

PHASE BEHAVIOR AND MORPHOLOGY OF MICROEMULSIONS IN A POLYMER- SURFACTANT SYSTEM

Alexa B Cano

NIST Center for Neutron Research,
Neutron Condensed Matter Science Group
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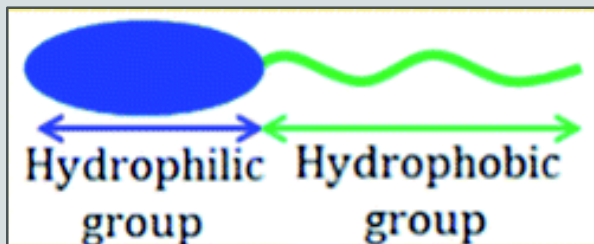


OVERVIEW

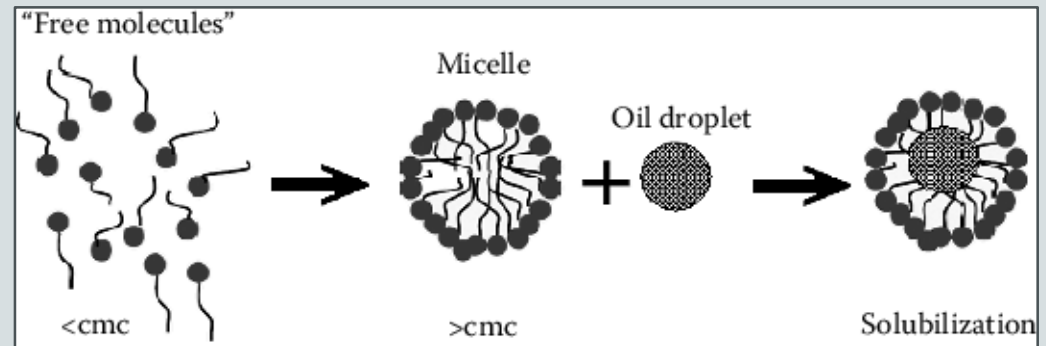
- Microemulsions and phase morphologies
- The Pluronic system
- Using SANS to determine structural phases
- Rheological properties related to structural phases

MICROEMULSIONS

- Ternary system of oil, water, and surfactant
- Self assembly
- Size domain: ~1-100 nm
- Prominent roles in pharmaceutical and industrial applications
- Understanding nanostructures is critical for understanding bulk material properties



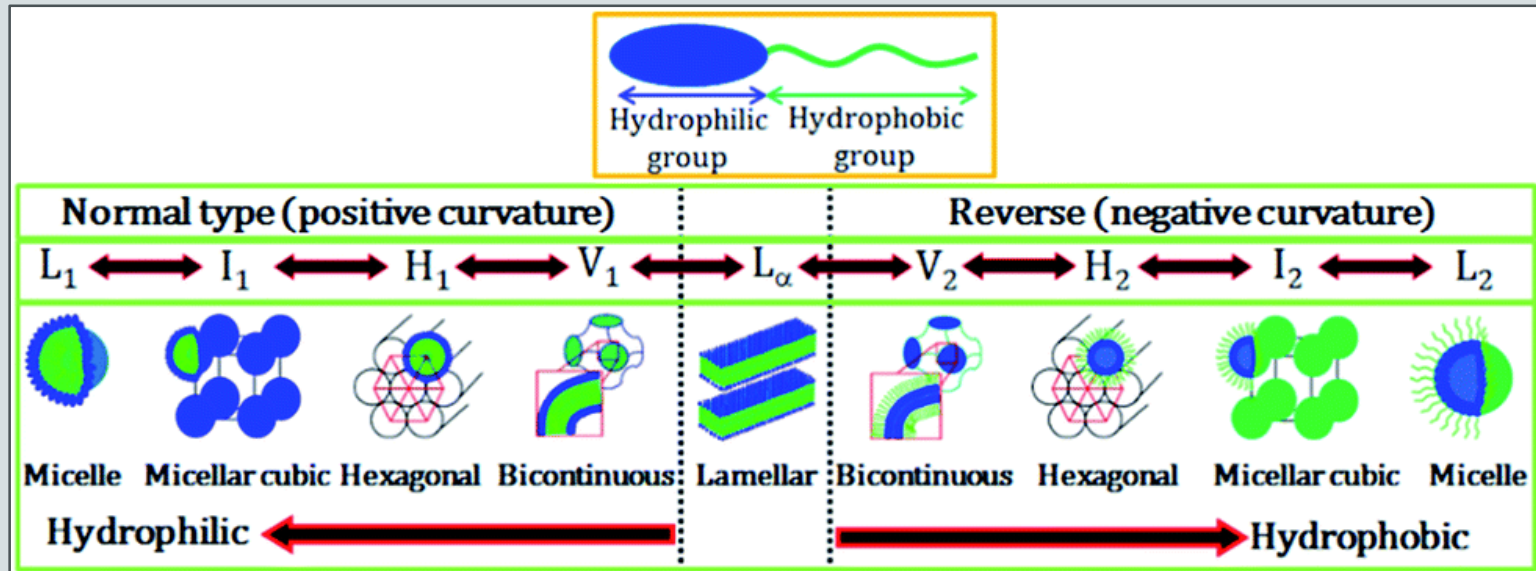
<https://www.semanticscholar.org/paper/Lyotropic-liquid-crystal-systems-in-drug-delivery%3A-Kim-Jahn/020c104ac5d99038e0493d684489fa737ad2bede>



