

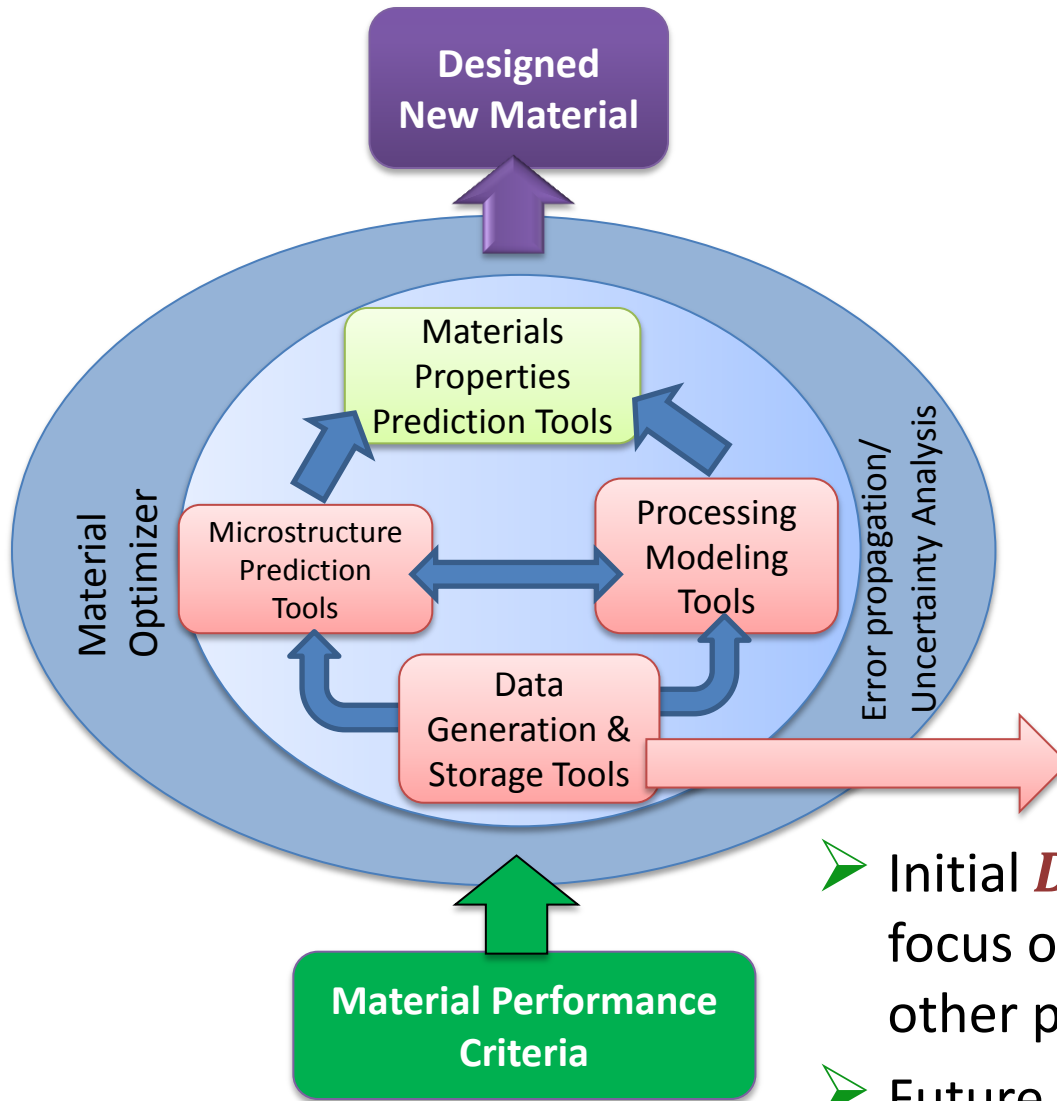
# MGI: Putting an End to Alloy Oops:\*

## NEED DATA

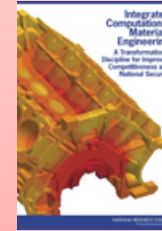
C. E. Campbell, U.R. Kattner, E. Lass,  
Metallurgy Division, NIST  
And  
Laura Bartolo, Kent State University

\*Doug Foxvog, ITL

# Materials Data for the MGI

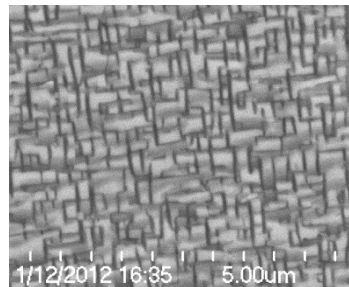
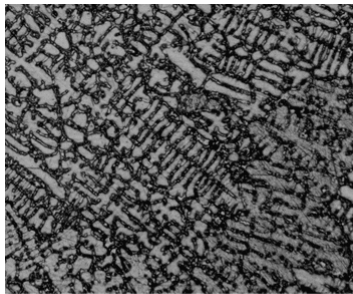
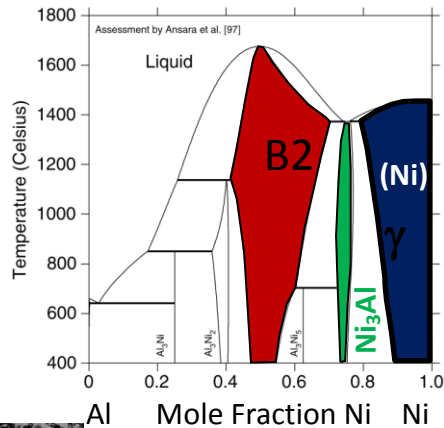
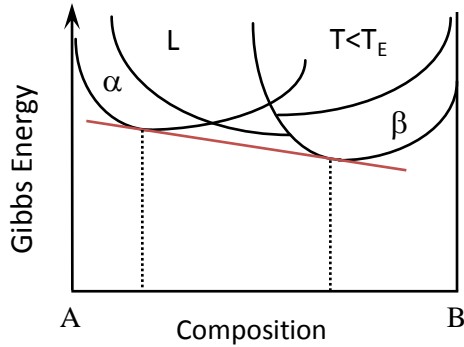


- Need to develop and curate materials information databases is well established.



