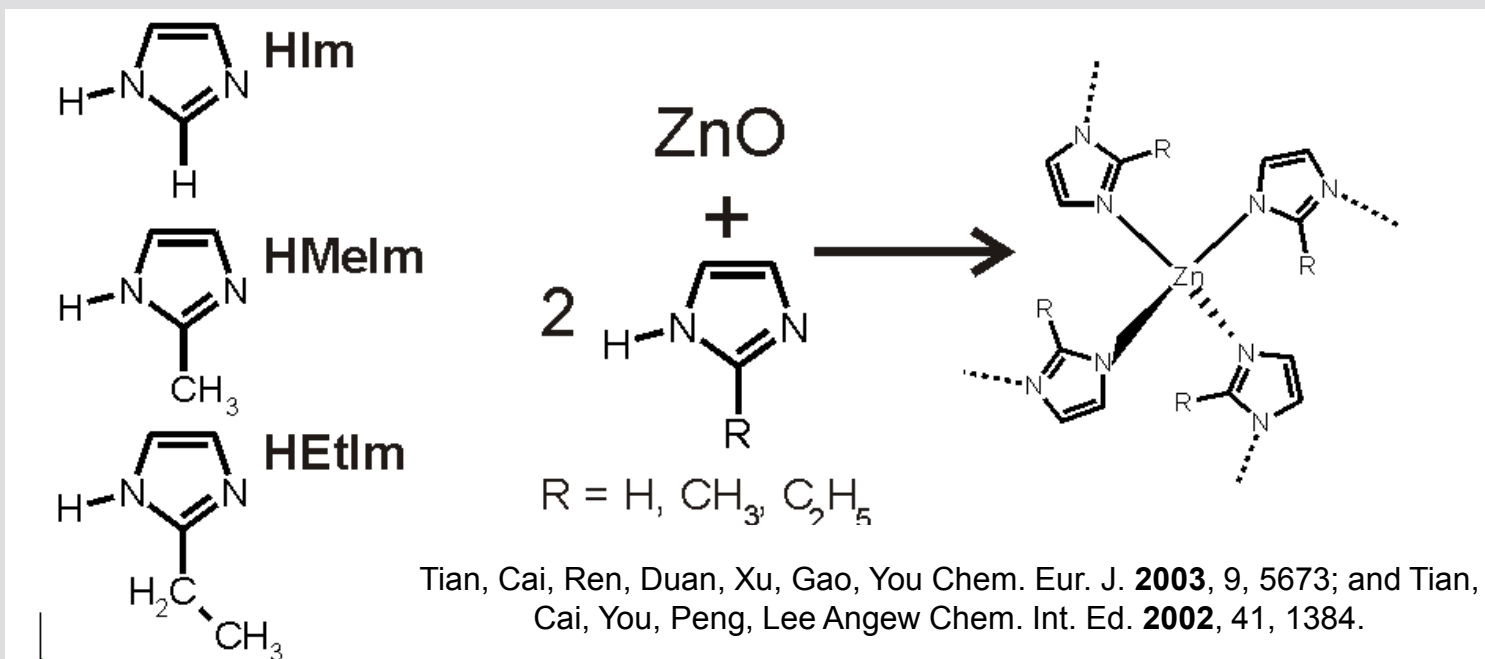
ILAG: ion- and liquid- assisted grinding





ZEOLITIC IMIDAZOLATE FRAMEWORKS (ZIFs)

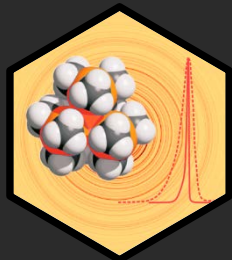
The bent geometry of the imidazole ligand, coupled with the tetrahedral coordination of Zn^{2+} results in a metal-organic analogy of silicate/zeolite structures



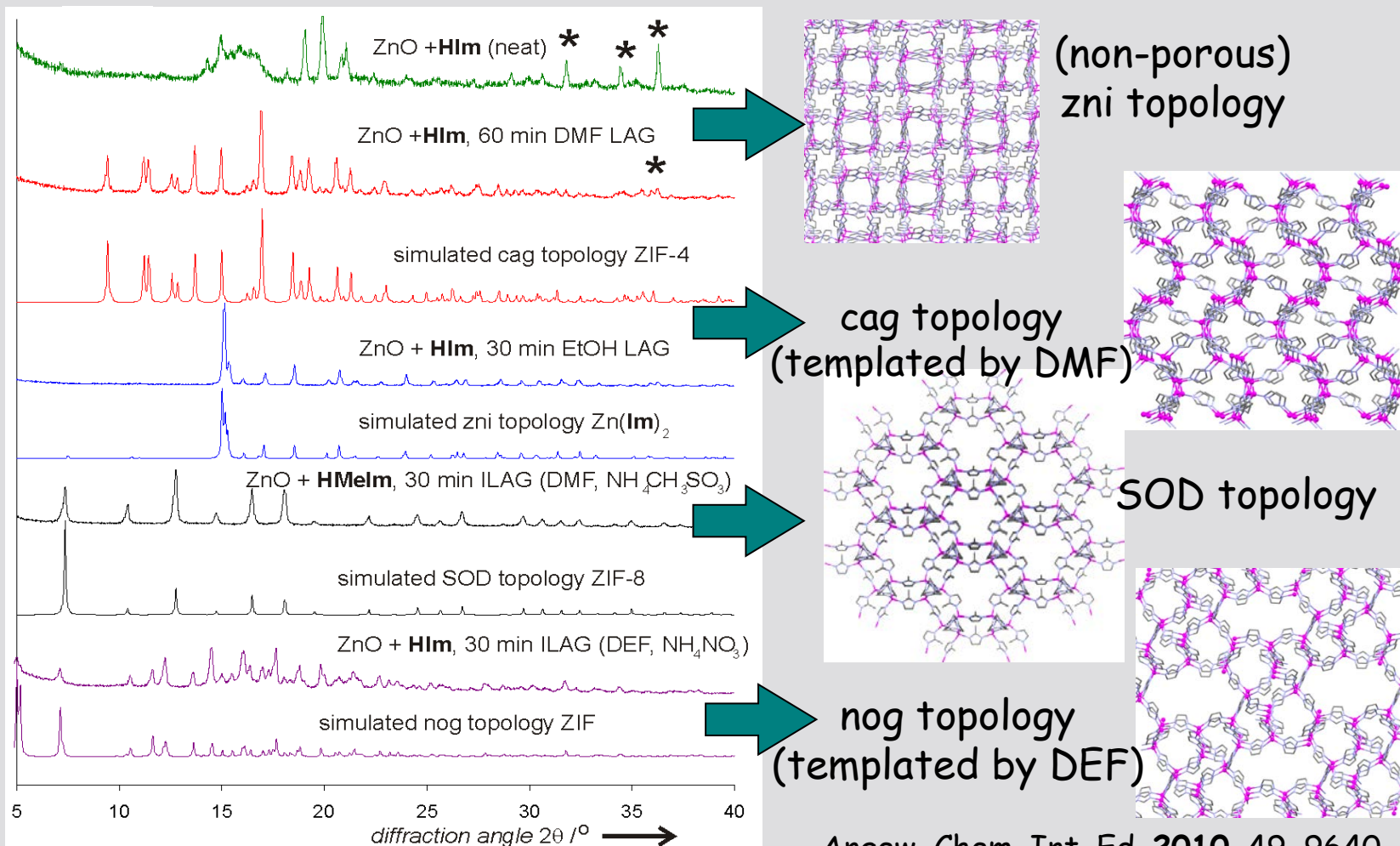
Normally obtained through solvothermal synthesis from zinc nitrate and imidazole ligand. Yields often below 40%

Yaghi, Furukawa, Wang *US Pat. Appl. No. US 2010/0186588 A1* (07/29/2010)

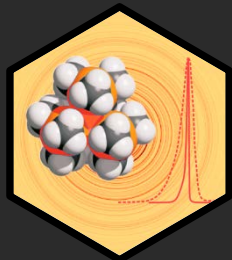
Park, Ni, Coté, Choi, Huang, Uribe-Romo, Chae, O'Keeffe, Yaghi *Proc. Natl. Acad. Sci.* **2006**, *103*, 10186.



EFFICIENCY AND TEMPLATING OF ZIF SYNTHESIS USING LAG/ILAG

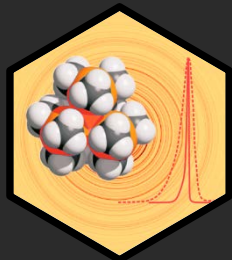


Angew. Chem. Int. Ed. 2010, 49, 9640.

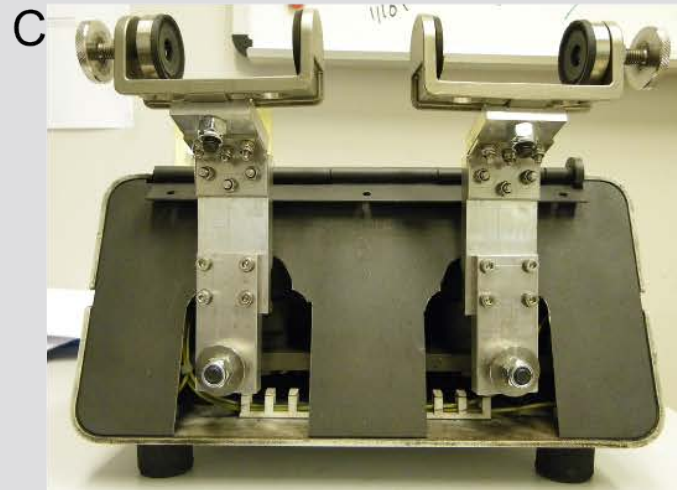


DISCOVERY OF NEW MATERIALS THROUGH ILAG

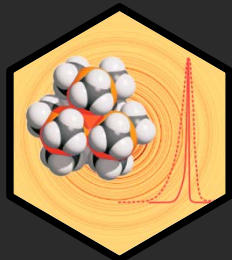




MECHANISTIC STUDIES – LOOKING INTO THE REACTION

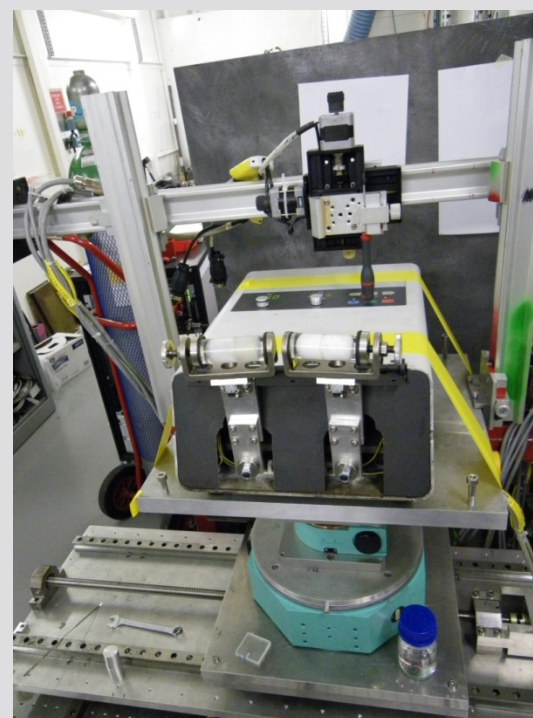


Modified Retsch MM200 mill



MECHANISTIC STUDIES – LOOKING INTO THE REACTION

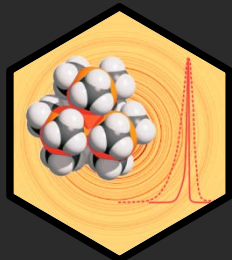
Visual observation is not very useful - switched to 90 keV X-ray beam
at ID15, ESRF, Grenoble



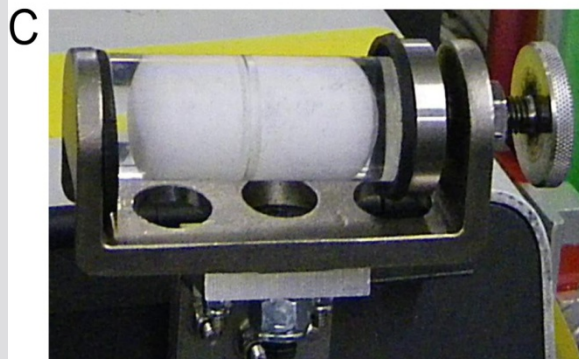
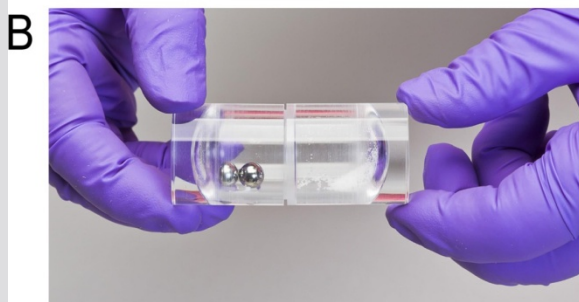
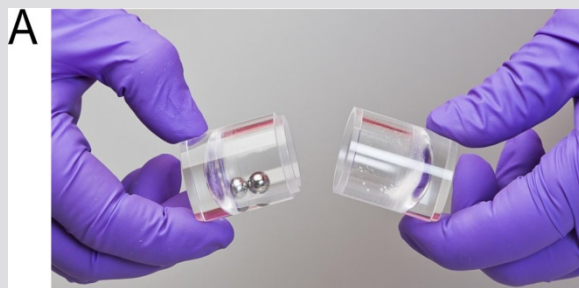
← mill

← movable
stand

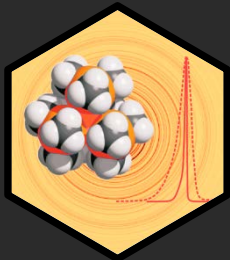
Nature Chem. 2013, 5, 66-73,
also C&EN 2012, 90: "X-ray vision for mechanochemical mills"



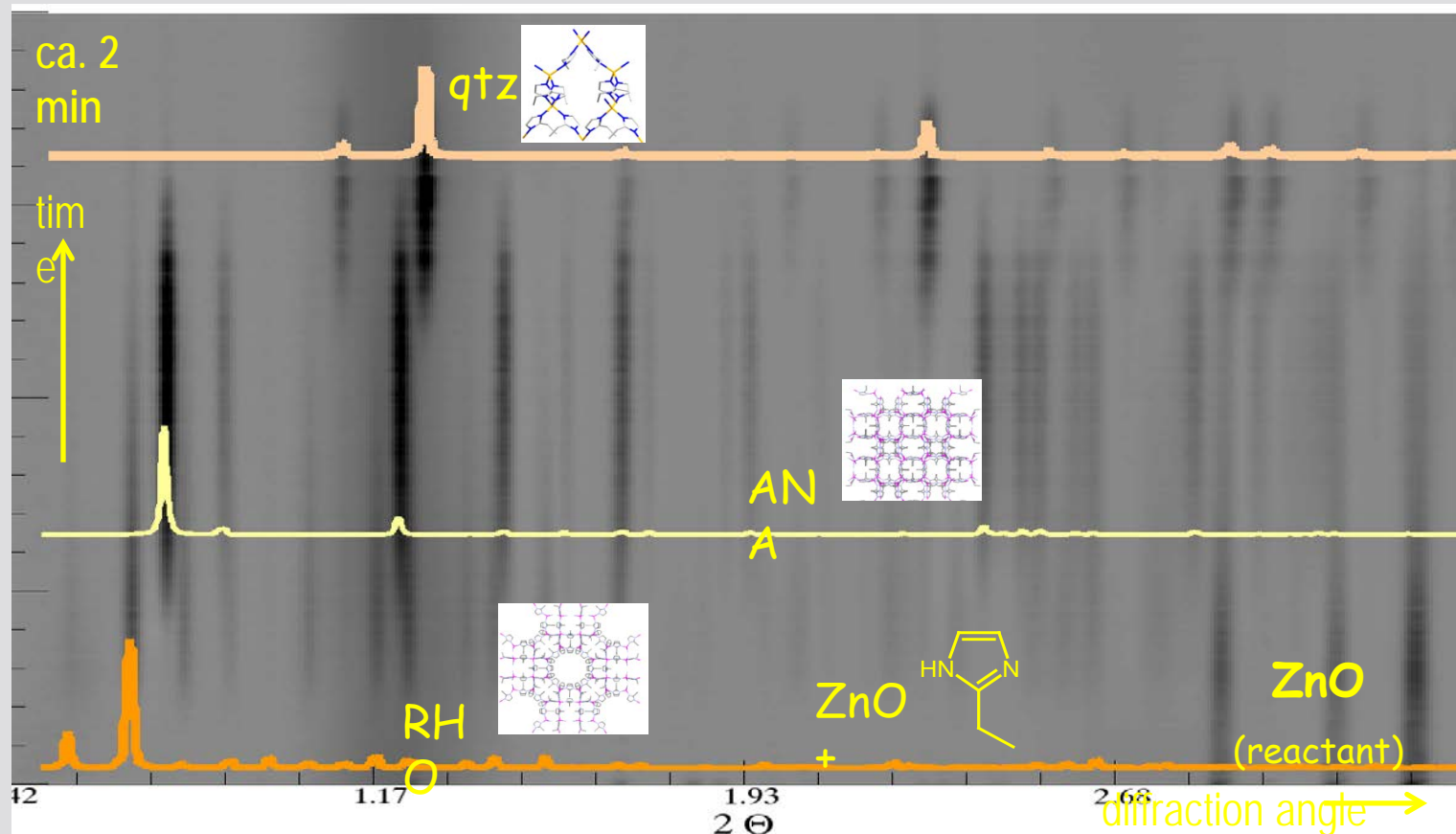
MECHANISTIC STUDIES - ...



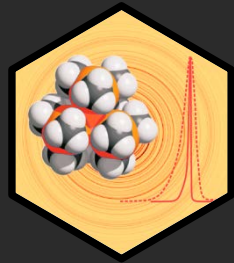
- (a) Proper handling of the two halves of the PMMA milling jar before snapping them together.
- (b) The milling jar after snapping the two halves together. The left side contains the milling liquid and milling balls. The right side contains the reactants (white powder).
- (c) horizontal positioning of the PMMA milling jar on the milling station.
- (d) the positioning of the mill and the milling jar with respect to the X-ray beam is verified using a teodolite.