- Initial *Data & Tools* development will focus on thermodynamics, kinetics and other phase-based material properties.
- Future work will build on this experience and expand into other areas.

# Why Start with Thermodynamics, Kinetics and Phase-based Material Properties



As cast  
microstructure

Annealed  
microstructure

Gibbs energy

$$G = g(T, P, N_i)$$

Entropy

$$S = -\left(\frac{\partial G}{\partial T}\right)_{P, N_i}$$

Enthalpy

$$H = G - T\left(\frac{\partial G}{\partial T}\right)_{P, N_i}$$

Heat capacity

$$C_P = -\left(\frac{\partial^2 G}{\partial T^2}\right)_{P, N_i}$$

Chemical potential

$$\mu_i = \left(\frac{\partial G}{\partial N_i}\right)_{P, T, N_{j \neq i}}$$

Volume

$$V = -\left(\frac{\partial G}{\partial P}\right)_{T, N_i}$$

Thermal expansion

$$\alpha = \frac{1}{V} \left(\frac{\partial^2 G}{\partial P \partial T}\right)_{N_i}$$

Isothermal compressibility

$$\kappa = -\frac{1}{V} \left(\frac{\partial^2 G}{\partial P^2}\right)_{T, N_i}$$

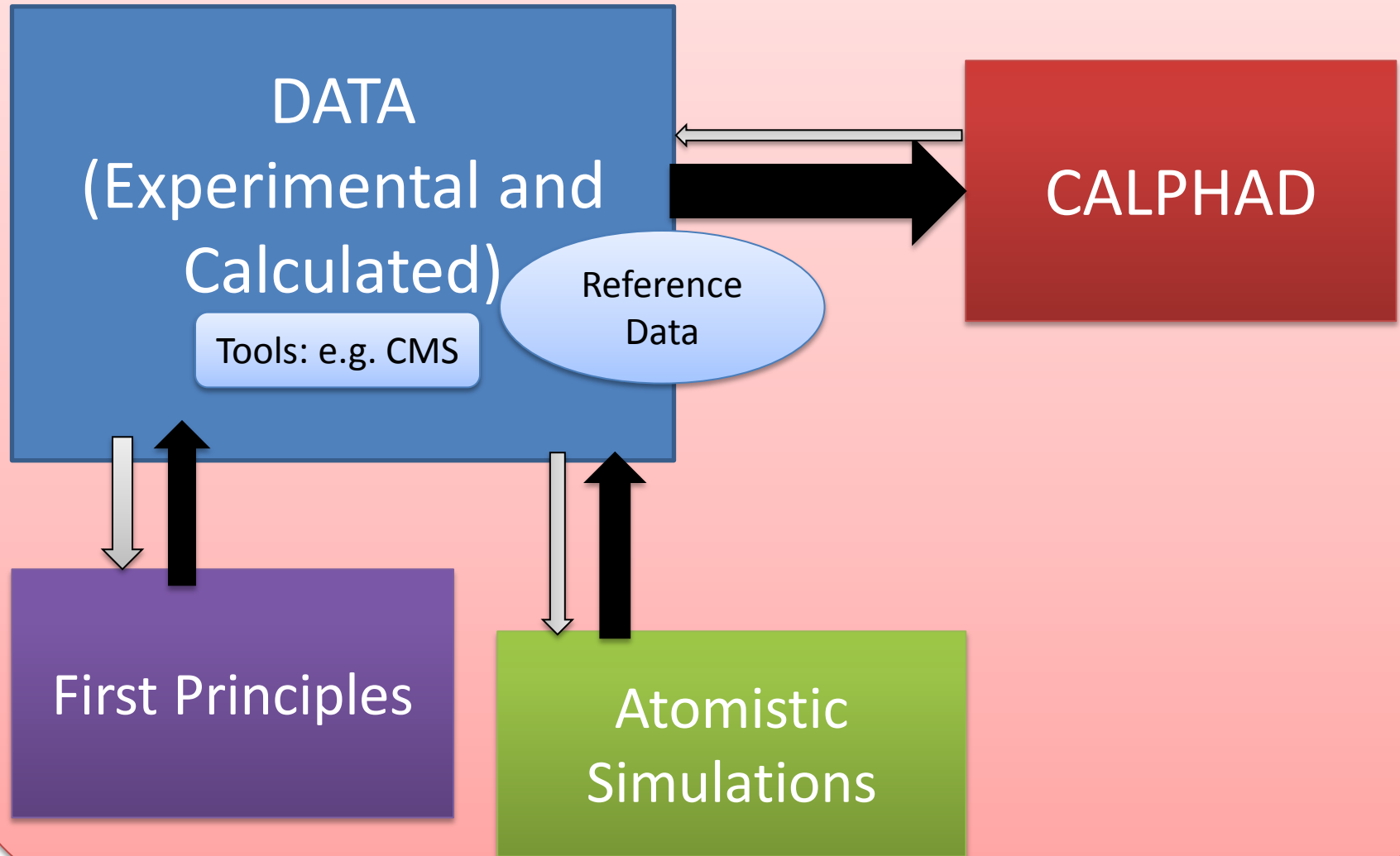
Bulk modulus

$$K = \frac{1}{\kappa}$$

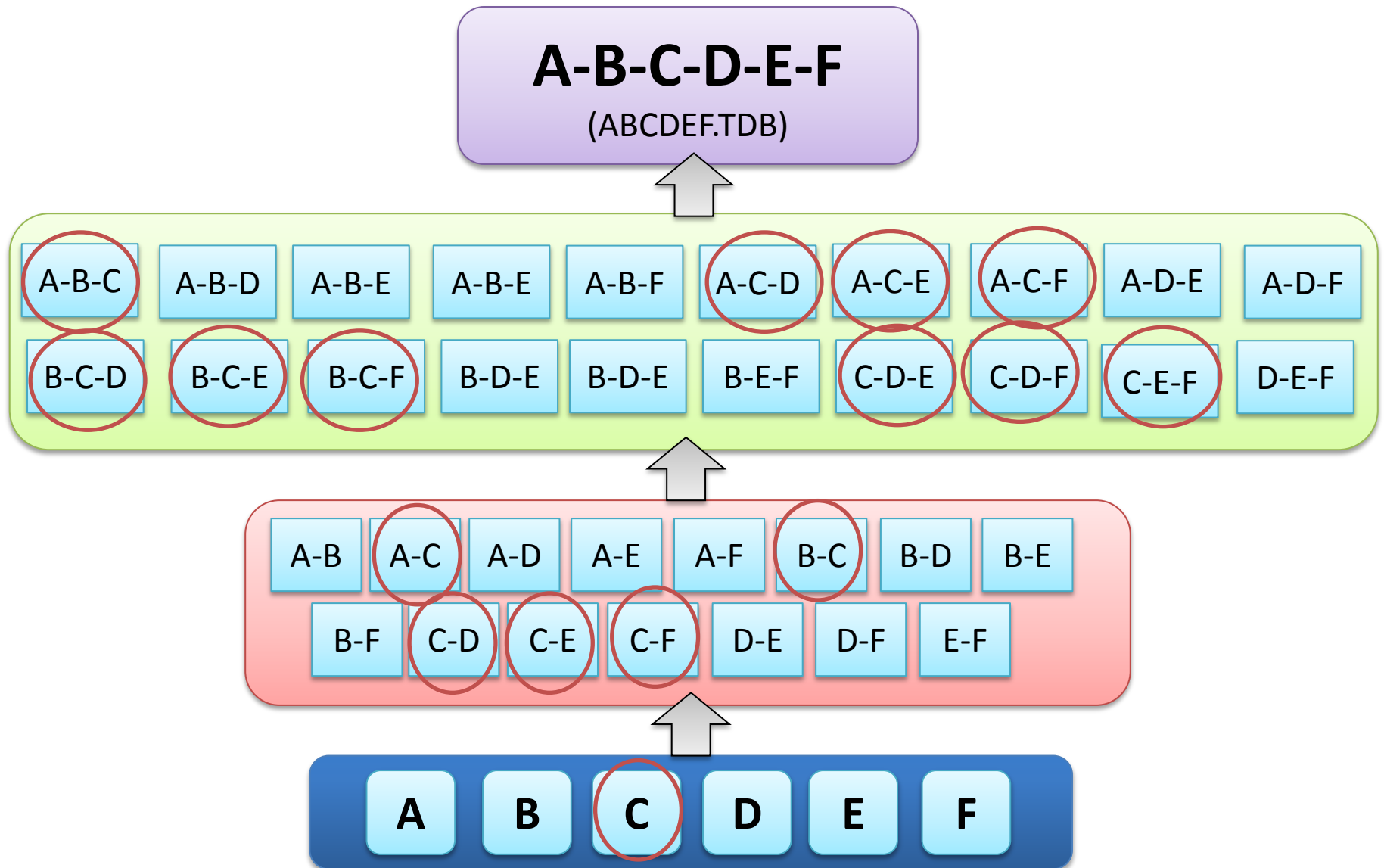
Intrinsic diffusivity

$${}^i D_{jk} = N_j M_j \frac{\partial \mu_j}{\partial N_k}$$

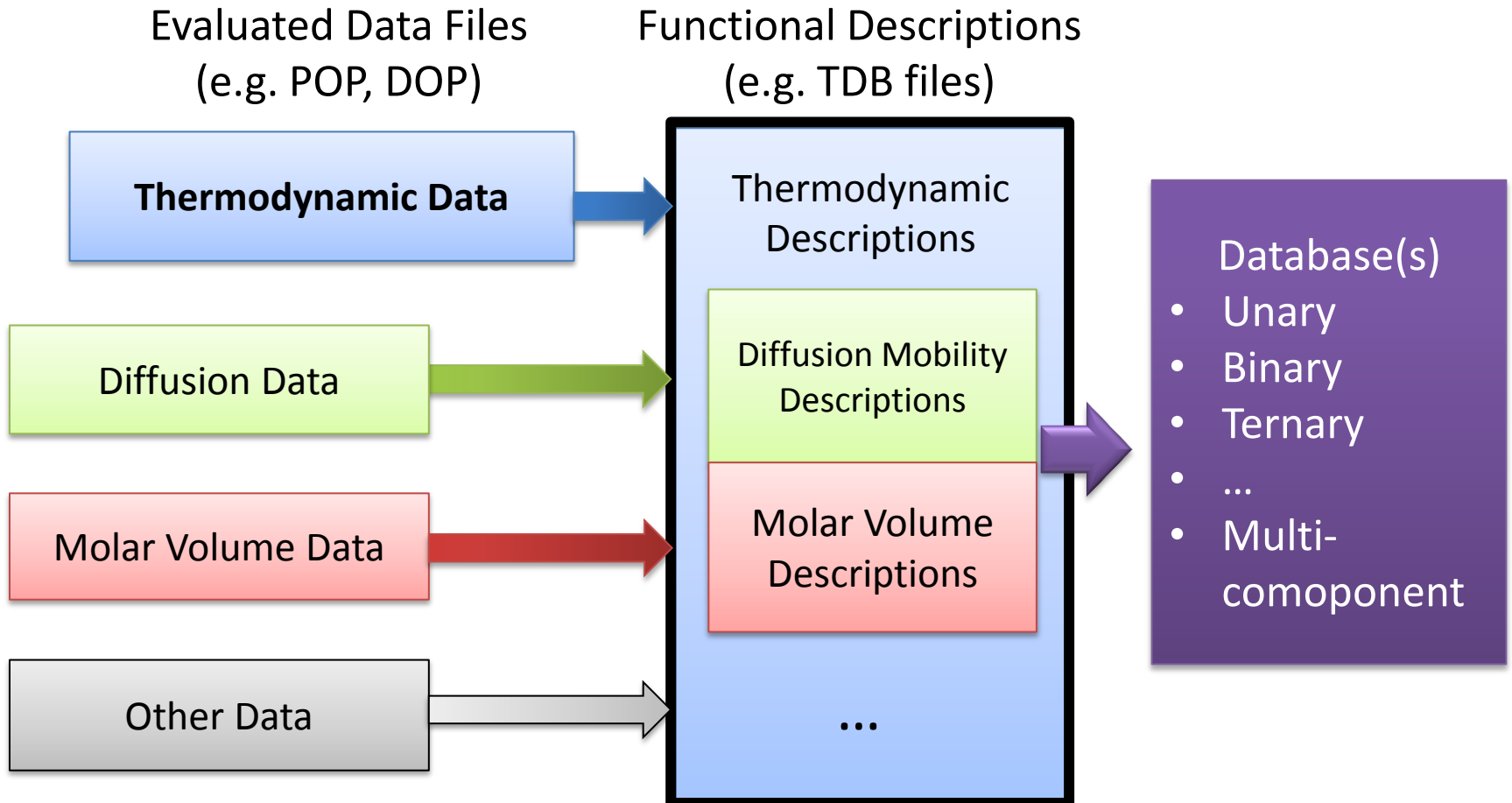
# Data Generation & Storage Tools



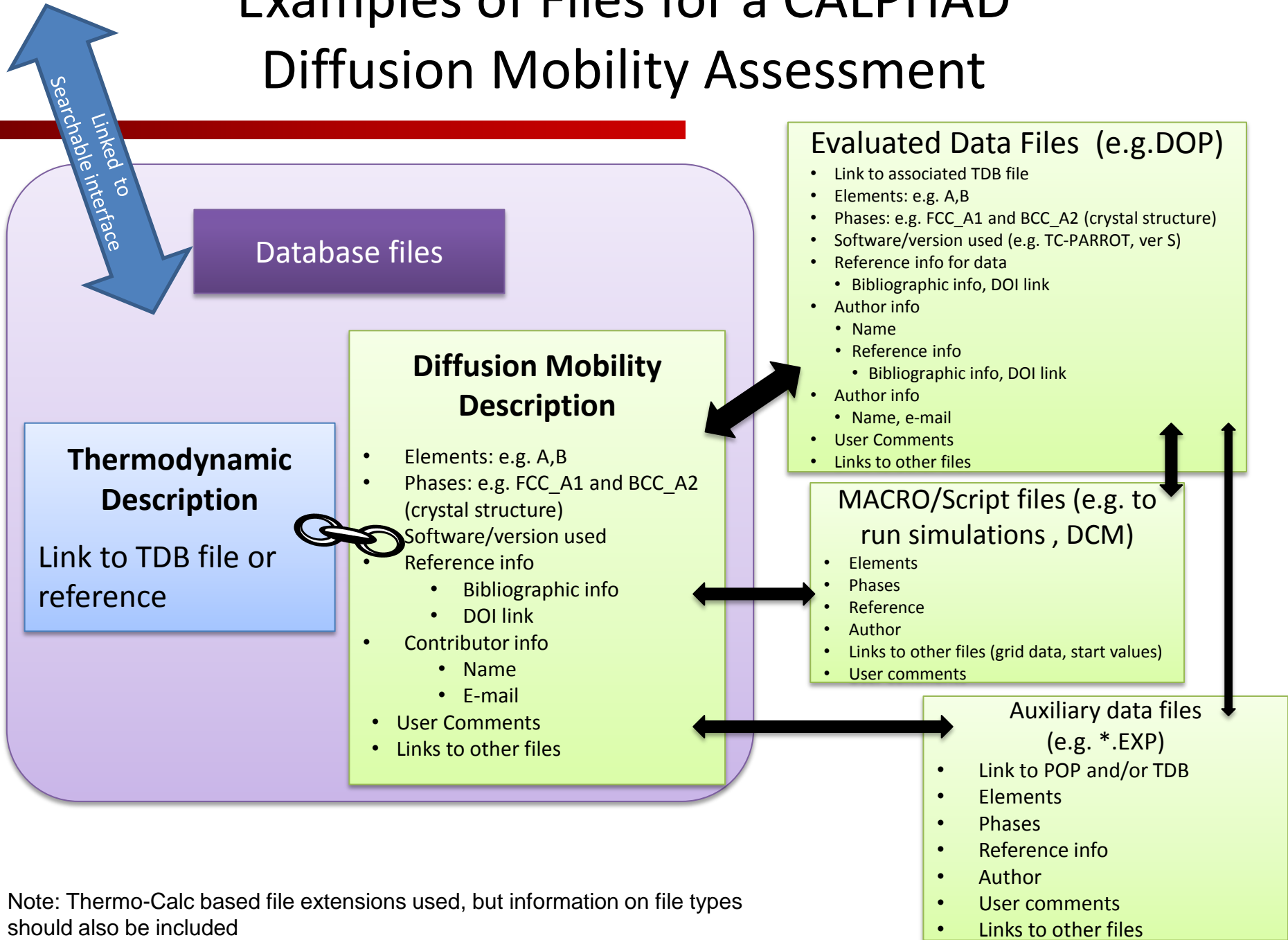