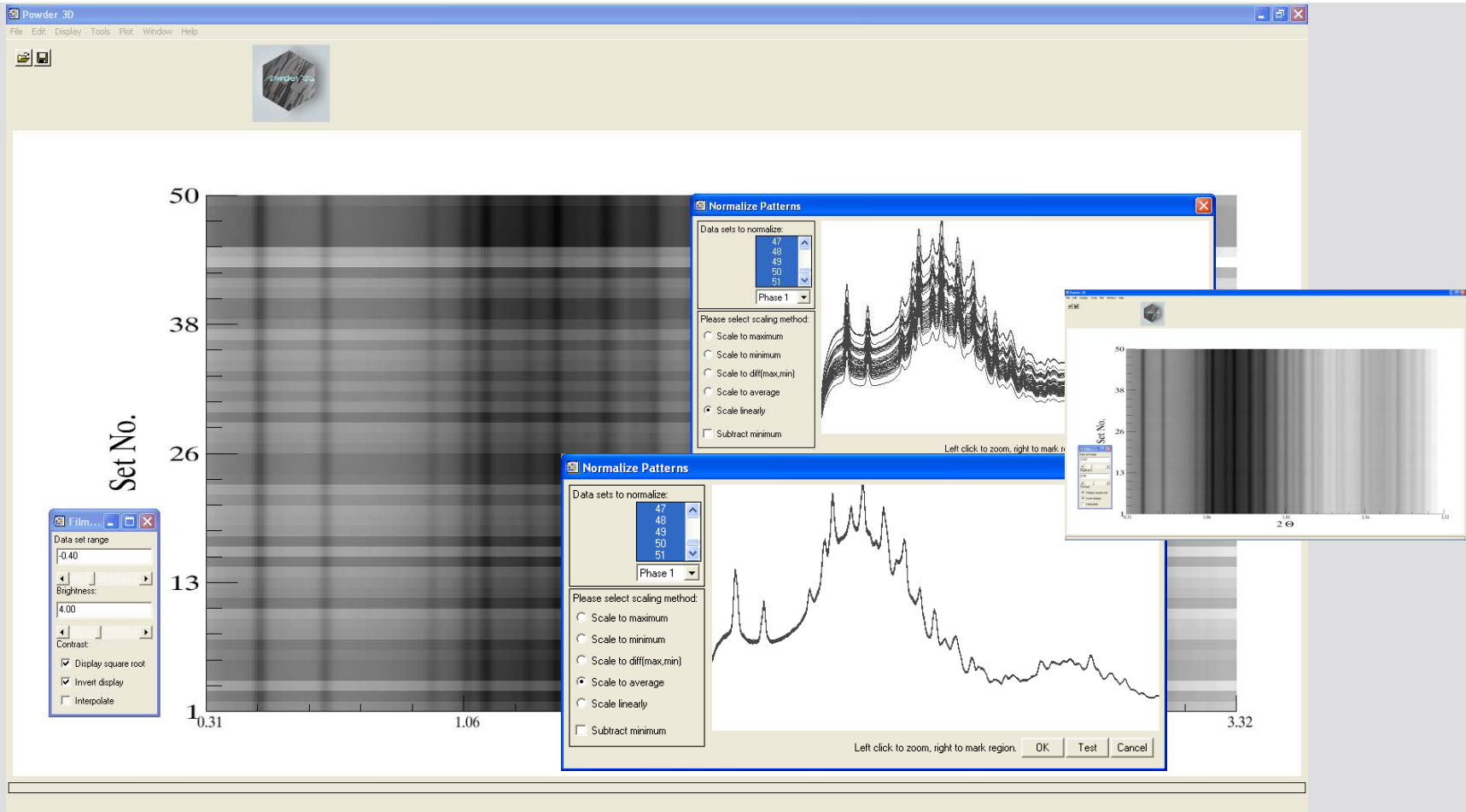
FORMATION AND TRANSFORMATION OF FRAMEWORK TOPOLOGIES



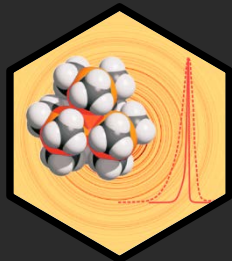
In situ measurements will provide mechanistic detail on the formation, transformation and stabilization of different framework types



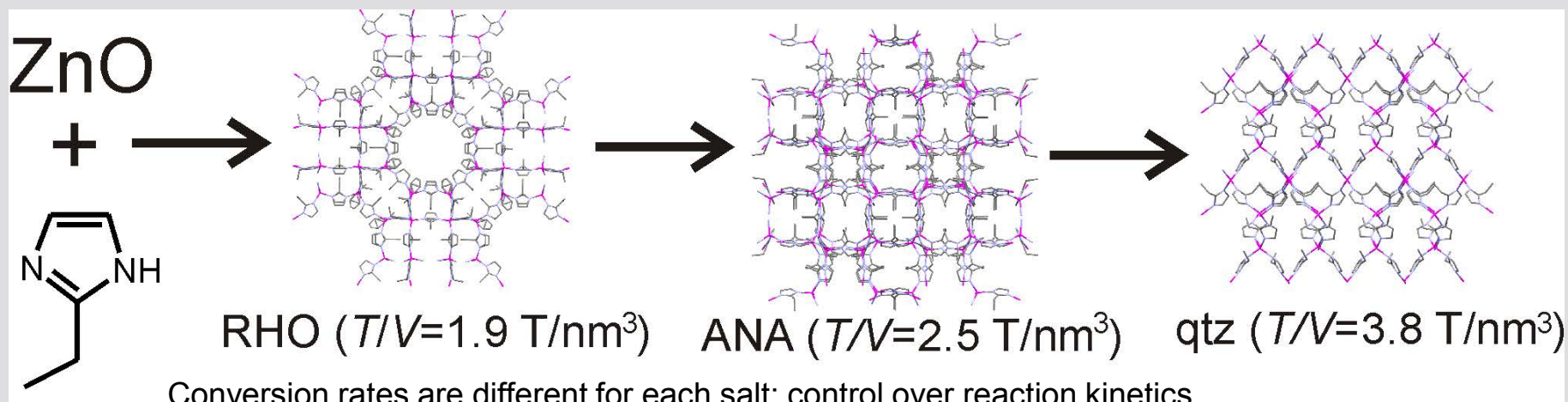
THE NEED FOR PROPER SCALING...



B. Hinrichsen, R.E. Dinnebier, M. Jansen, Powder3D, An easy to use program for data reduction and graphical presentation of large numbers of powder diffraction patterns, 2006, Z. Krist., Supplement Issue 23 (EPDIC-9 proceedings), 231-236.

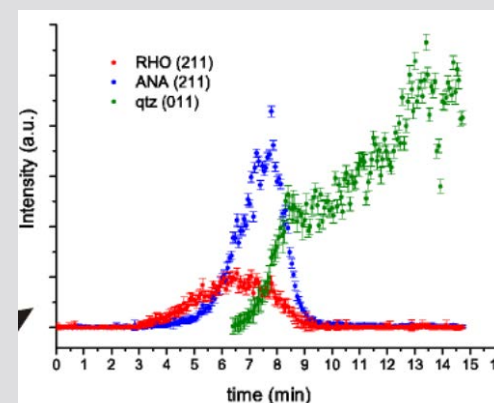
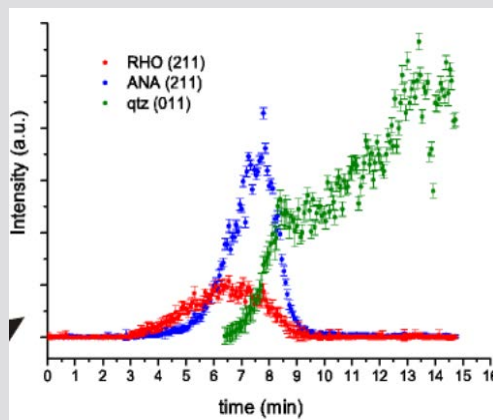
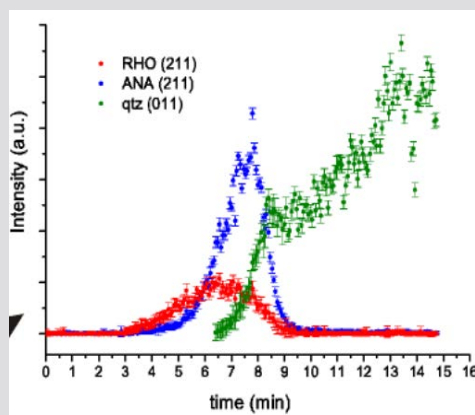


INTERMEDIATES AND KINETIC ANALYSIS USING TIME-RESOLVED PXRD

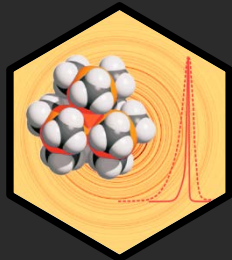


More liquid

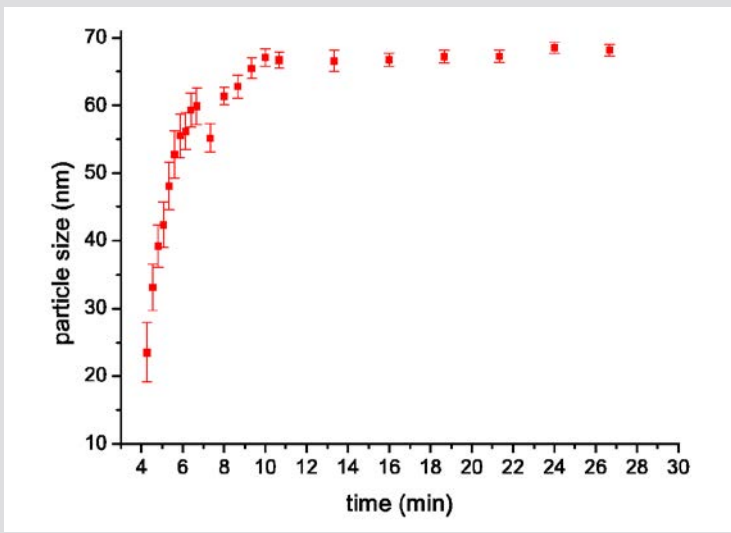
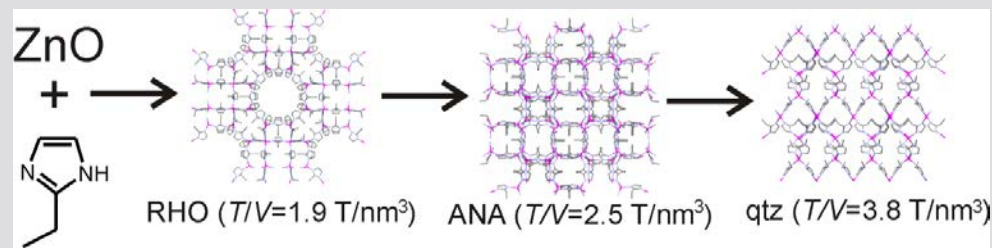
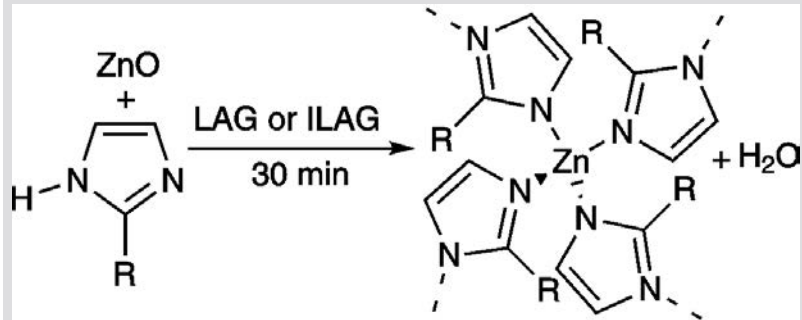
Less liquid



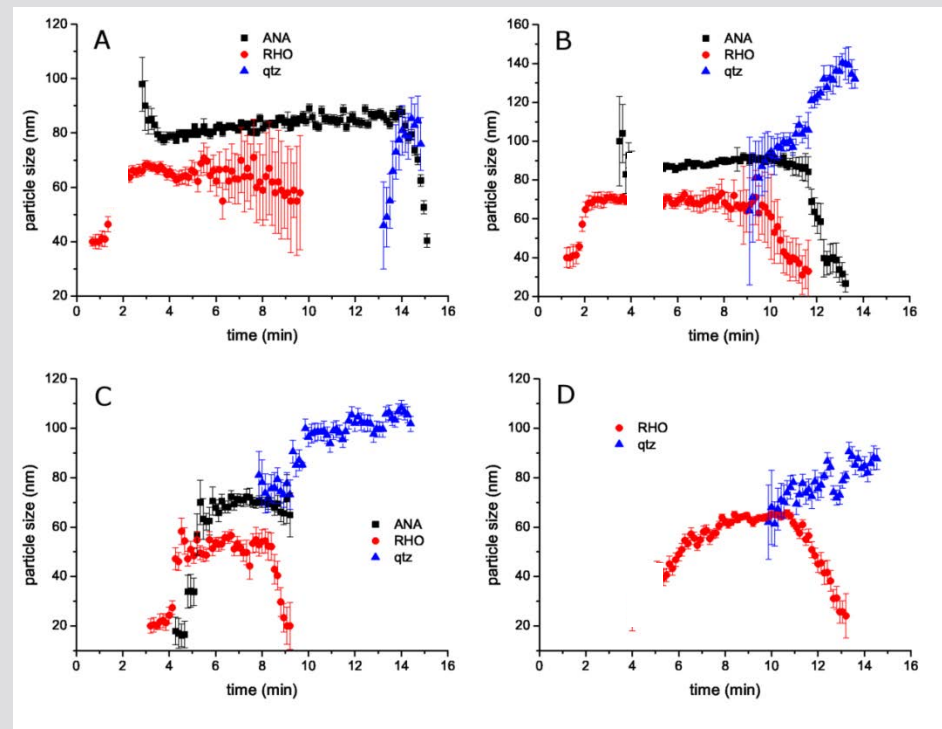
Direct monitoring of how reaction kinetics is affected by different parameters



PARTICLE SIZE EVOLUTION BY TIME-RESOLVED PXRD



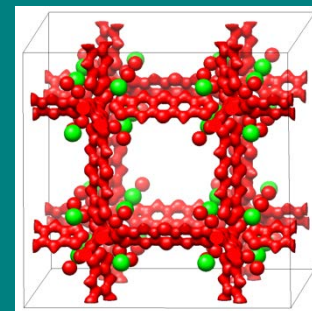
Direct monitoring of how particle size changes throughout milling

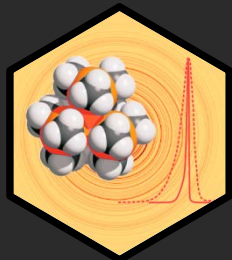


NOVEL CHARACTERIZATION OF THE ADSORPTION SITES IN LARGE PORE METAL-ORGANIC FRAMEWORKS: COMBINATION OF X-RAY POWDER DIFFRACTION AND THERMAL DESORPTION SPECTROSCOPY

Soleimani-Dorcheh, A., R. E. Dinnebier, A. Kuc,
O. Magdysyuk, F. Adams, D. Denysenko, T. Heine,
D. Volkmer, W. Donner and M. Hirscher

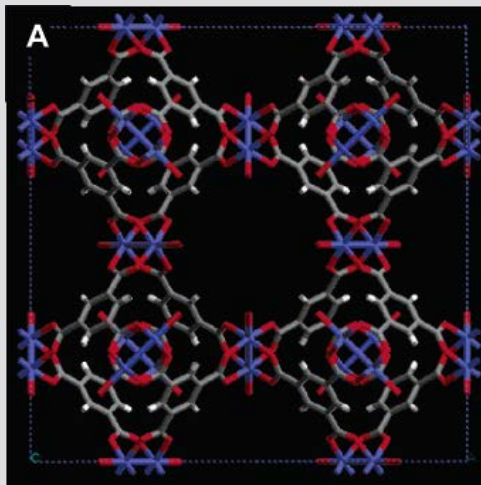
2012, *Physical Chemistry Chemical Physics* 14, 37, 12892-12897