# Need for File Repository



# Structure of CALPHAD Database Files

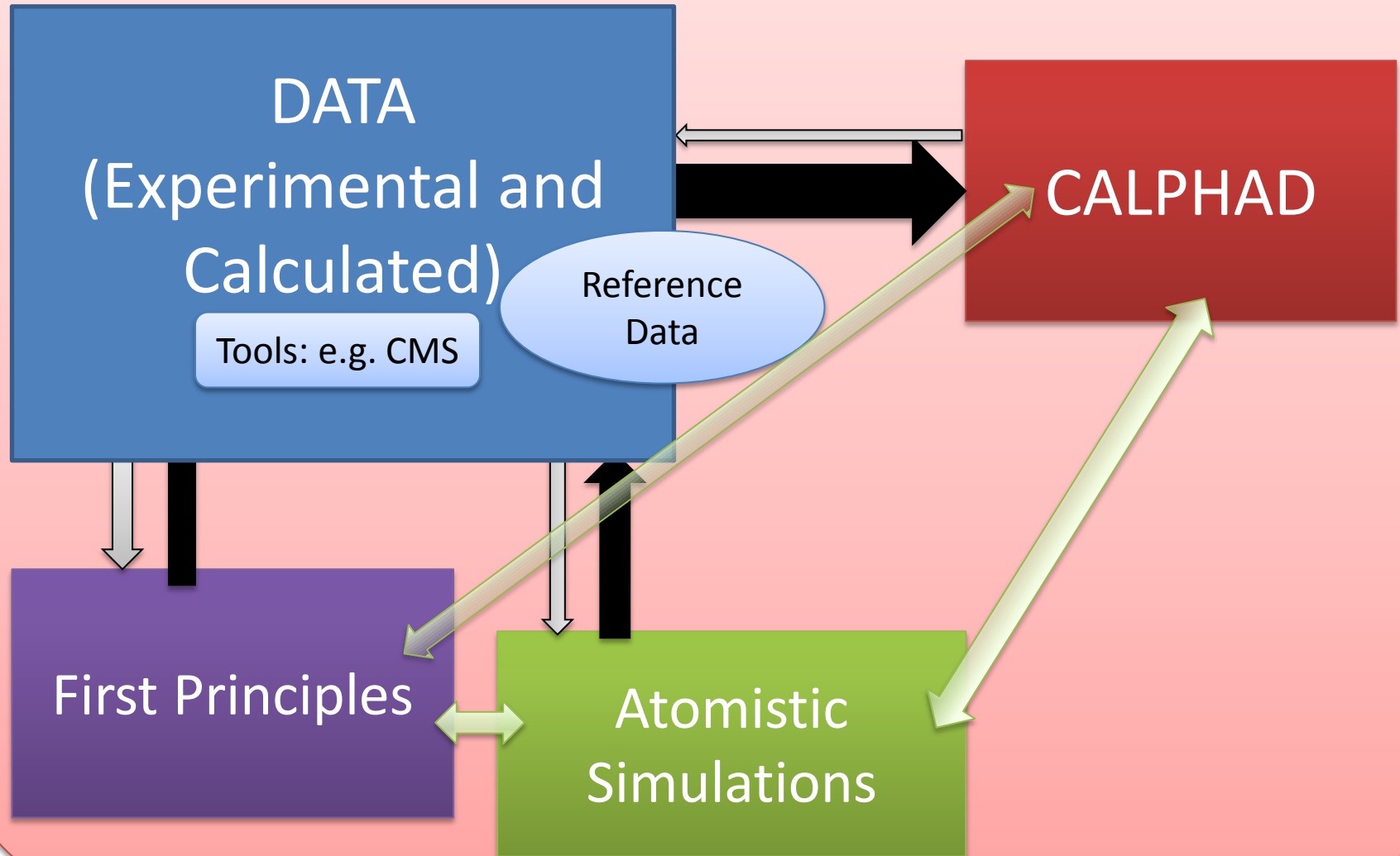


# Examples of Files for a CALPHAD Diffusion Mobility Assessment



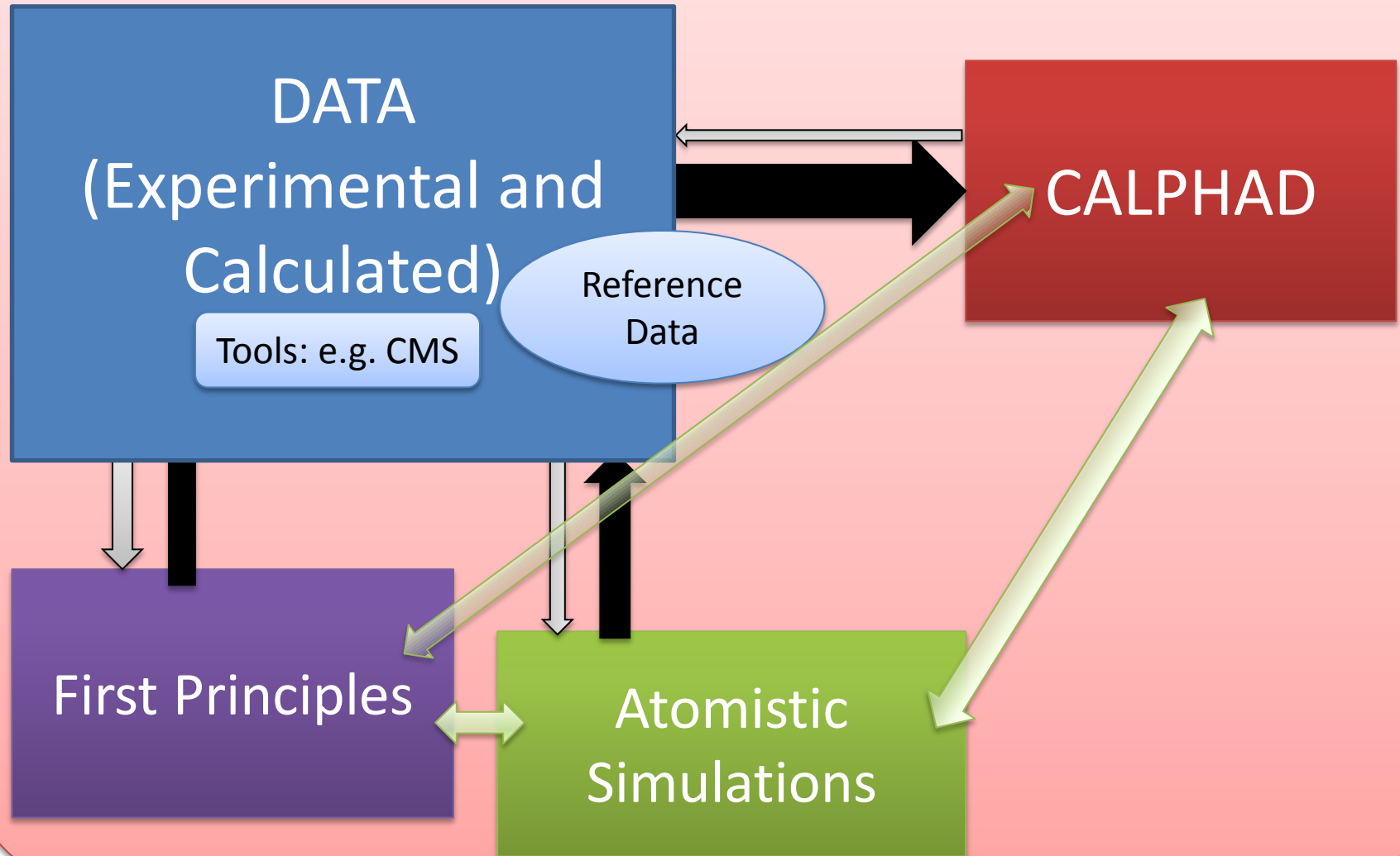
Note: Thermo-Calc based file extensions used, but information on file types should also be included