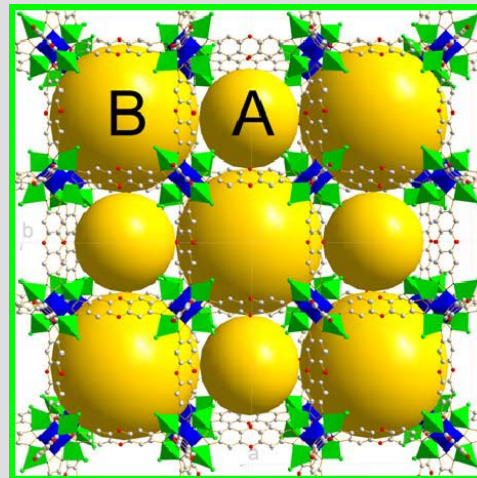


SELECTED MOF'S

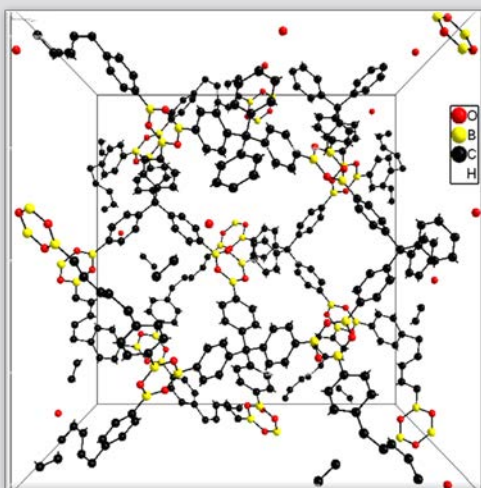
HKUST-1



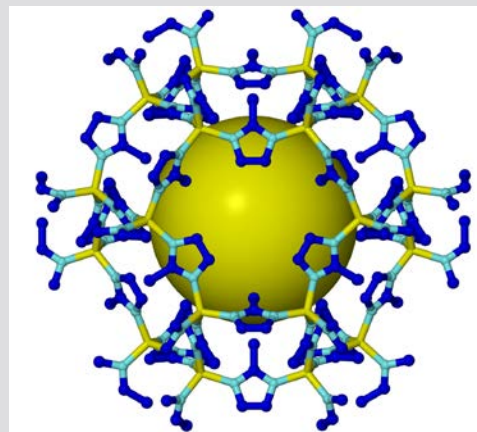
MFU-4l

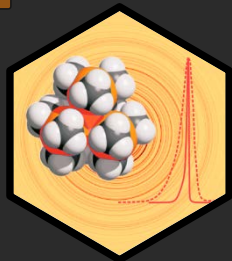


COF-102

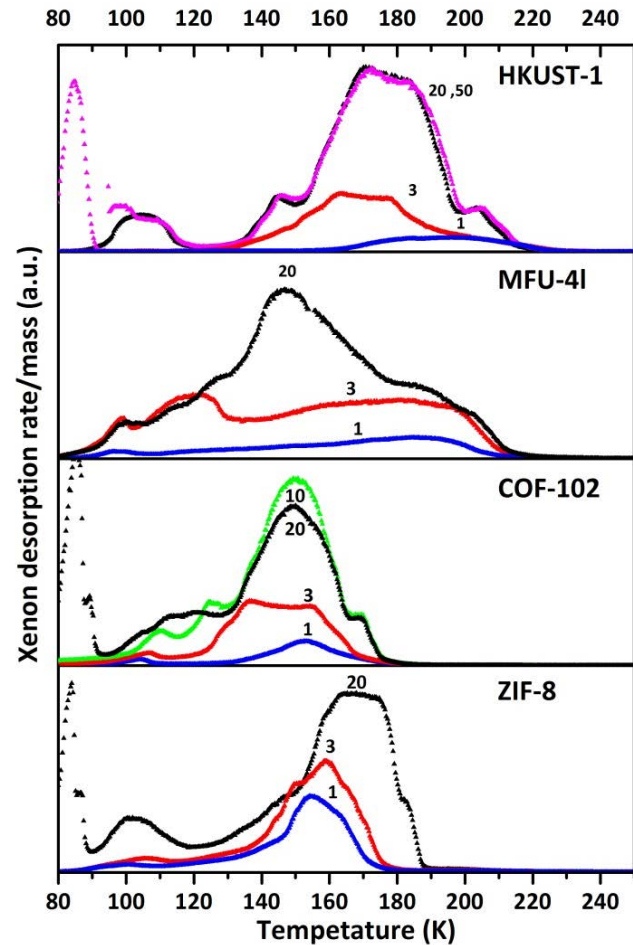
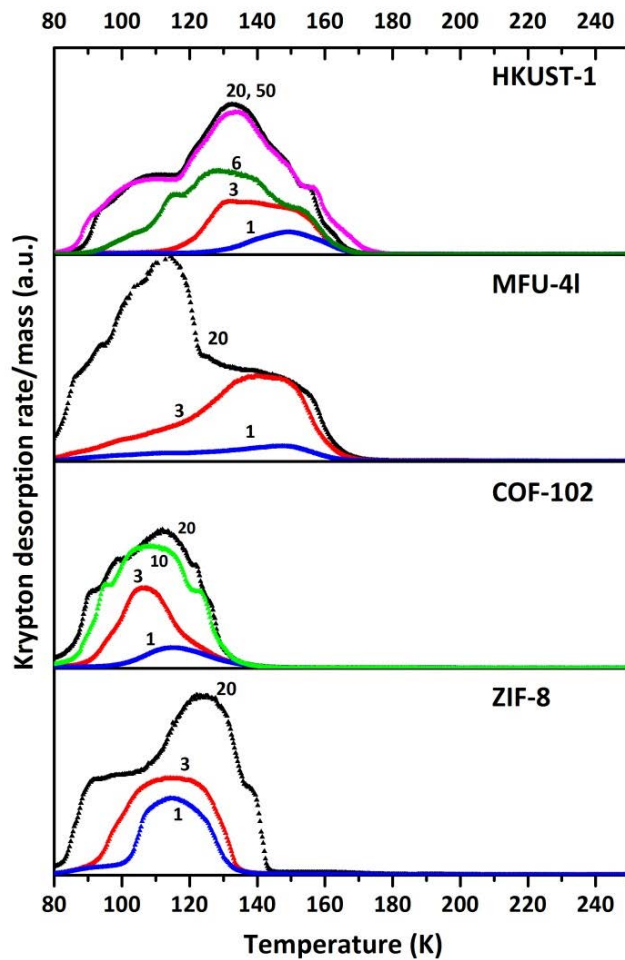


ZIF-8

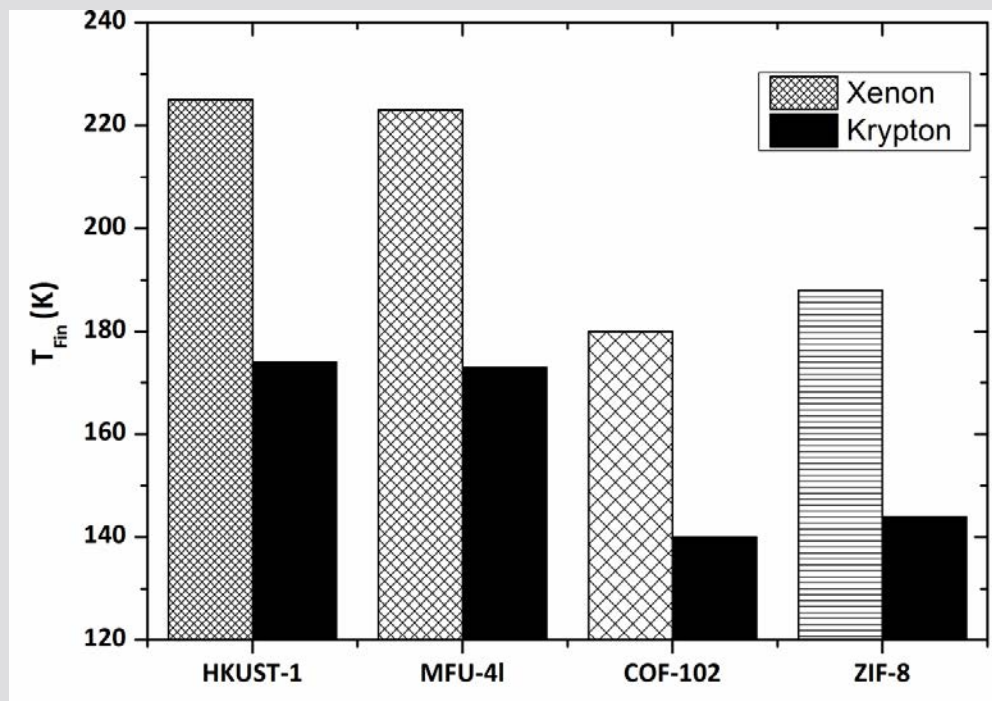
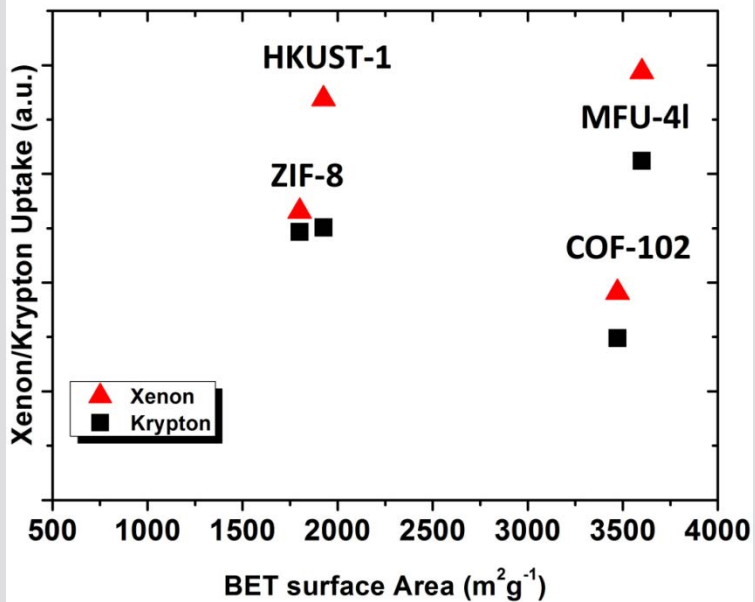
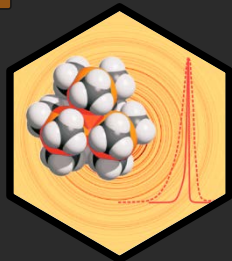




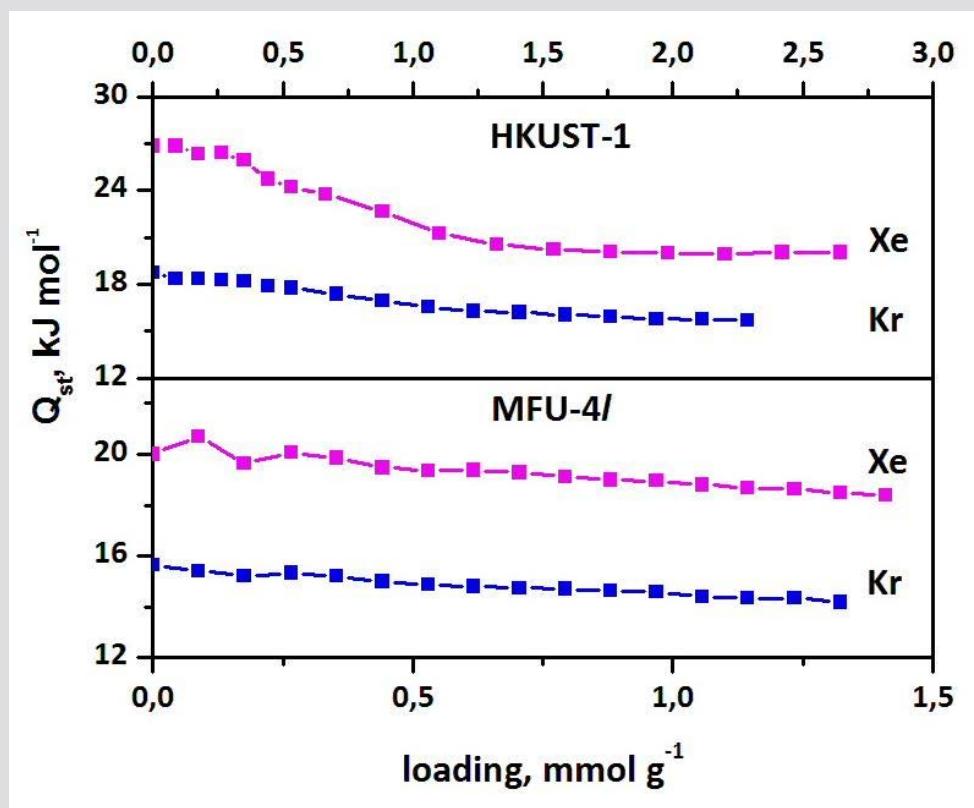
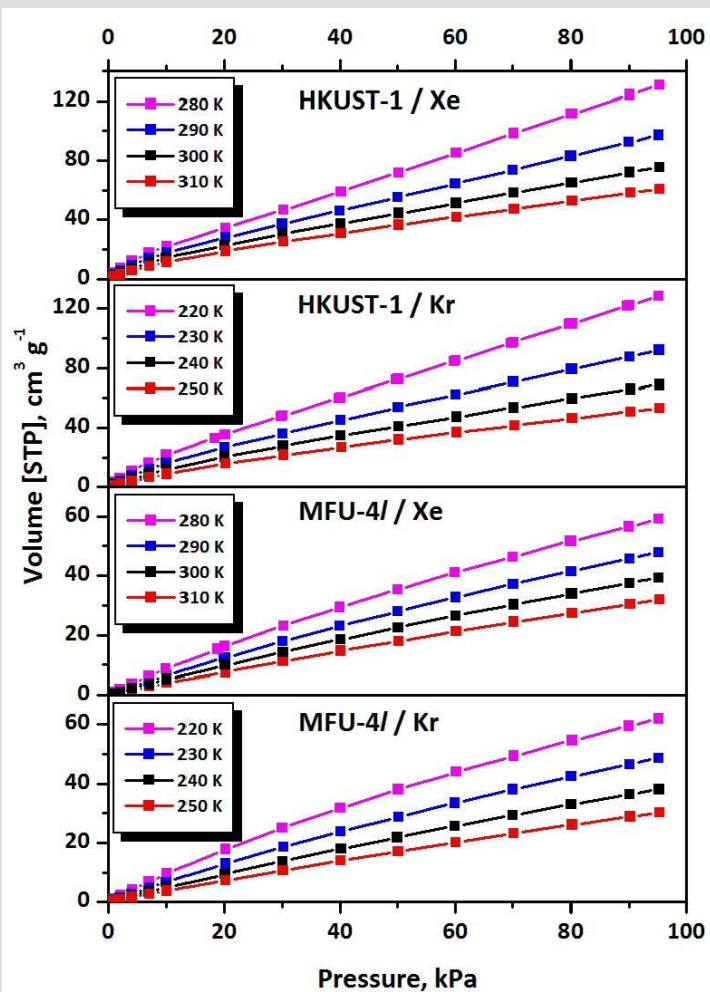
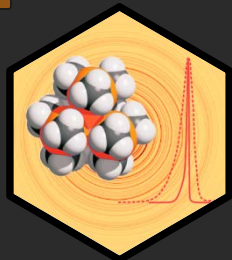
KR AND XE DESORPTION SPECTRA

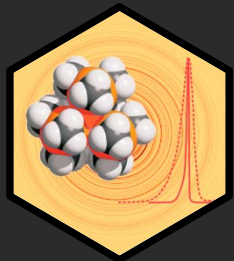


UPTAKE AND DESORPTION TEMPERATURE



ADSORPTIONS ISOTHERMEN

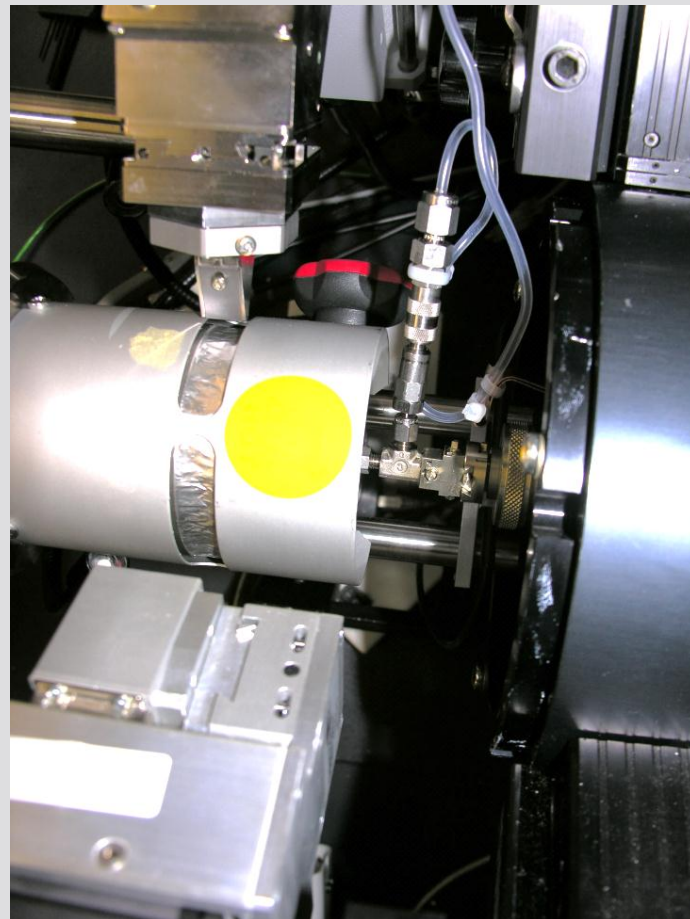


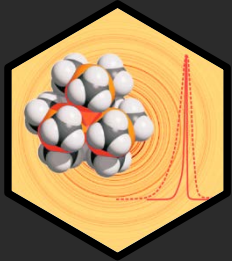


XENON ADSORPTION SITES IN MFU-4L

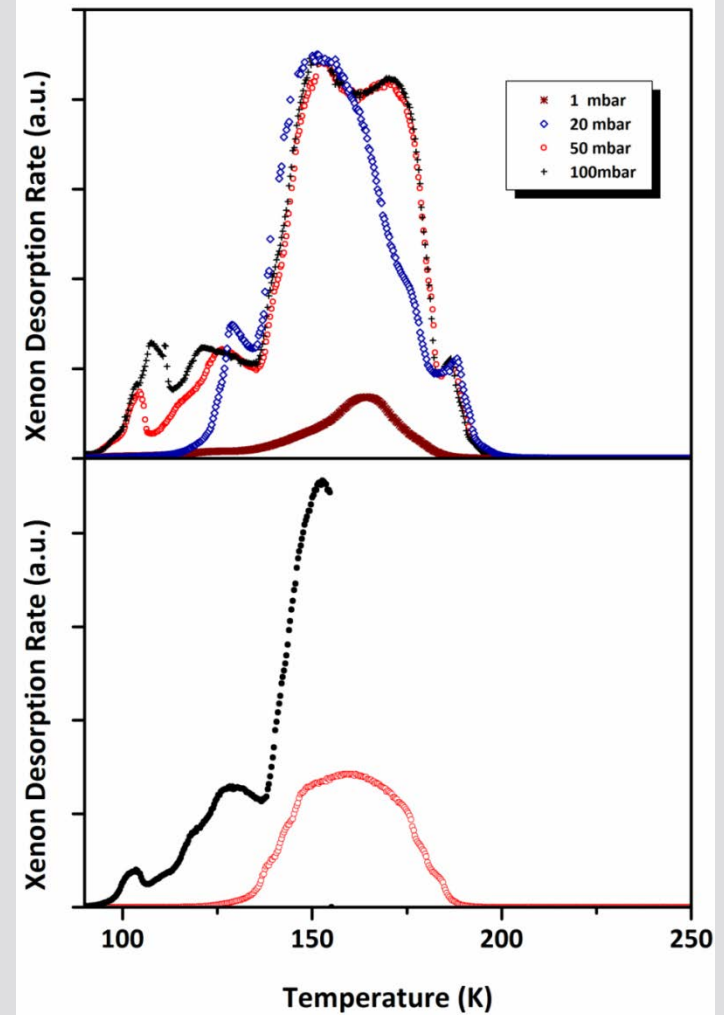
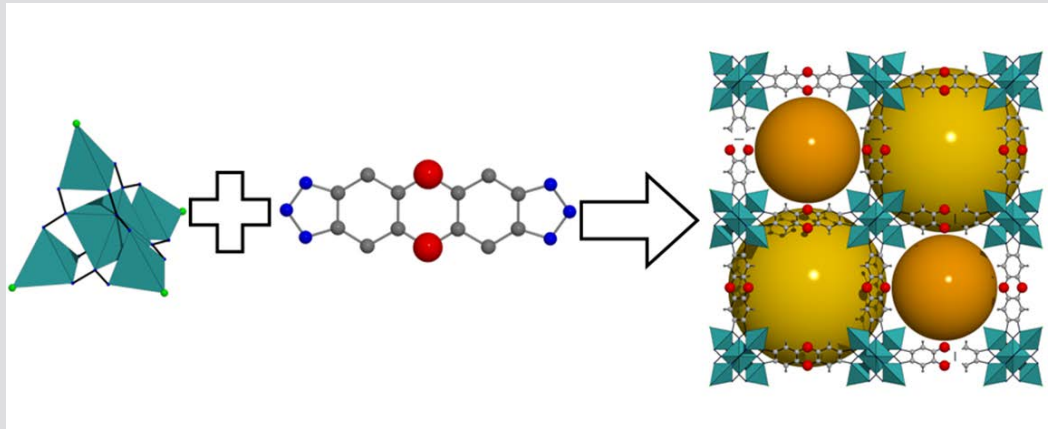
Methods:

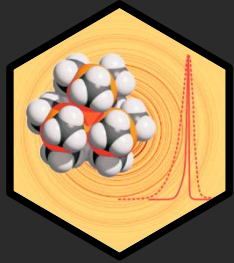
1. Thermal desorption spectroscopy
2. In-situ X-ray powder diffraction
3. Quantum mechanical calculation





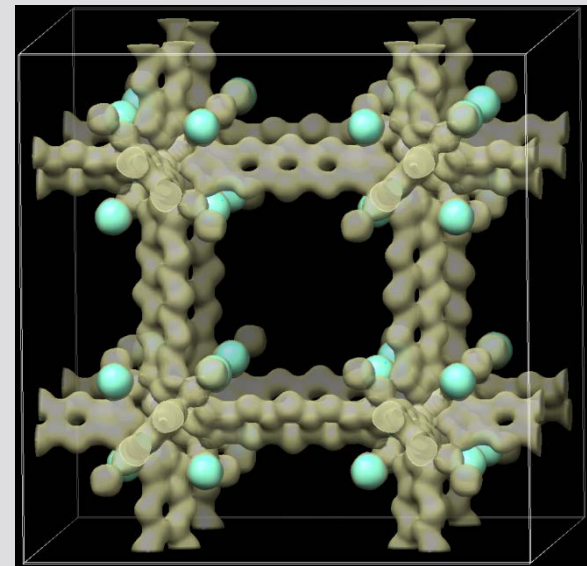
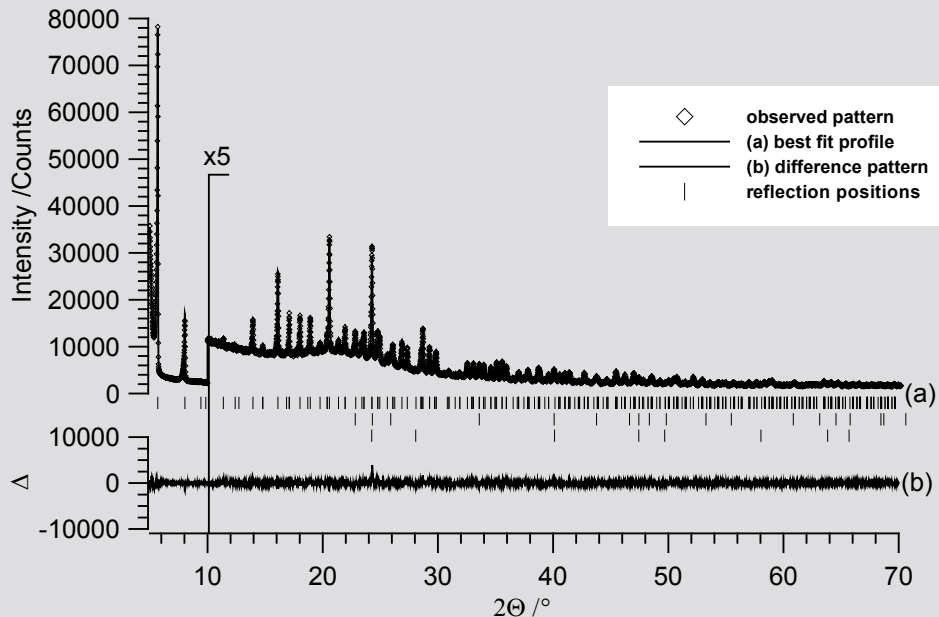
XE ADSORPTION ON MFU-4LARGE

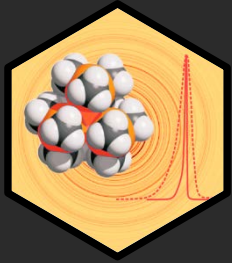




In-situ X-ray Powder Diffraction on MFU-41

- Loaded at 20mbar
- Different patterns at 110 and 150K
- **Only one adsorption site for Xenon at $2/3, 2/3, 2/3$ (32f positions)**
- **Site Occupancy: at 110K $\approx 100\%$ and at 150K $= 23\%$**





OPTIMIZED DFT CALCULATION

Xenon Adsorption Energy:

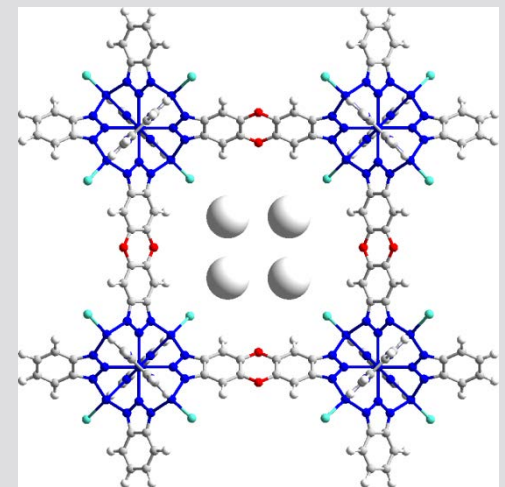
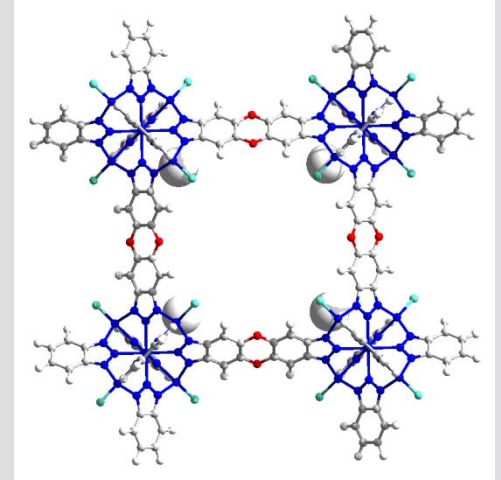
Large pore: 96 KJmole⁻¹

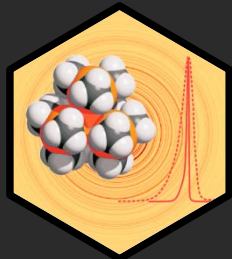
Small Pore: Equal to Xe-Xe Van der Waals force



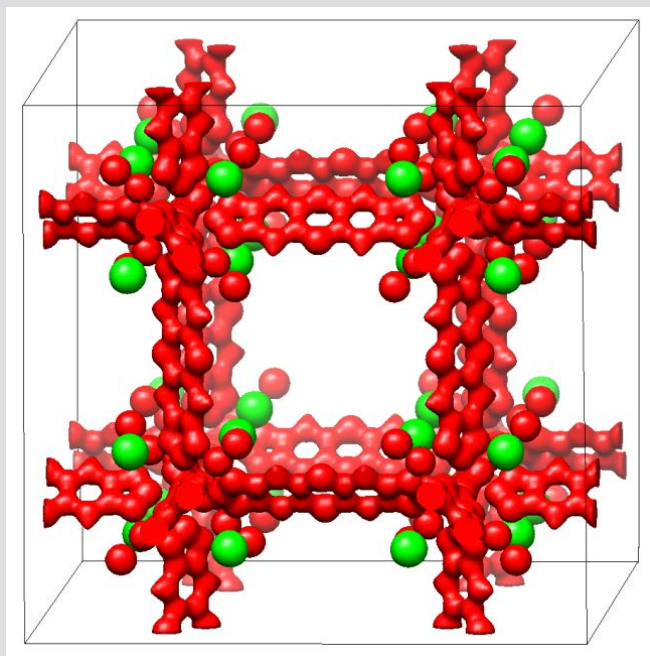
Xenon does not adsorb in the small pores!

Unpublished work

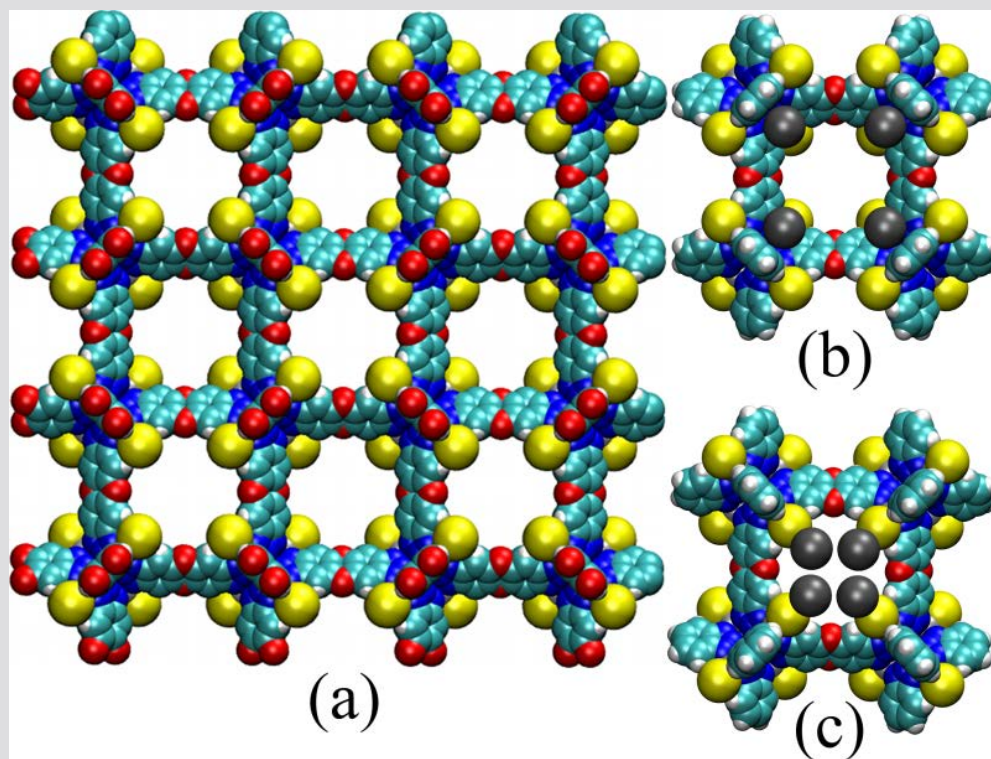




XRD AND THEORY



Xe sites from MEM-analysis



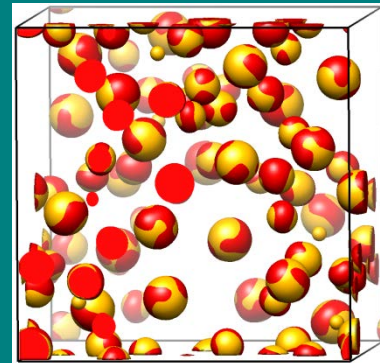
Xe sites from DFT calculations

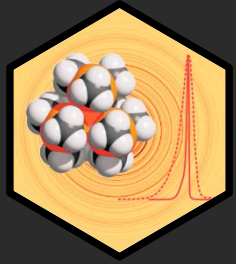
ADVANCES IN THE ANALYSIS OF IN-SITU XRPD DATA USING THE METHOD OF MAXIMUM ENTROPY (MEM)

**Oksana Magdysyuk, Ali Samy, Sander van Smaalen,
Martin Jansen, Pavel Kazin, Robert E. Dinnebier**

2010, ACTA CRYST. B, 66, 184-195.

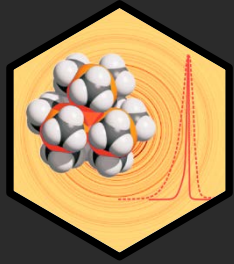
2012, Z. Kristallogr. 227, 321-333.





OUTLINE

- Introduction
- MEM versus Fourier analysis
- The effect of biasing
- Selected Examples
- Using the MEM to
 - locate intercalated atoms with low occupancy
 - analyse disorder
- Conclusions



CONCEPT OF MEM

The maximum entropy method (MEM) can be used to extract the maximum amount of information from a limited set of data by maximizing the entropy in an iterative process:

$$S = - \sum_{i=1}^{N_{pix}} \rho_i \ln \frac{\rho_i}{\rho_i^{prior}}$$

$$N_1 \times N_2 \times N_3 = N_{pix}$$

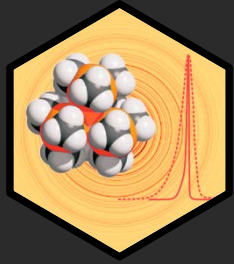
- the electron density is sampled at the points of the grid over the unit cell

$$\rho_i^{prior}$$

- the prior density

Prior density:

- 1) flat
- 2) any desirable distribution of the available electrons over the unit cell



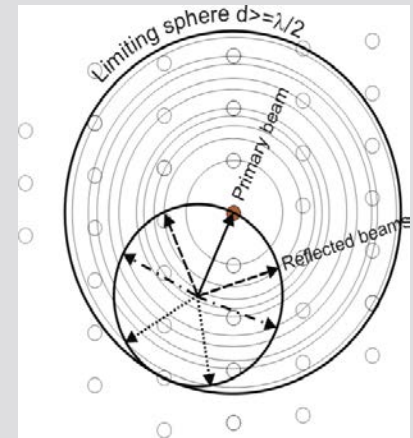
THE PROBLEM OF POWDER DIFFRACTION - SUMMARY -

Powder: Single crystal reciprocal lattice is smeared into spherical shells

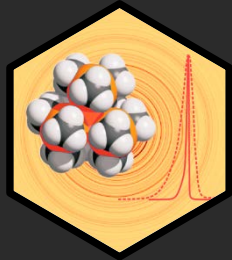
$$\rightarrow \mathbf{d} \Rightarrow |\mathbf{d}| = d \approx \sin \theta^{-1}$$

→ Different degrees of overlap:

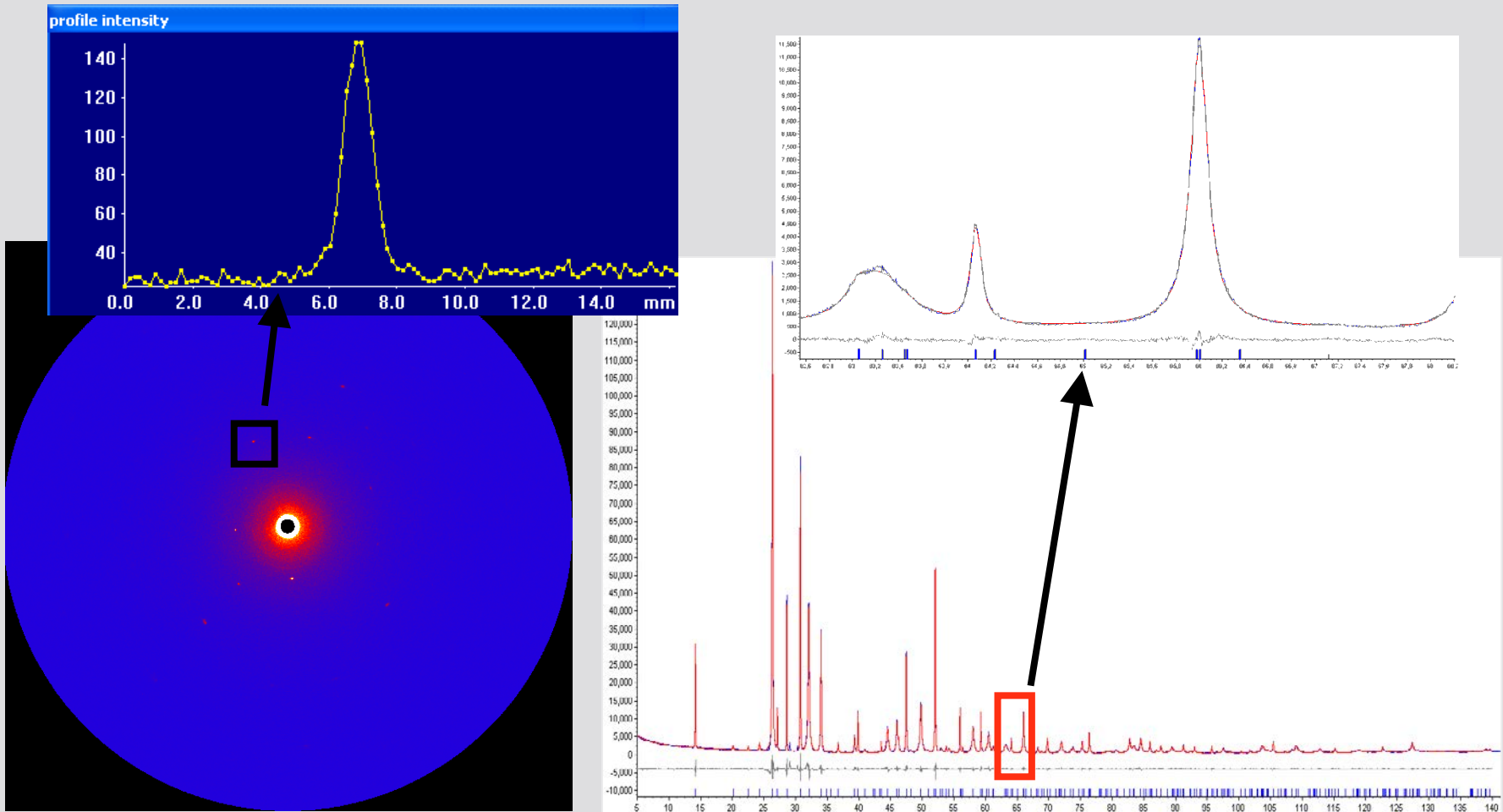
- **multiplicity** (just a factor between 2 and 48)
- **systematic** (e.g. 511, 333 in the cubic case)
- **accidental** (depending on lattice parameters and scattering angle)



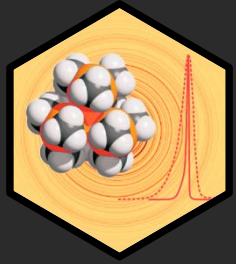
→ Not only the phases of the structure factors are unknown, but also many amplitudes



THE PROBLEM OF POWDER DIFFRACTION - AN EXAMPLE -



Example: single crystal and powder data of Pb_3O_4

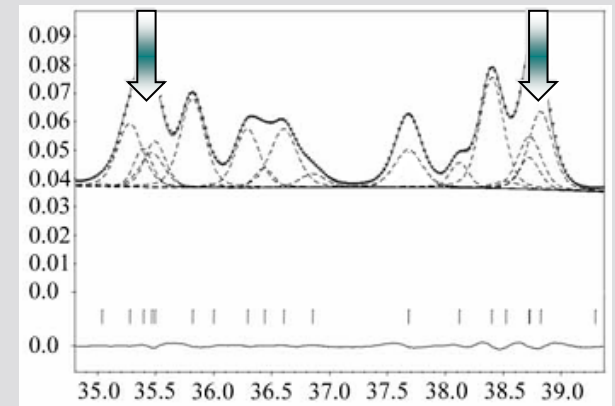


THE GOAL OF THE MEM IS TO FIND THE ELECTRON DENSITY THAT MAXIMIZES THE ENTROPY S SUBJECT TO VARIOUS CONSTRAINTS

1) $\rho_{total} = \sum_{k=1}^{N_{pix}} \rho_k$ - normalization of the electron density