# Data Generation & Storage Tools

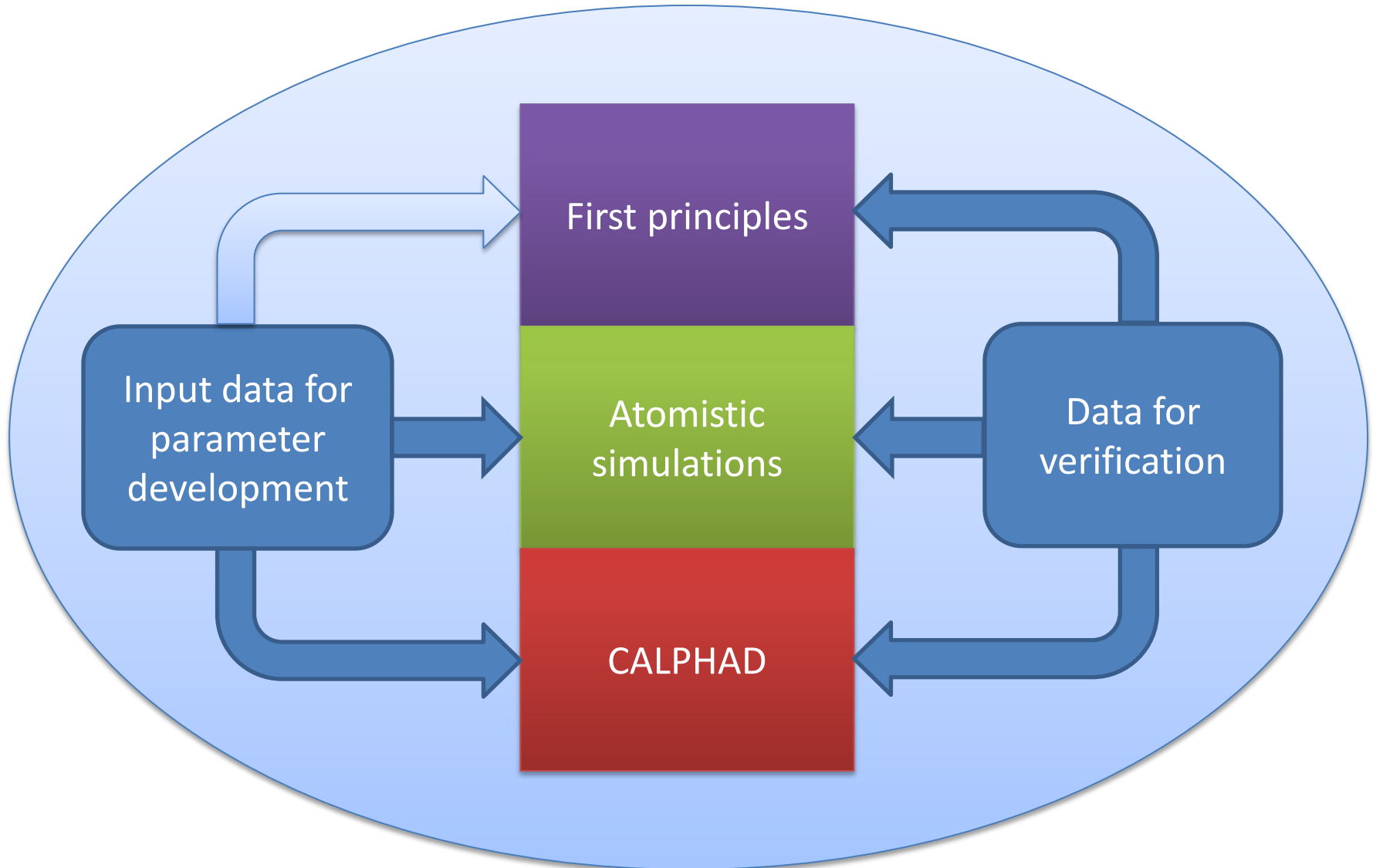




# Data Generation & Storage Tools



# Computational Methods and Data



# Information Need to Describe General Data Entry

## Data

- Elements present
- Type of value (e.g. enthalpy, heat of formation, phase boundary, diffusivity, lattice parameter, bulk moduli)
  - Experimental or computational method
  - Type of measurement (direct or indirect)
- Number of phases present
- Datum value and error
  - Type (single value or series)
  - Units
  - Actual value(s) and error(s)
- For each phase present
  - Phase name
  - Composition and formula
  - Crystal structure
  - Lattice parameter
- Temperature and error
- Pressure and error

## Metadata

- Type of material
  - Bulk composition
  - Material purity
  - Sample preparation
  - Microstructure
    - Single crystal
    - Polycrystalline (grain size, dislocation density)
    - Non-crystalline
- Data manipulation details (if any, e.g. reference state corrections, analysis method to determine interdiffusion coefficient)
- Reporting format (raw data, digitized data, other)
- Reference (DOI or text ; one must be present)
- Additional information

**Need flexible formats  
that can evolve with  
changing data needs!**

# Identifiers

- **Crystal Structure**

- Materials consist of distinct phases which are characterized by their crystal structures.
- Crystallographic data (*lattice parameter(s), space group symmetry, positional parameters*) for a phase convey the identities of constituent atoms as well as their exact positions in space.
- Information on the identities and arrangement of constituent atoms (crystallographic data) provide the initial critical input for modeling materials behavior.
- Use the InChI identifier (Space group & wycoft sites)

- **Material** (Composition, Heat Treatment, + ???)

- InChI type identifier

# Diffusion Data Types

- Tracer Diffusivity
- Intrinsic Diffusivity
- Interdiffusion
- Grain boundary diffusion
- Activation Energies
- Diffusion Couple Composition Profiles
- Layer widths

Could be reported as a function or individual determined diffusivities.

Need to know temperature, composition, phases, present, and diffusing species

## Metadata Need:

- Type of material
  - Bulk composition
  - Material purity
  - Sample preparation
  - Microstructure information
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# Things to Consider

- Naming conventions for files
  - ABC\_Smith05.xxx
- Files to reproduce a specific DICTRA simulation store together in a zip or separate linked files in workspace

# Concepts for Workspace/Workflow

- An individuals can have multiple different workspaces.
  - AlCu-Thermodynamics; CoNi-Diffusion, ReNi-Experimental.  
AlTi-FirstPrinciples
- Owner of the workspace defines who has access to the workspace and what the level of access is.
- Workspaces can be shared between different groups/worskspaces
- Workspaces would have some basic structure elements specific to the type of workspace.
  - Thermodynamics, Diffusion, Experiments (types of workspace)
  - Common structural elements: People, Types of Resources, Systems

# Discussion Points – What's workable?

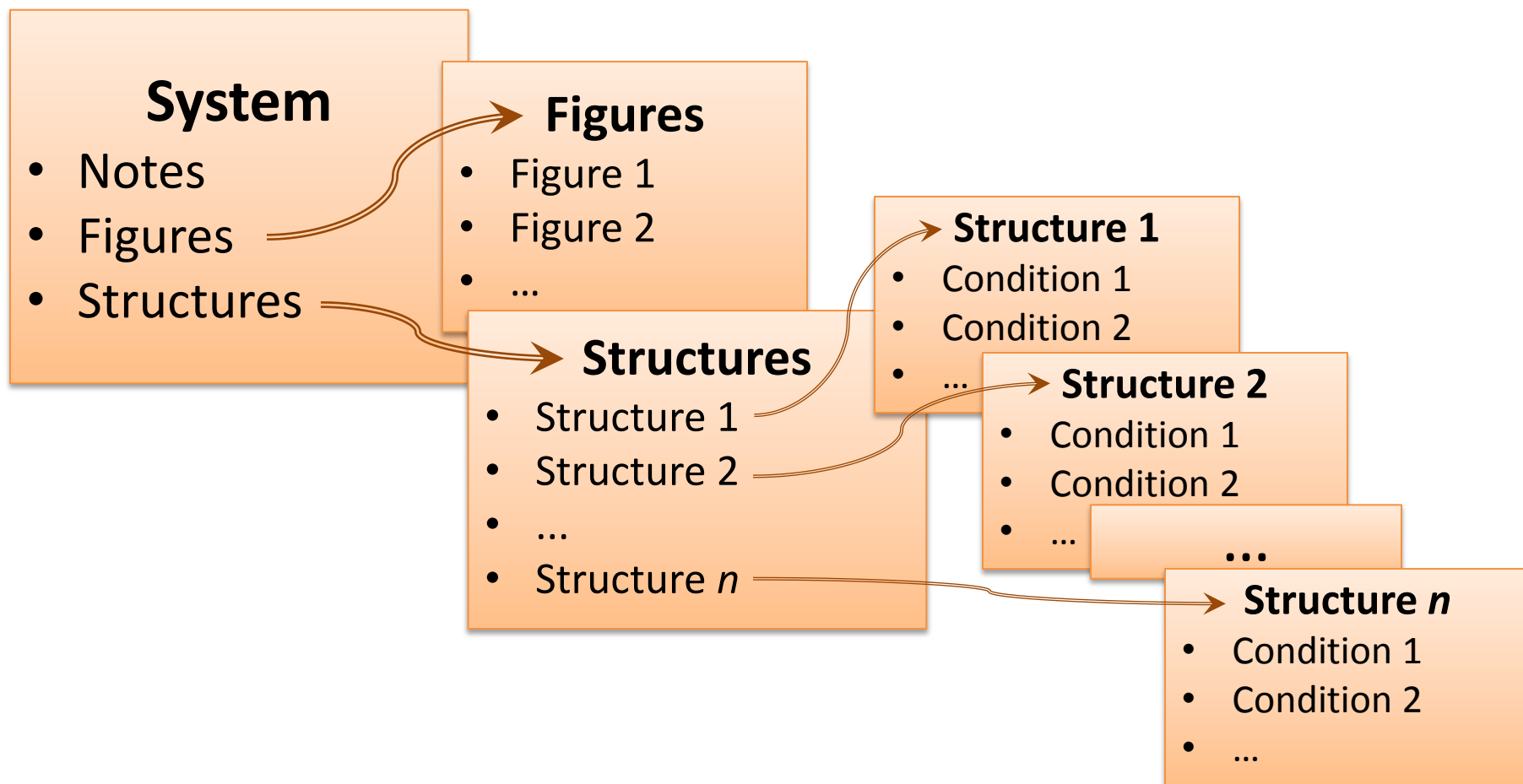
- - Naming conventions for files & workspaces
- - Files to reproduce a specific DICTRA
- - Some files stored in zip; others as separate linked files
- - Start with 8 categories:
  - People; Phased Based Materials Properties; Phases; Software; Systems; Techniques; Treatments; Types of Resources



# File Repository for ATAT-MAPS Data Structure

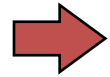
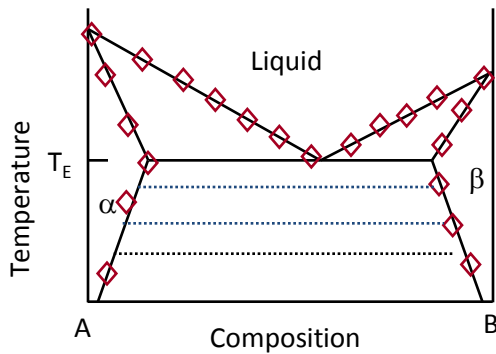
## Chemical Systems:

Au-Cu, NaCl-KCL, SiC-AlN, AlN-GaN-InN, ...

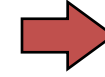
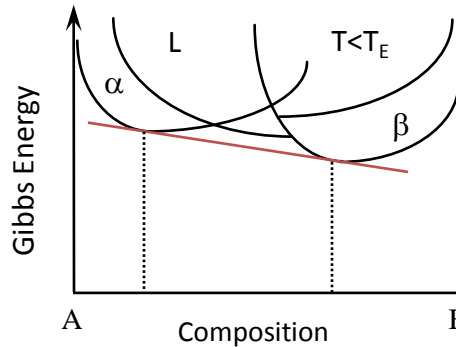


# Original CALPHAD Approach

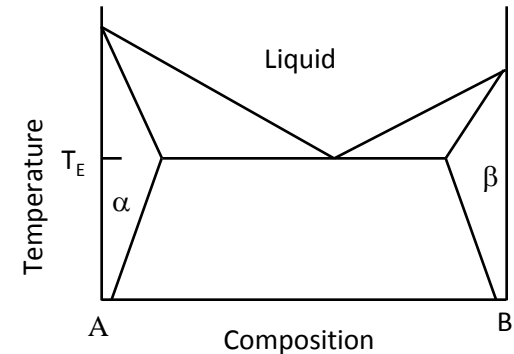
Experimental phase diagram and thermochemical data



Determine Gibbs energy functions for each phase:  $G = f(x, T, P)$

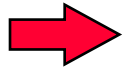


Calculated phase diagram



$$G^\phi = G^0 + G^{ideal} + G^{excess}$$

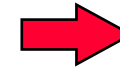
Binaries



Ternaries



Quaternaries



$n^{\text{th}}$  Order Systems

True quaternary compounds are rare in metallic systems

⇒ Assessment of ternary systems is usually sufficient for the description of a multicomponent system

⇒ Same methodology can be applied to the description of other property data