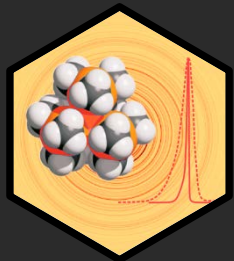
2) $C_F = -1 + \frac{1}{N_F} \sum_{i=1}^{N_F} w_i \left(\frac{|F_{obs}(H_i) - F_{MEM}(H_i)|}{\sigma(H_i)} \right)^2$

constraint C_F is based on the observed phased structure factors



3) $G^j = \sqrt{\frac{\sum_{k=1}^{N_G^j} m_k |F(H_k)|^2}{\sum m_k}}$ constraint G is the "structure factor" of a group of overlapping reflections

$$C_{FG} = -1 + \frac{1}{N_{all}} \sum_{i=1}^{N_F} w_i \left(\frac{|F_{obs}(H_i) - F_{MEM}(H_i)|}{\sigma(H_i)} \right)^2 + \frac{1}{N_{all}} \sum_{j=N_F+1}^{N_{all}} \left(\frac{|G_{obs}^j - G_{MEM}^j|}{\sigma(H_i)} \right)^2$$



BASICS OF THE MAXIMUM ENTROPY METHOD (MEM)

Maximize:

$$Q(\rho) = S(\rho) - \sum_{j=1}^{N_c} \lambda_j C_j(\rho), \quad \longrightarrow \quad \frac{\partial Q}{\partial \rho_i} = 0$$

$$\rho_i = \frac{N_{el} N_{pix}}{V} \tau_i \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i}\right) / \sum_{i=1}^{N_{pix}} \tau_i \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i}\right) \quad (3)$$

Set of N_{pix}
nonlinear equations

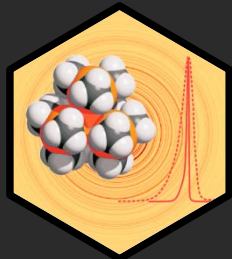
$$\tau_i^{n+1} \approx \rho_i^n \quad \text{Approximation for next iteration}$$

Iteration:

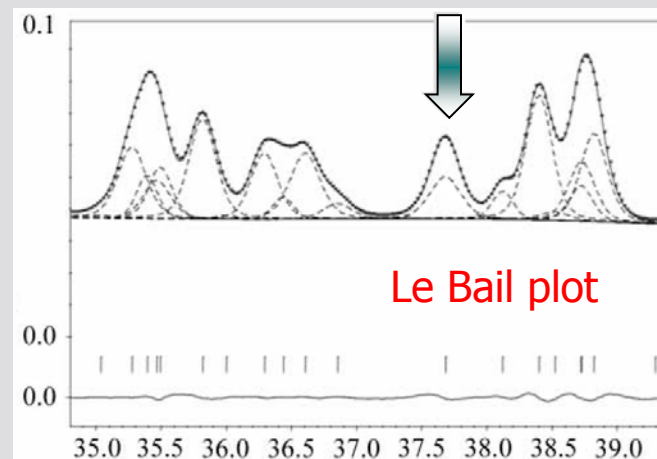
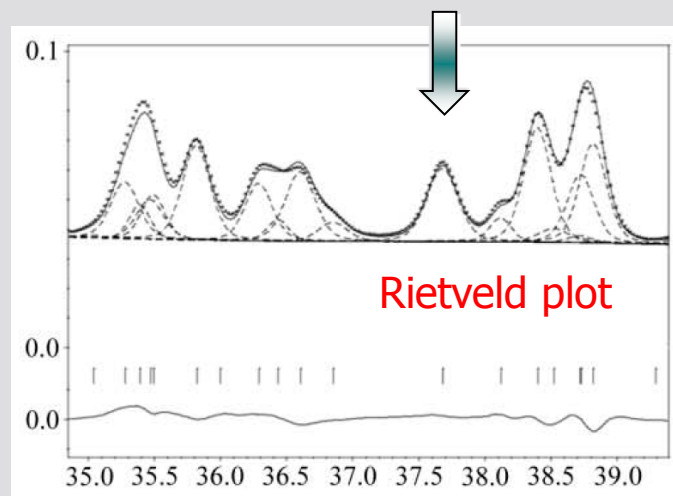
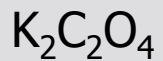
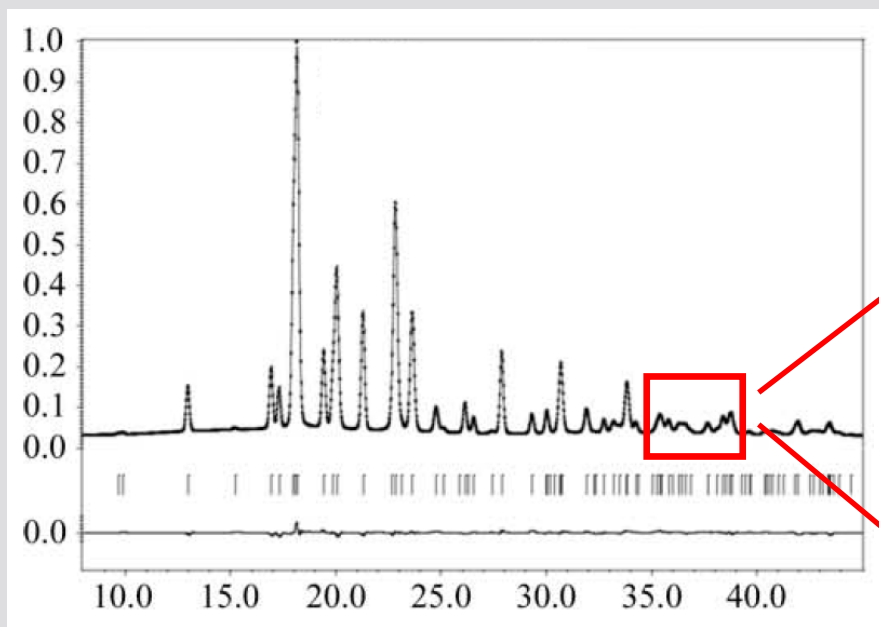
$$\rho_i^{n+1} = \frac{N_{el} N_{pix}}{V} \rho_i^{(n)} \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i} \Big|_{\rho_i^{(n)}}\right) / \sum_i \rho_i^{(n)} \exp\left(-\lambda_F \frac{\partial C_F}{\partial \rho_i} \Big|_{\rho_i^{(n)}}\right) \quad \text{Sakata \& Sato algorithm (1990)}$$

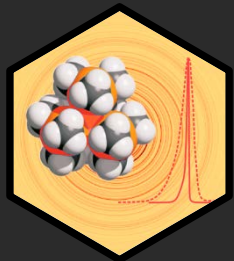
The iteration is started with $\rho_i^{(1)} = \tau_i$ and the new density $\rho_i^{(n+1)}$ is calculated from the prior density $\rho_i^{(n)}$, the value of the constraint decreases each cycle until the condition of $C_F \leq 1$ is fulfilled

ρ_i	Electron density
τ_i	Prior density
N_{pix}	no of pixels of cell volume V
N_{el}	no of electron/unit cell = F000
w	weight factor
$\sigma(H)$	standard error of F_{obs}
F_{obs}	observed structure factors
F_{MEM}	MEM structure factors
λ	Lagrange multipliers
N_c	no of constraints



POWDER DIFFRACTION: OVERLAPPING REFLECTIONS





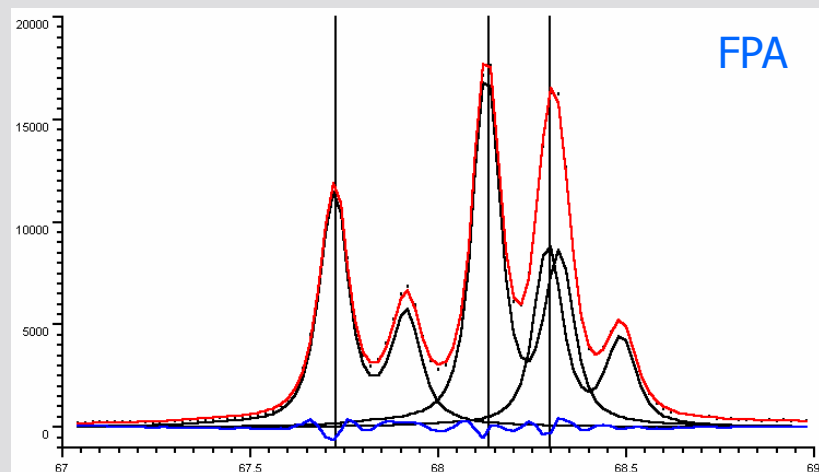
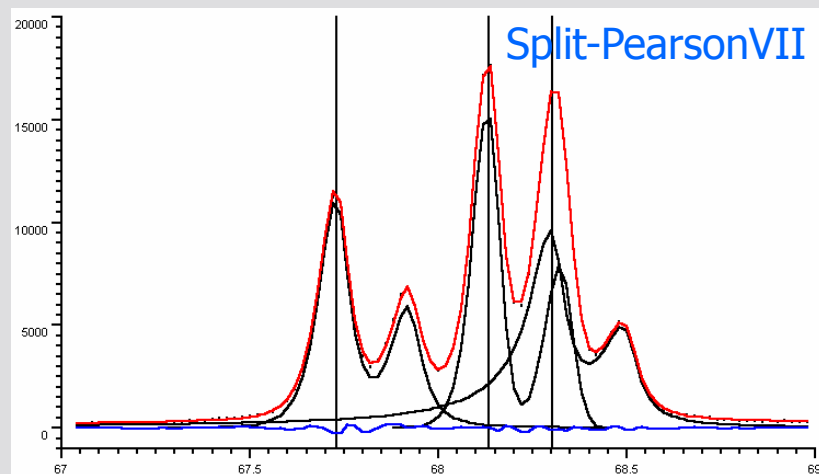
PROFILE ANALYSIS OF X-RAY POWDER PATTERN

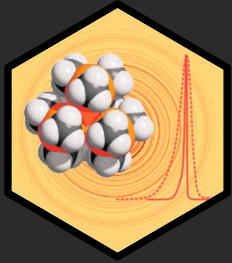
Advantages of the Fundamental Parameters Approach (FPA):

In contrast to empirical/conventional profile fitting methods, the refined numerical parameters have physical meaning

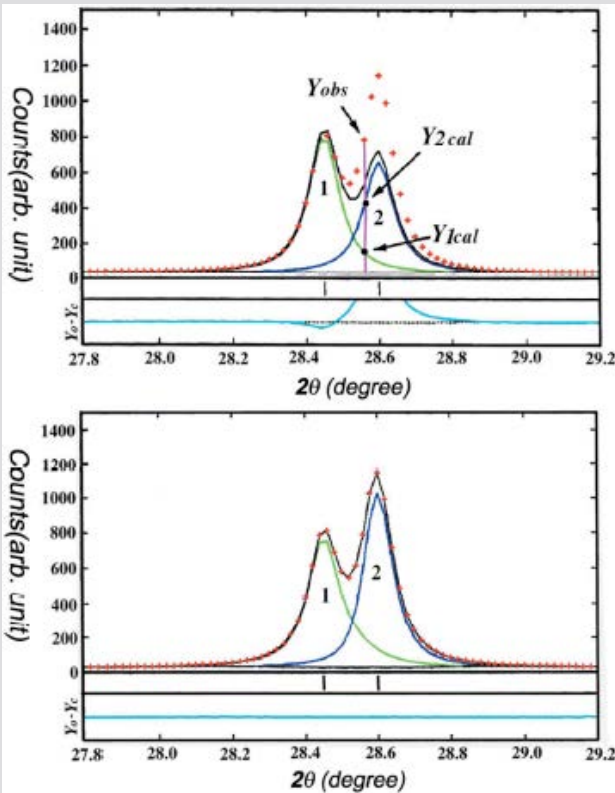
FPA greatly reduces the number of parameters refined and hence minimizes parameter correlation

Example: quartz





DECOMPOSITION OF OVERLAPPING REFLECTIONS



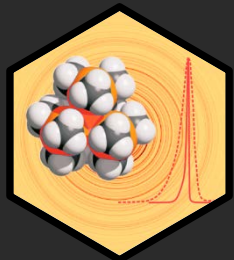
$$Y_i^{obs}(2\theta_j) = \frac{Y_i^{obs}(2\theta_j)Y_i^{calc}}{\sum_N Y_i^{calc}(2\theta_j)}$$

Rietveld refinement:

- the peak area is proportional to the square of the structure-factor amplitude
- for overlapping peaks, the contribution for a given reflection is weighted by the calculated peak contribution for that reflection divided by the sum of the calculated peak values for each contributing reflection ("F_{obs}" might be wrong)

Le Bail fit:

- there is no initial structural model
- the values of the integrated intensities are determined iteratively after each refinement cycle
- for fully overlapping peaks, intensities are partitioned equally



MEM RECONSTRUCTION OF DIFFERENT TYPES OF ELECTRON DENSITY MAPS FROM POWDER DIFFRACTION DATA

1) $F_{\text{calc}}, f^{\text{calc}}$



$\rho_{\text{calc}}^{\text{MEM}}$

Completely model biased

2) $F_{\text{obs}}, f^{\text{calc}}$



$\rho_{\text{obs}}^{\text{MEM}}$

3) $F_{\text{obs}}+G, f^{\text{calc}}$



$\rho_{\text{obs}+G}^{\text{MEM}}$

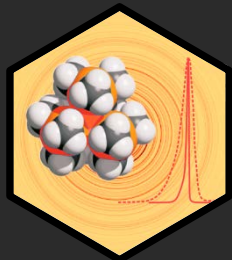
4) $F_{\text{LeBail}}+G, f^{\text{calc}}$



$\rho_{\text{LeBail}+G}^{\text{MEM}}$

Least model biased





FOURIER RECONSTRUCTION OF DIFFERENT TYPES OF ELECTRON DENSITY MAPS FROM POWDER DIFFRACTION DATA

1) $F_{\text{calc}}, f^{\text{calc}}$



$$\rho_{\text{calc}}^{\text{Fourier}}$$

Completely
model biased

2) $F_{\text{obs}}, f^{\text{calc}}$



$$\rho_{\text{obs}}^{\text{Fourier}}$$

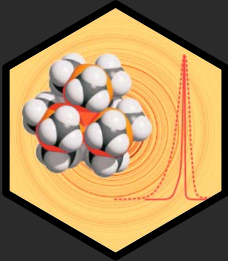
3) $F_{\text{obs}} - F_{\text{calc}}, f^{\text{calc}}$



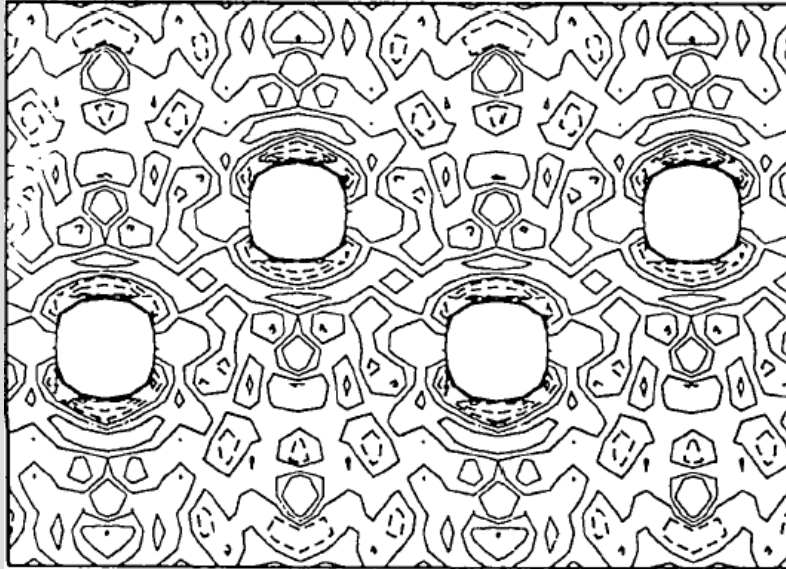
$$\rho_{\text{diff}}^{\text{Fourier}}$$

Least
model biased

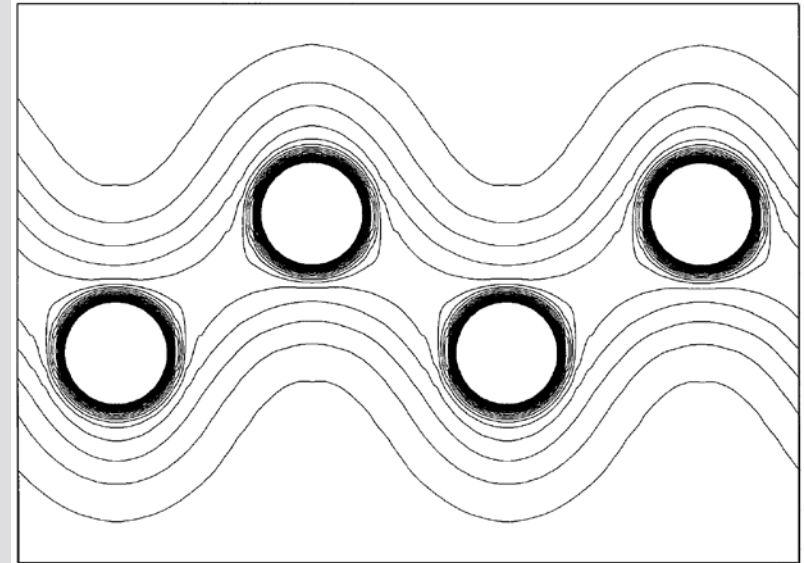




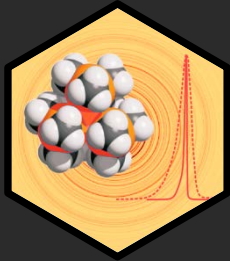
MODELED AND EXPERIMENTAL FOURIER MAPS



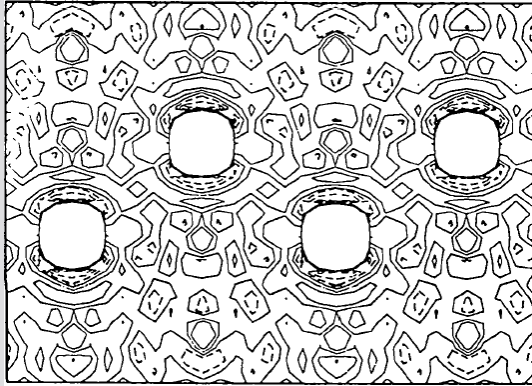
Fourier map of experimental data of silicon in the (110) plane obtained by a Fourier synthesis of the structure factors for which $\sin\theta/\lambda = 0.86 \text{ \AA}^{-1}$. Contour intervals are at $0.1 e/\text{\AA}^3$ (cutoff level $1.6 e/\text{\AA}^3$).



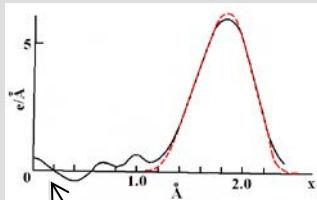
Fourier map of model data of silicon in the (110) plane obtained by a Fourier synthesis of the structure factors for which $\sin\theta/\lambda = 5.5 \text{ \AA}^{-1}$. Contour intervals are at $0.1 e/\text{\AA}^3$ (cutoff level $2.5 e/\text{\AA}^3$).



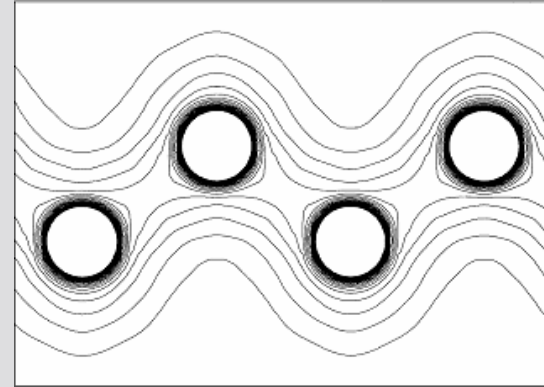
DETERMINATION OF THE ELECTRON-DENSITY DISTRIBUTION FOURIER MAP ↔ MEM MAP



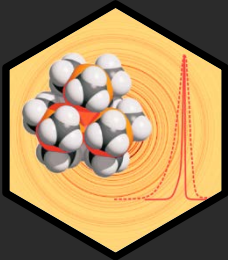
- a perfect Fourier map would require a complete set of structure factors up to at least $\sin \theta/\lambda = 5.0 \text{ \AA}^{-1}$
- allows the location of missing atoms
- may show strong peaks that do not correspond to atoms



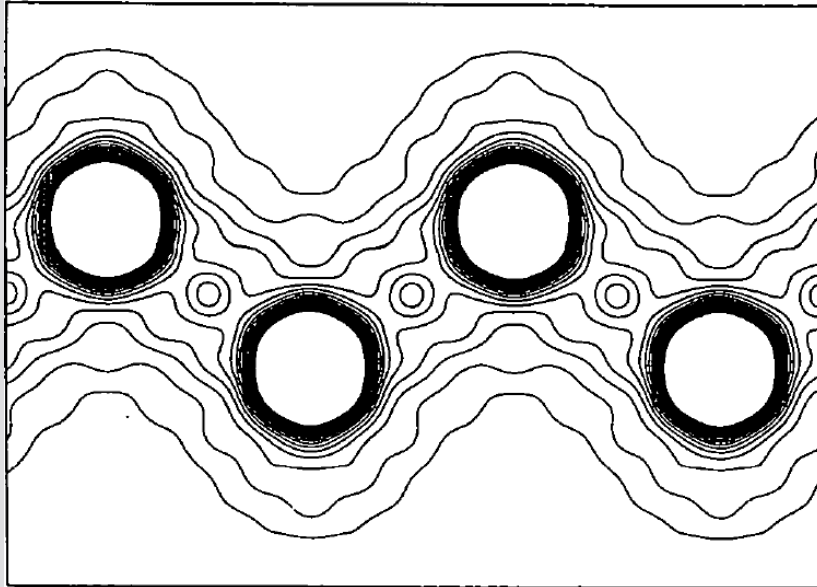
series termination errors



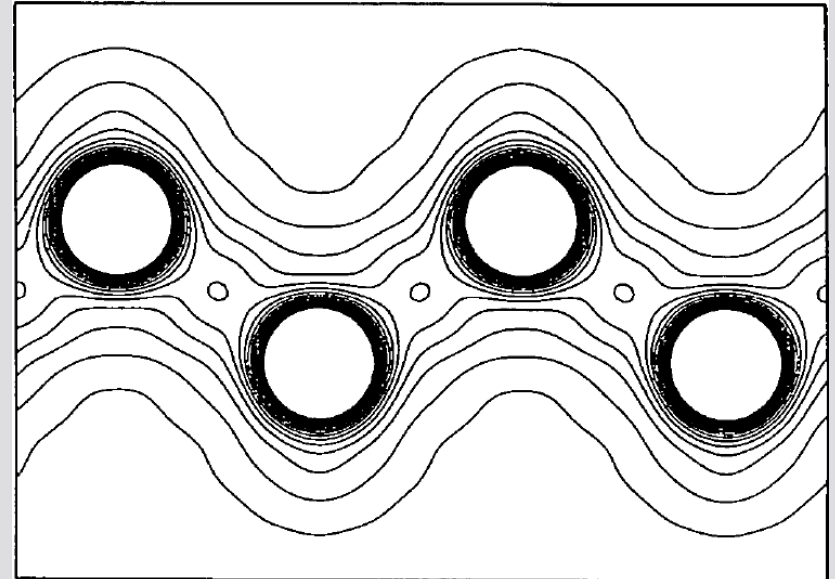
- allows to incorporate prior information
- does not involve numerical Fourier transform
=> no series termination errors.
- can successfully work with structure factors from powder data with a resolution of $\sin \theta/\lambda \approx 0.6 \text{ \AA}^{-1}$.
- allows the location of missing atoms
- peaks that do not correspond to atoms in the current model always have the density of the noise level



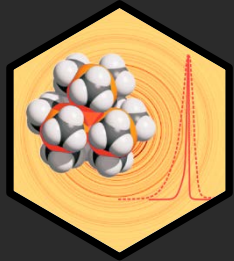
MEM CALCULATIONS WITH FLAT PRIOR AND DIFFERENT RESOLUTION $\sin\theta/\lambda$



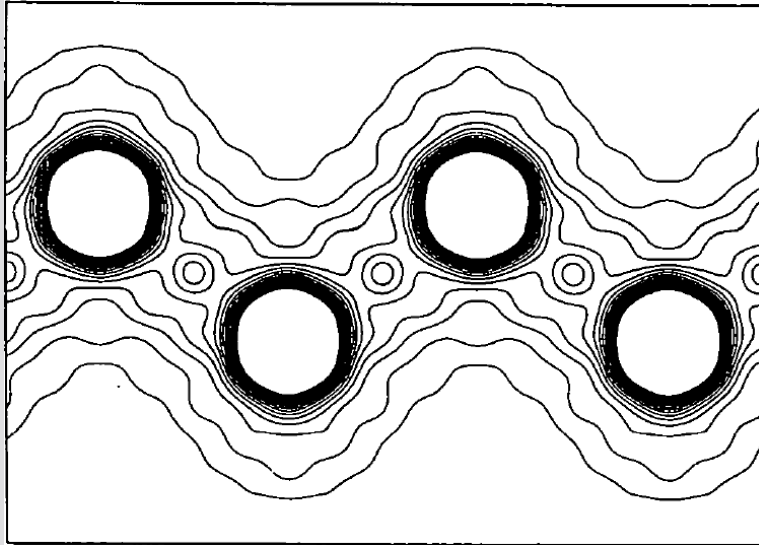
MEM map of experimental data of silicon in the (110) plane obtained using the flat prior and the calculated structure factors for which $\sin\theta/\lambda = 0.86 \text{ \AA}^{-1}$. Contour intervals are at $0.1 e/\text{\AA}^3$ (cutoff level $2.5 e/\text{\AA}^3$).



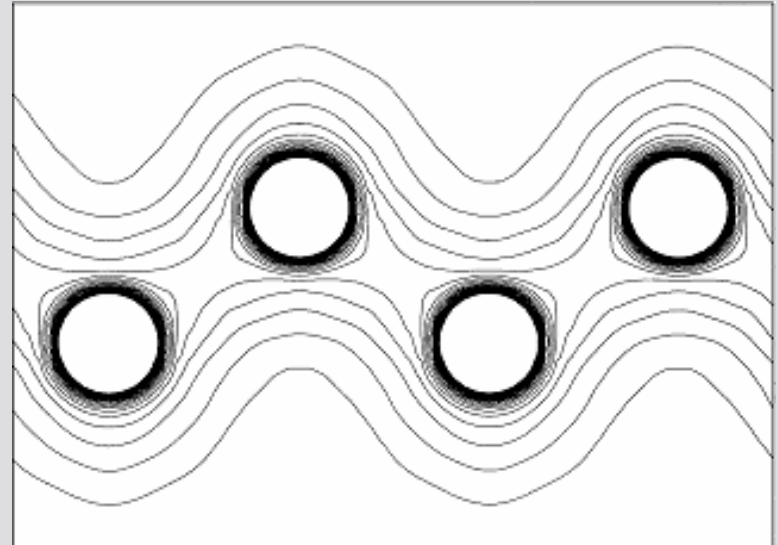
Same as left side but $\sin\theta/\lambda = 1.04 \text{ \AA}^{-1}$.
→ just 2 more Bragg reflections !!!



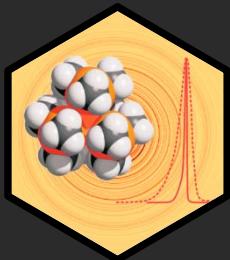
PRIOR DENSITY IN MEM CALCULATIONS



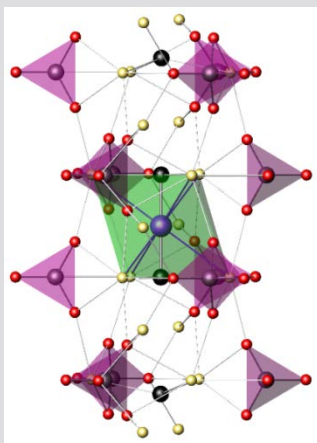
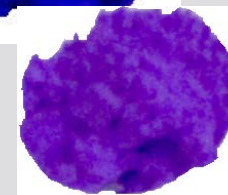
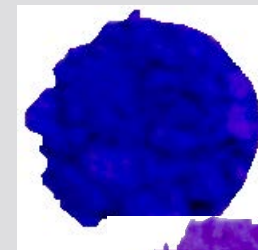
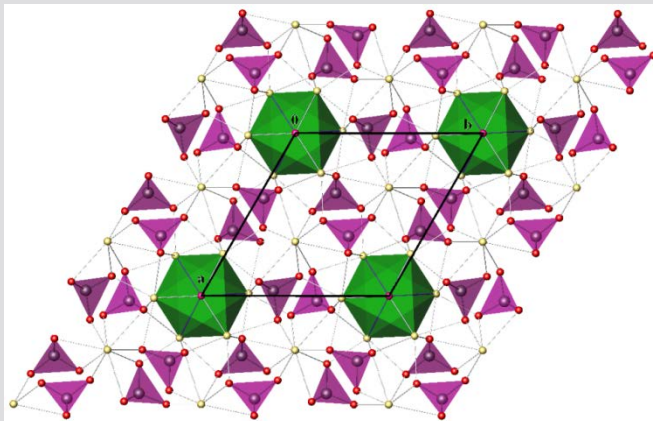
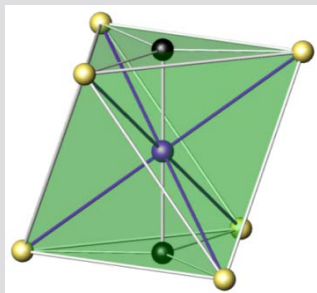
MEM map of experimental data of silicon in the (110) plane obtained using the flat prior and the calculated structure factors for which $\sin\theta/\lambda = 0.86 \text{ \AA}^{-1}$. Contour intervals are at $0.1 e/\text{\AA}^3$ (cutoff level $2.5 e/\text{\AA}^3$).



Same as left side but using the procrystal density (ISAM).



ELECTRON-DENSITY DISTRIBUTION IN APATITES WITH INTERCALATED METAL ATOMS

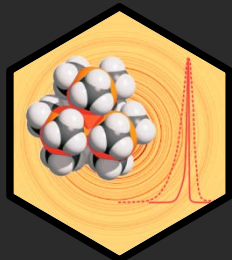


Molecular formula	$\text{Sr}_5(\text{PO}_4)_3\text{Cu}_x\text{O}_{2x}(\text{OH})_{1-2x}$
Space group	$P6_3/m$ (176)
Z	2
$a / \text{\AA}$	9.77
$c / \text{\AA}$	7.28

Karpov, A.S.; Nuss, J.; Jansen, M.; Kazin, P.E.; Tretyakov, Yu.D. // Solid State Science. 2003. V.5. P.1277-1283.

Kazin, P.E.; Karpov, A.S.; Jansen, M.; Nuss, J.; Tretyakov, Yu.D. // Z. Anorg. Allg. Chem. 2003. V. 629. P. 344-352.

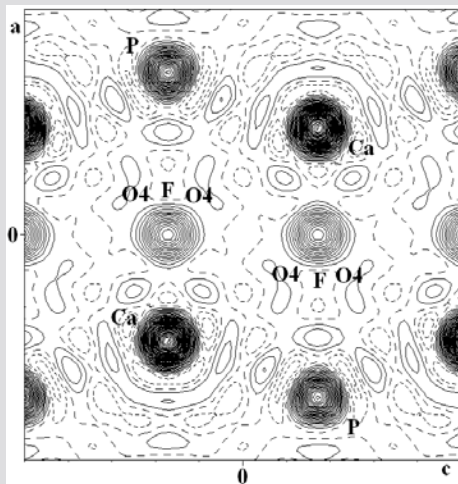
Kazin, P.E.; Gazizova, O.R.; Karpov, A.S.; Jansen, M.; Tretyakov, Yu.D. // Solid State Science. 2007. V. 9. P. 82-87.



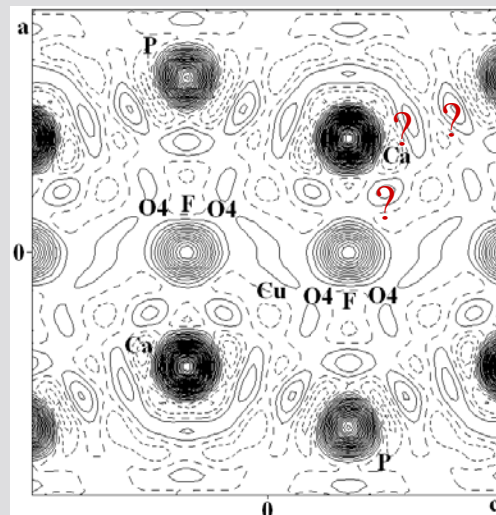
LOCALIZATION OF MISSING ATOMS IN THE INCOMPLETE CRYSTAL STRUCTURE OF APATITE



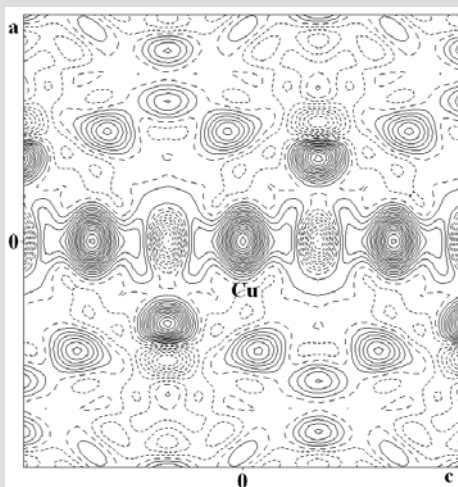
Fourier
 ρ_{calc}



Fourier
 ρ_{obs}



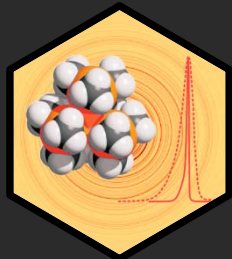
Fourier
 ρ_{diff}



There is no Fourier map that allows to locate the Cu atoms unambiguously

X-ray laboratory data
 $\sin\theta/\lambda = 0.55 \text{ \AA}^{-1}$

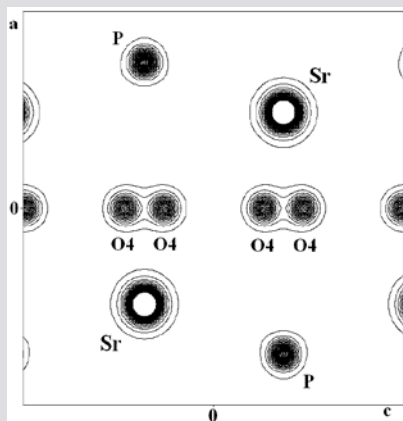
Contour levels: from 1 to 50 $\text{e}/\text{\AA}^3$, step 1 $\text{e}/\text{\AA}^3$



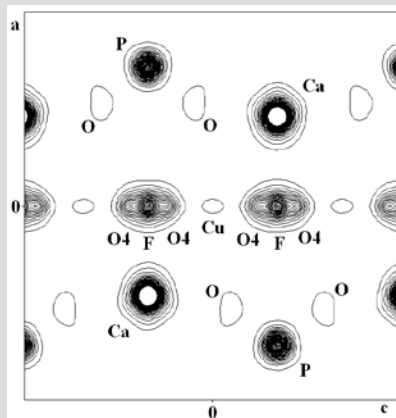
LOCALIZATION OF MISSING ATOMS IN THE INCOMPLETE CRYSTAL STRUCTURE OF APATITE



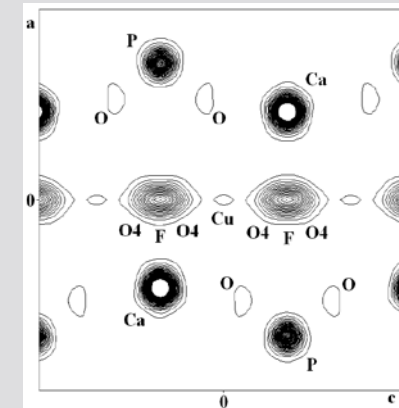
ρ_{calc}^{MEM}



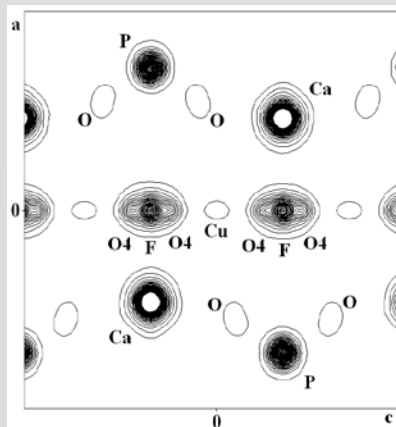
ρ_{obs}^{MEM}
procrystal
density



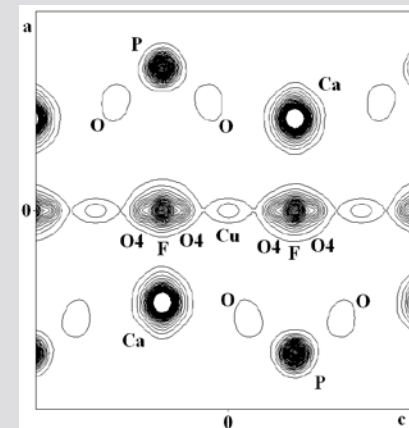
ρ_{obs}^{MEM}
flat prior



ρ_{obs+G}^{MEM}
procrystal
density



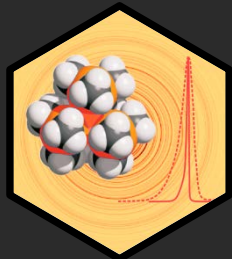
$\rho_{LeBail+G}^{MEM}$
procrystal
density



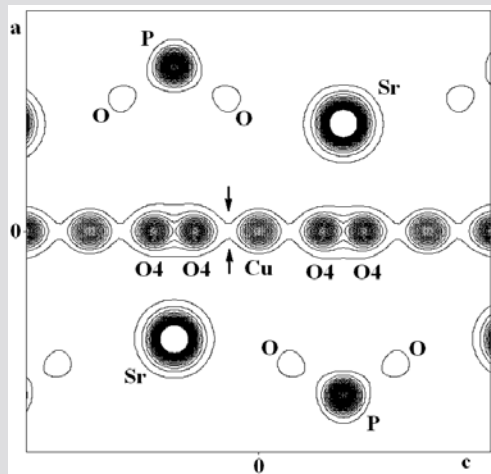
X-ray laboratory data
 $\sin\theta/\lambda = 0.55 \text{ \AA}^{-1}$

Contour levels: from 1 to 50 e/Å³, step 1 e/Å³

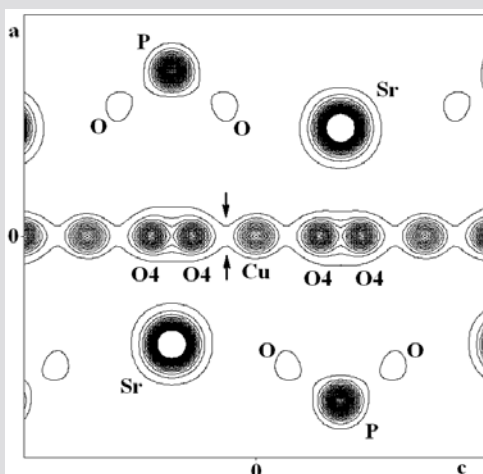
DETERMINATION OF THE EDD OF INTERCALATED COPPER ATOMS USING MEM MAPS



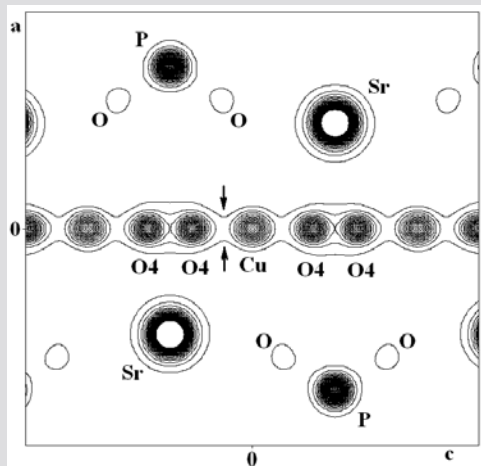
1



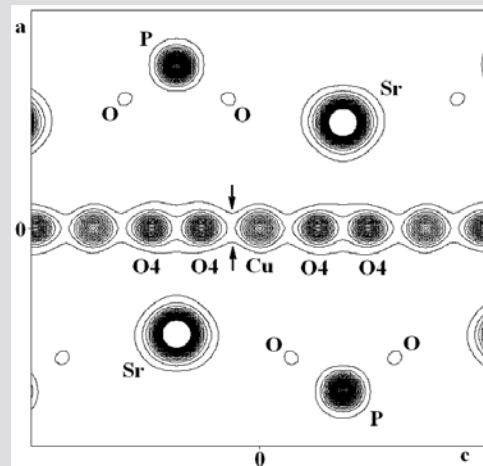
2



3



4



X-ray **laboratory** data
 $\sin\theta/\lambda = 0.55 \text{ \AA}^{-1}$

All compounds have the same
content of Cu = 0.1, but
different amount of peroxide

1 – 1.9 % of peroxide

2 – 3 % of peroxide

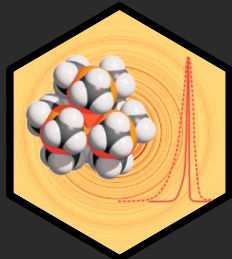
3 – 15 % of peroxide

4 – >15 % of peroxide

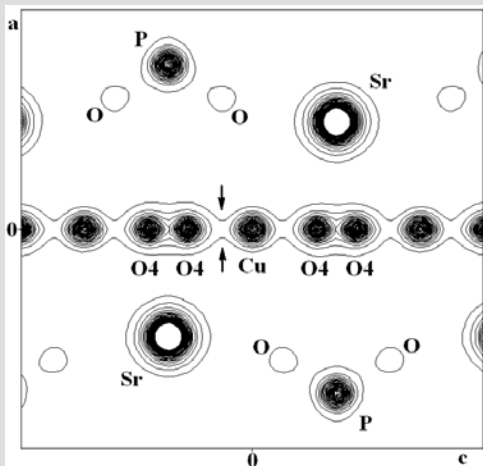
The distribution of the electron
density is distorted due to the
increasing content of peroxide
molecules

Contour levels: from 1 to 50 $e/\text{\AA}^3$, step 1 $e/\text{\AA}^3$

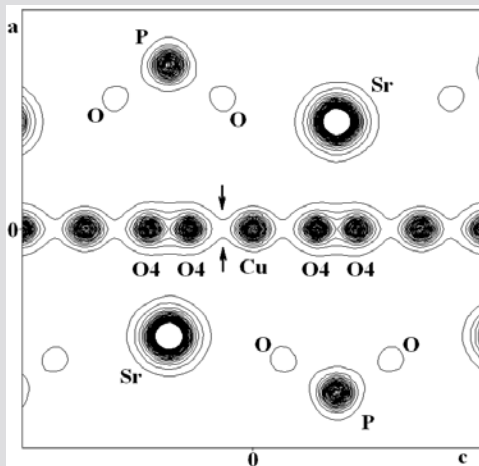
DETERMINATION OF THE EDD OF INTERCALATED COPPER ATOMS USING MEM MAPS



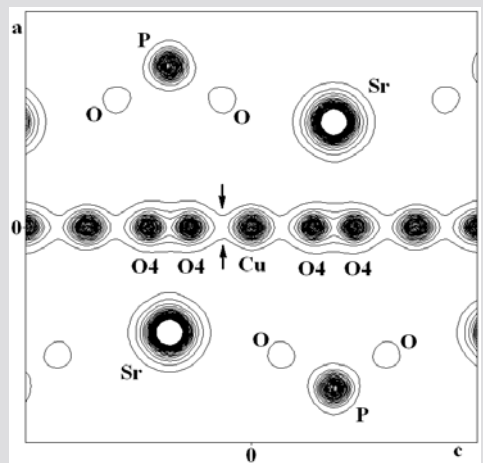
1



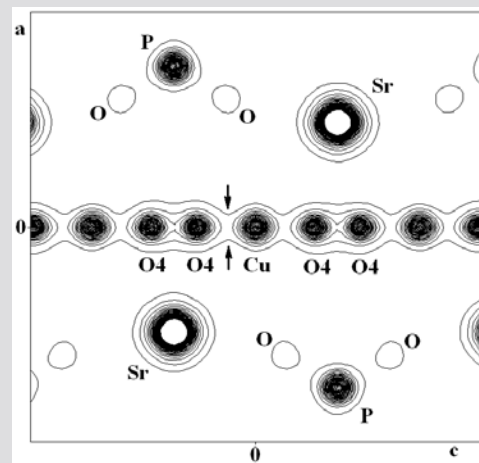
2



3



4



Synchrotron data $\sin\theta/\lambda = 0.55 \text{ \AA}^{-1}$

All compounds have the same content of Cu = 0.1, but different amount of peroxide

1 – 1.9 % of peroxide

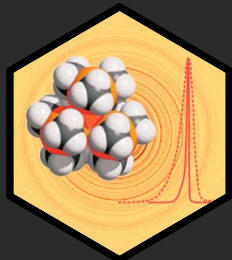
2 – 3 % of peroxide

3 – 15 % of peroxide

4 – >15 % of peroxide

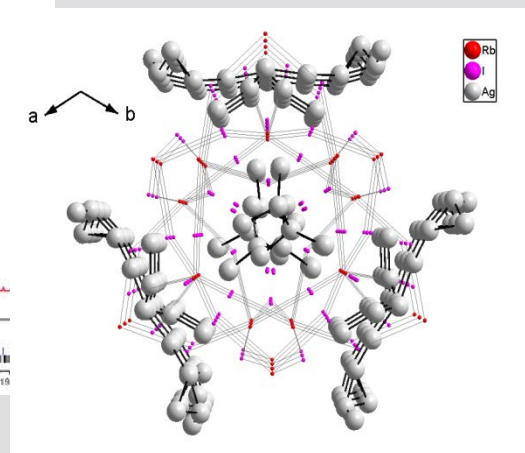
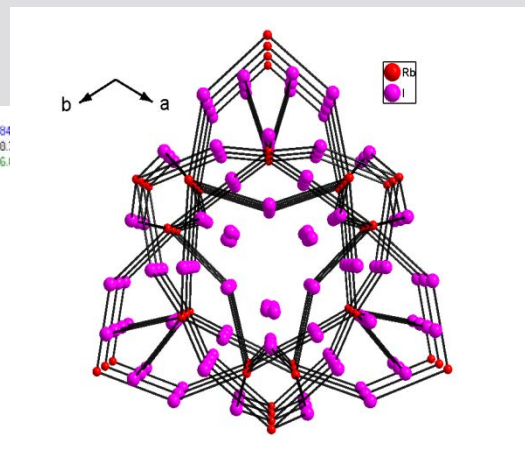
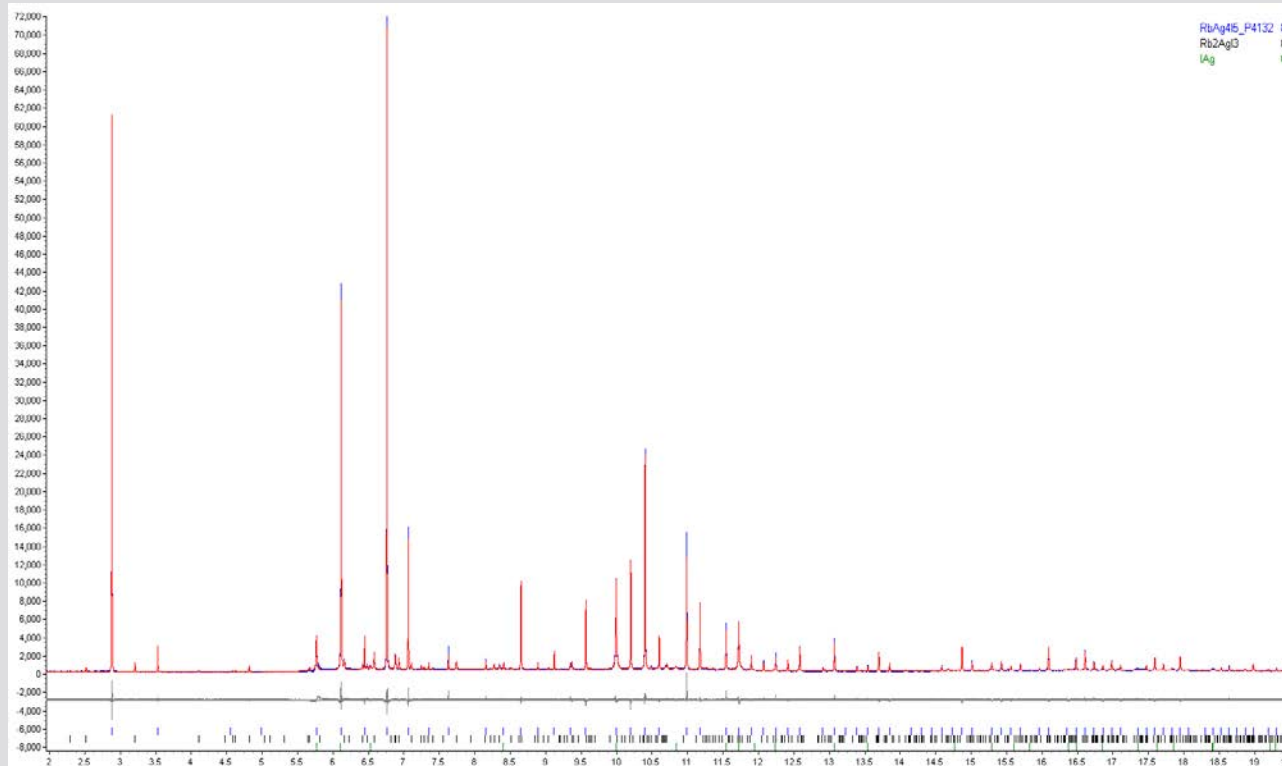
The distribution of the electron density is distorted due to the increasing content of peroxide molecules

Contour levels: from 1 to 50 e/\AA^3 , step 1 e/\AA^3

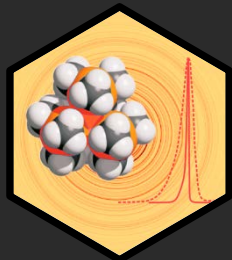


MEM MAP FROM POWDER ↔ EDD FROM SINGLE CRYSTAL

\square -RbAg₄I₅ ionic conductor P4₁32 $a=11.2500 \text{ \AA}$



Parameters of the Rietveld refinement:
 $R_{wp} = 7.13\%$, $R_p = 5.23\%$, $R_I = 3.10\%$, $GooF = 1.64$

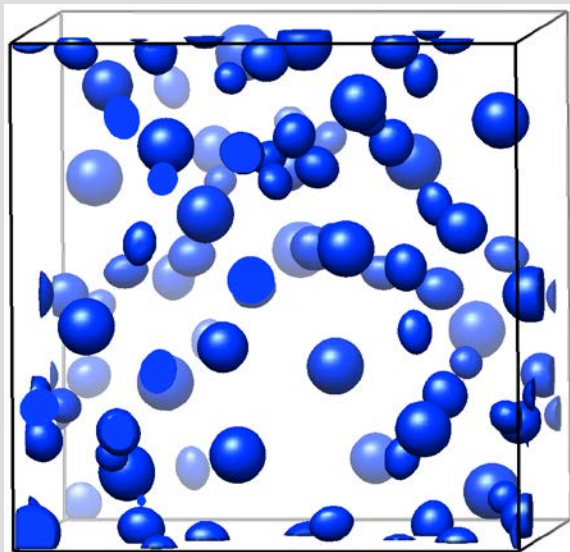
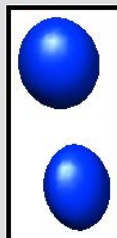


SINGLE CRYSTAL EDD

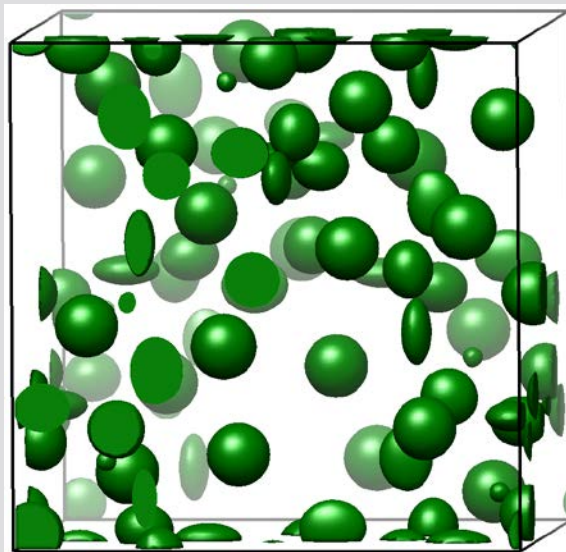
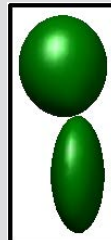
↔ RIETVELD FROM POWDER
↔ MEM MAP FROM POWDER

Ag1

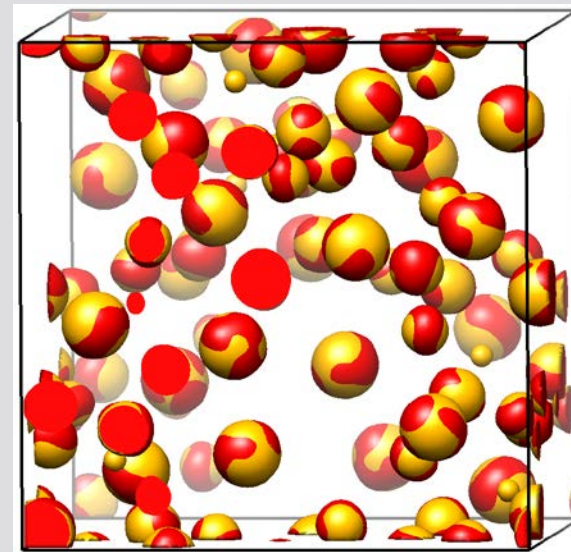
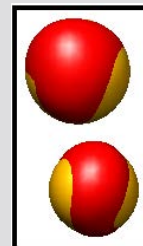
Ag2



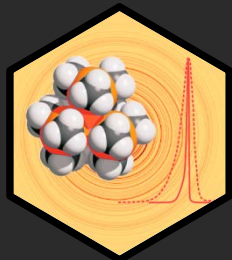
single crystal EDD



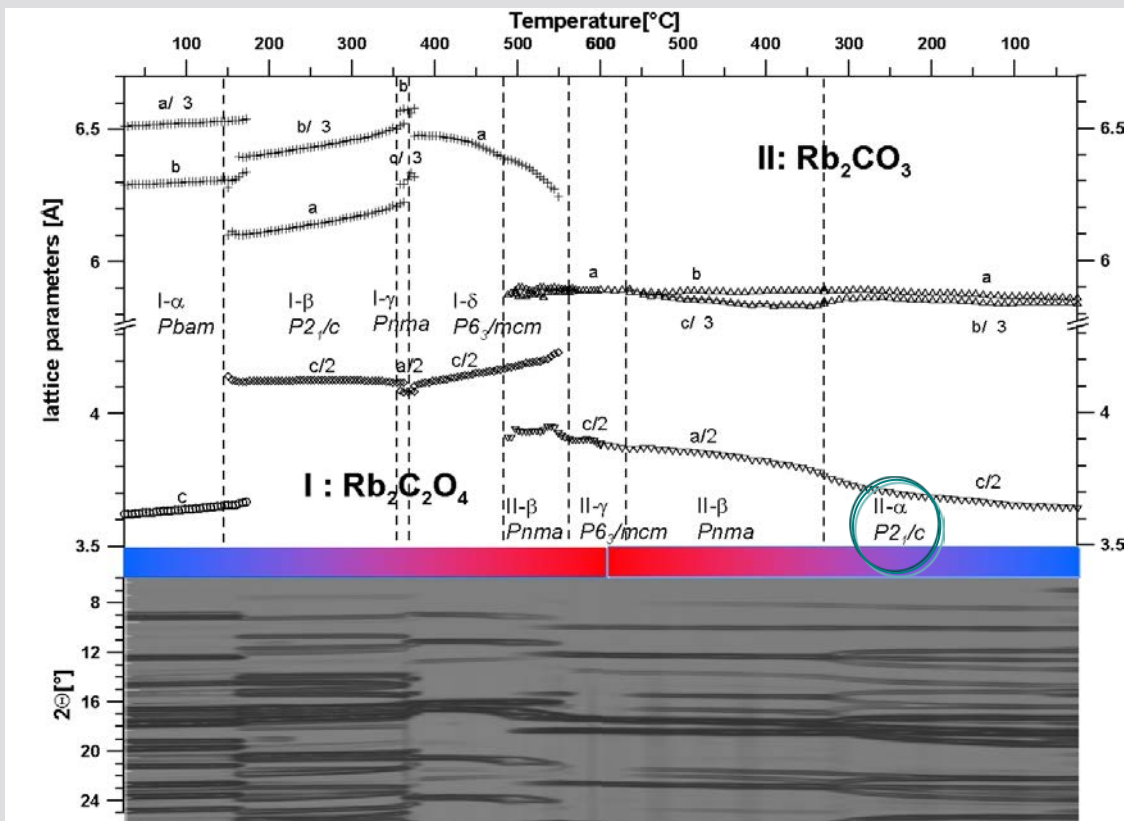
anisotropic ADPs from
Rietveld refinement



red – MEM reconstructed ED
yellow – ISAM

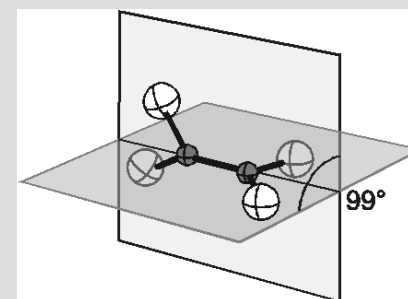


THE HIGH TEMPERATURE PHASES OF RUBIDIUM OXALATE BY IN SITU POWDER DIFFRACTION



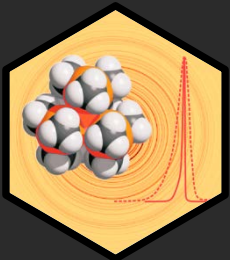
T: 25°-450°-25°

200 scans in 3½ hrs



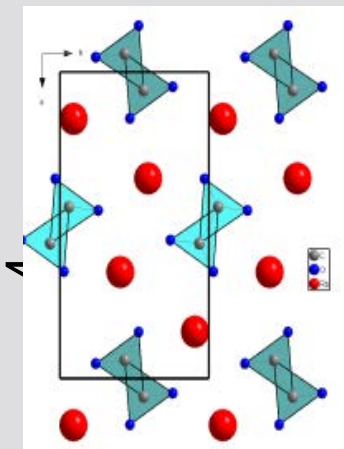
R. E. Dinnebier, S. Vensky, M. Jansen, and J. Hanson, Crystal Structures of and Topological Aspects on the High Temperature Phases and the Decomposition Products of the Alkali Oxalates $M_2[C_2O_4]$, $M=(K, Rb, Cs)$, 2005, *Chemistry, a European Journal*, 11, 1119 – 1129.

A. Samy, R. E. Dinnebier, S. van Smaalen, and M. Jansen, The Maximum Entropy Method and Charge Flipping, a powerful combination to visualize the true nature of structural disorder from *in situ* X-ray powder diffraction data. (2010) *Acta Cryst. B*.

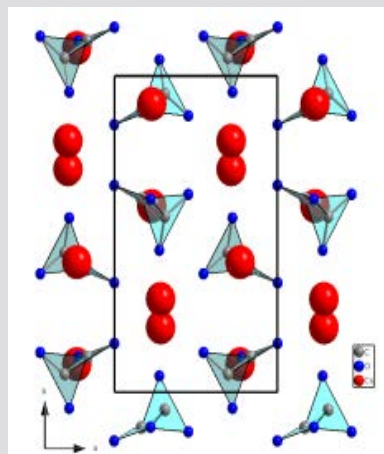


THE HIGH TEMPERATURE PHASES AND DECOMPOSITION PRODUCTS OF RUBIDIUM OXALATE

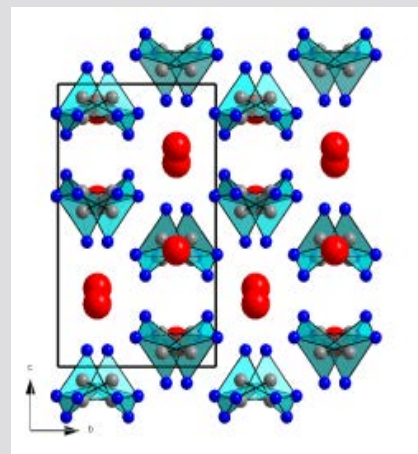
$\text{Rb}_2\text{C}_2\text{O}_7$



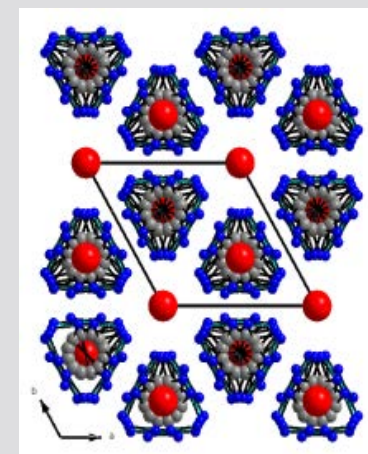
$Pbam$ \rightarrow



$P21_1/c$ \leftrightarrow

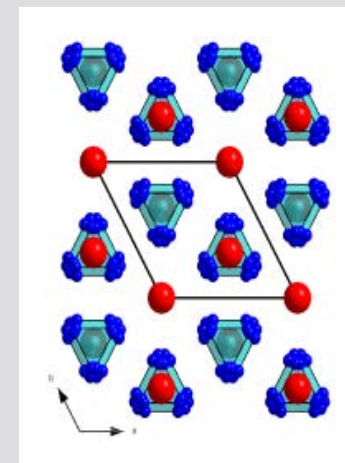
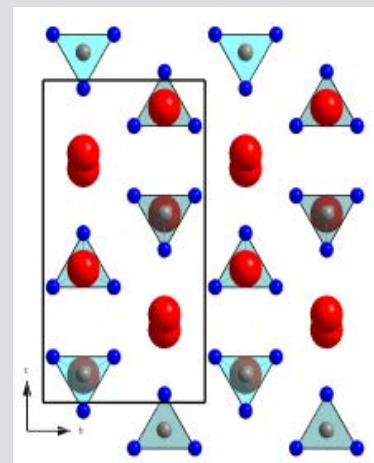
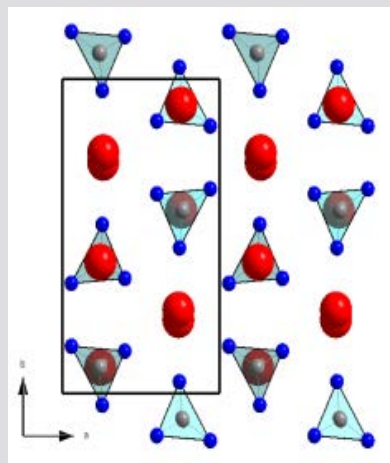


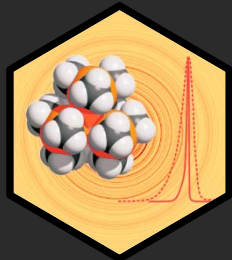
$Pnma$ \leftrightarrow



$P6_3/mmc$

Rb_2CO_3





RESULTS OF MEM-CALCULATIONS BASED ON MODEL-RIETVELD-REFINEMENT

Prior: IAM

Strong-biased

MEM-Fcalc:

All F biased

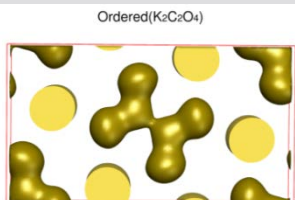
MEM-Fobs:

F_{obs} partially biased

MEM-Fobs-G-constr.

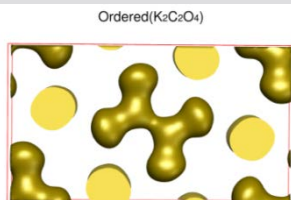
Less biased

δ -K₂C₂O₄
(*Pbam*)
ordered
T = 295 K



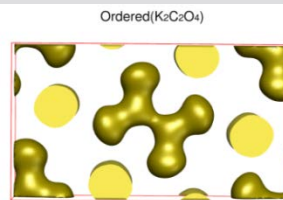
Procrystal-prior; iso-level=1.0, V/A=30.10/142.2

(a1)



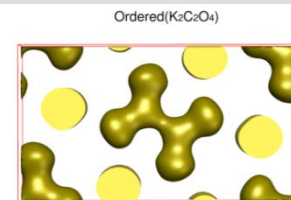
MEM-Fcalc; iso-level=1.0, V/A=29.41/139.7

(a1)



MEM-Fobs; iso-level=1.0; V/A=29.69/140.3

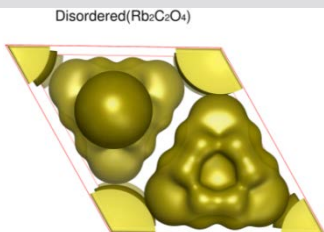
(a2)



MEM-Fobs+G; iso-level=1.0, V/A=29.51/139.6

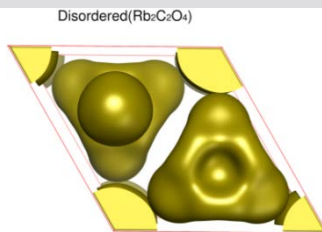
(a3)

α -Rb₂C₂O₄
(*P6₃/mmc*)
disordered
T = 683 K



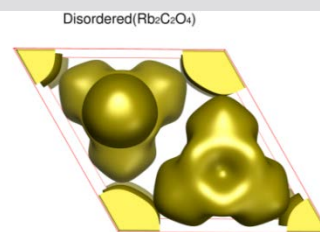
Procrystal-Prior; iso-level=0.6; V/A= 95.56/250.93

(b1)



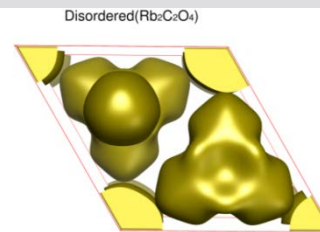
MEM-Fcalc; iso-level=0.6; V/A= 90.07/237.8

(b1)



MEM-Fobs; iso-level=0.6; V/A= 87.40/226.9

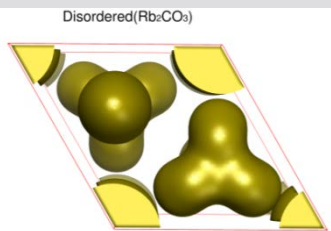
(b2)



MEM-Fobs+G; iso-level=0.6; V/A= 87.62/226.6

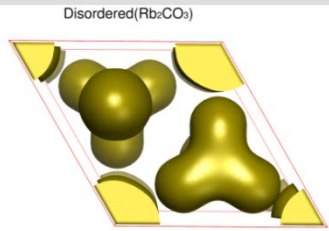
(b3)

α -Rb₂CO₃
(*P6₃/mmc*)
disordered
T = 860 K



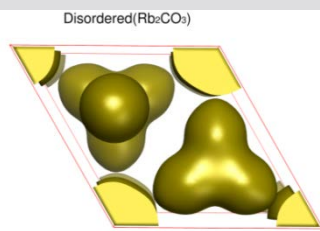
Procrystal-prior; iso-level=0.6; V/A= 63.85/195.5

(c1)



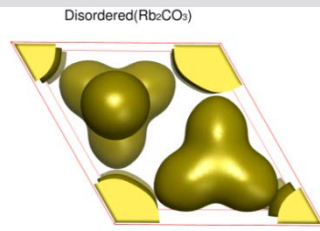
MEM-Fcalc; iso-level=0.6; V/A= 61.57 / 191.2

(c1)



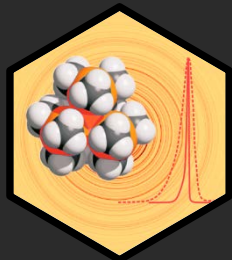
MEM-Fobs; iso-level=0.6; V/A= 61.47/189.6

(c2)



MEM-Fobs+G; iso-level=0.6; V/A= 61.57/189.9

(c3)



CHARGE FLIPPING + MEM

$F_{\text{LeBail}+G}, f^{\text{Rietveld}}$



$\rho_{\text{LeBail}+G}^{\text{MEM}}$

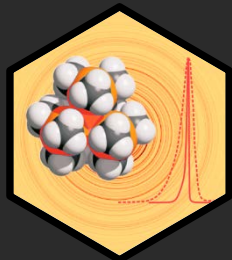
$F_{\text{LeBail}+G}, f^{\text{Charge Flipping}}$



$\rho_{\text{LeBail}+G}^{\text{MEM}}$

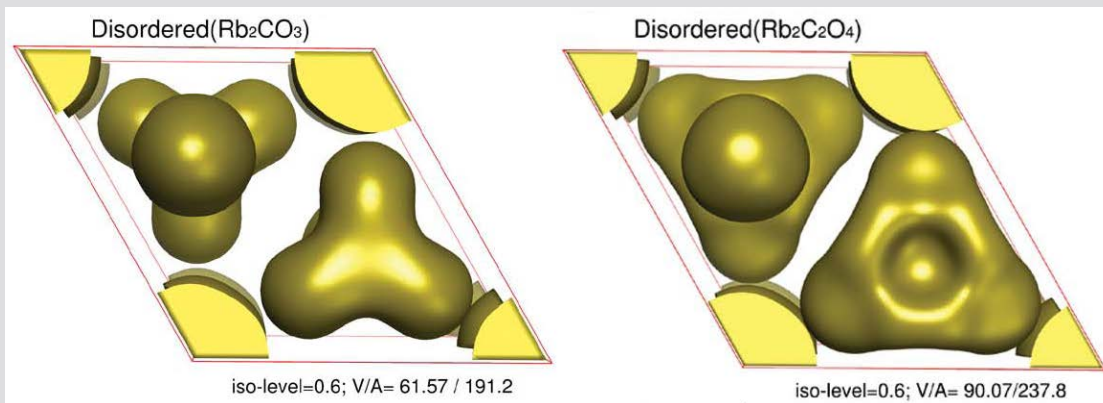
- 1) structure-factor amplitudes are obtained by Le Bail fit
- 2) structure-factor phases are determined by charge flipping with histogram matching

Completely *ab-initio* electron-density distribution!!!

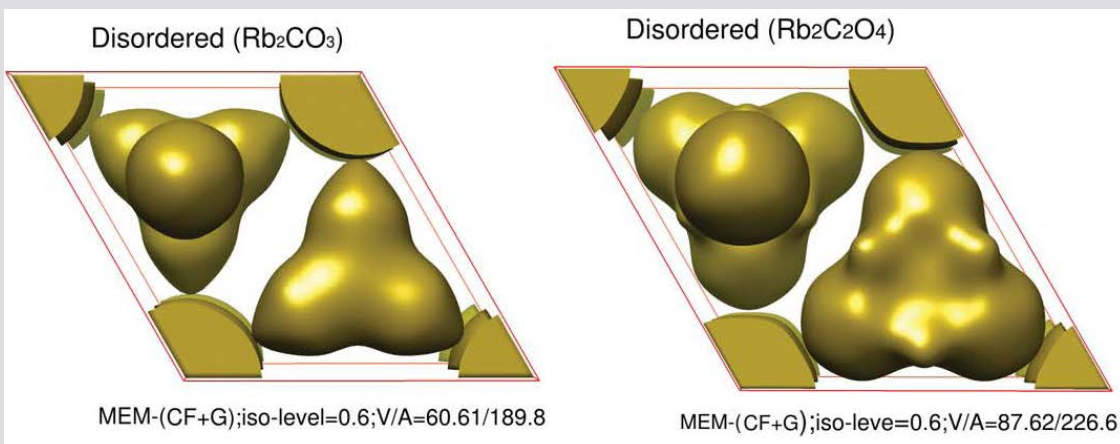


CHARGE FLIPPING + MEM

Distribution of the electron densities from Rietveld refinement:



Distribution of the electron densities from Charge Flipping + MEM:



research papers

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Maximum entropy method and charge flipping, a powerful combination to visualize the true nature of structural disorder from *in situ* X-ray powder diffraction data

In a systematic approach, the ability of the Maximum Entropy Method (MEM) to reconstruct the most probable electron density of highly disordered crystal structures from X-ray powder diffraction data was evaluated. As a case study, the ambient temperature crystal structures of disordered α - $\text{Rb}_2[\text{C}_2\text{O}_4]$ and α - $\text{Rb}_2[\text{CO}_3]$ and ordered β - $\text{K}_2[\text{C}_2\text{O}_4]$ were investigated in detail with the aim of revealing the 'true' nature of the apparent disorder. Different combinations of F (based on phased structure factors) and G constraints (based on structure-factor amplitudes) from different sources were

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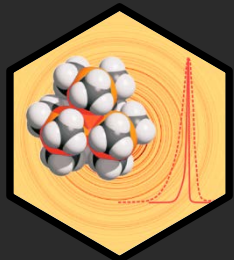
321

Reconstructions of electron density by the Maximum Entropy Method from X-ray powder diffraction data based on incomplete and complete crystal structure models: a case study of apatites with different intercalated metal atoms

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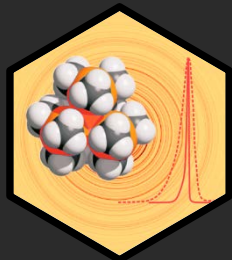
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SOME CONCLUSIONS

- ❖ New detectors and better X-ray sources allow to fully track structural changes in the sub-second regime.
- ❖ New software for data reduction (e.g. Powder3D-IP), visualization (e.g. Powder-3D) and sequential/parametric WPPF refinement (e.g. Topas 5.0) allow to handle (semi-)automatically huge numbers of powder patterns.
- ❖ Rietveld refinement and MEM allow for the investigation of advanced structural features (inaccessible to Fourier analysis) like : disorder, diffusion pathways in ionic conductors or distribution of electron density.
- ❖ The MEM is very well suited to locate even small amounts of intercalated atoms in crystal structures from XRPD
- ❖ Complex structure determination from XRPD by the combination of CF and MEM is possible.



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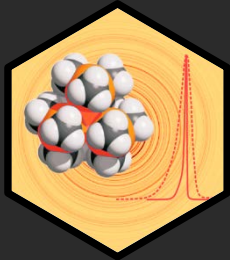
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