[https://sites.ualberta.ca/~csps/JPPS8\(2\)/C.Rangel-Yagui/solubilization.htm](https://sites.ualberta.ca/~csps/JPPS8(2)/C.Rangel-Yagui/solubilization.htm)

LYOTROPIC LIQUID CRYSTALLINE MESOPHASES

- Oil-in-water or water-in-oil (reverse microemulsion)
- Intermediate phases depending on surface curvature
- Preferred surface curvature is given by volume ratio between hydrophobic and hydrophilic domains



PLURONICS AS POLYMERIC SURFACTANTS

- System of interest: Pluronic P84 – p-xylene – water
- 60 wt% and 70 wt% pluronic microemulsions
- Exhibits many phases
- Phase boundaries may be identified by turbidity in response to temperature

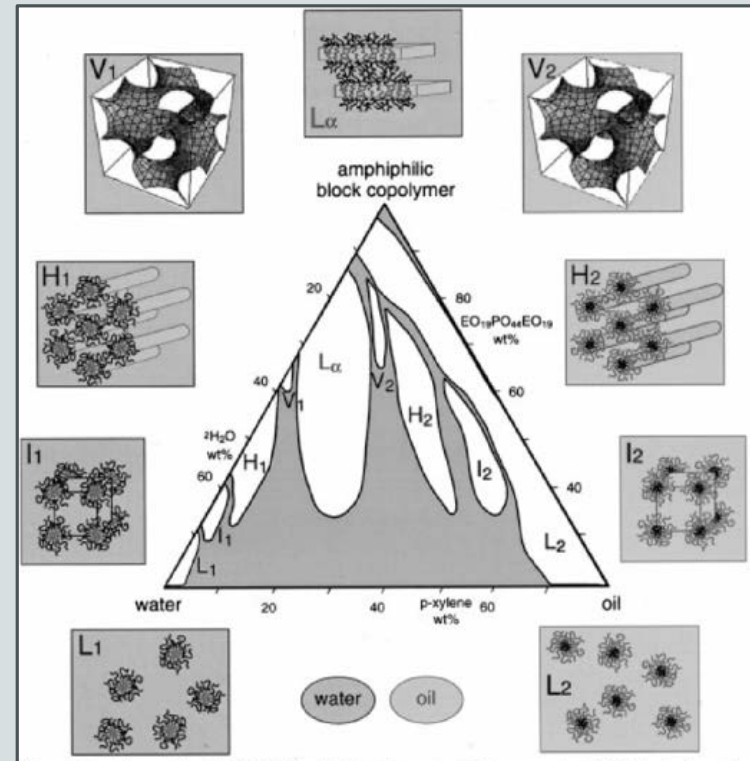
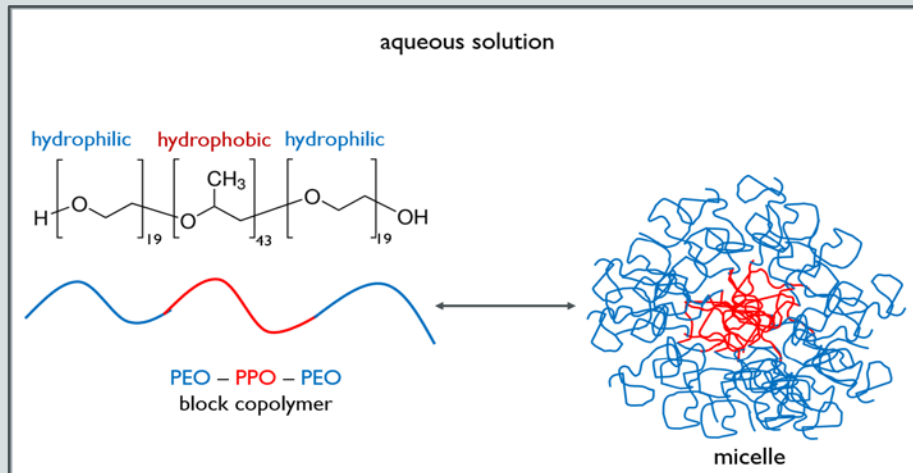
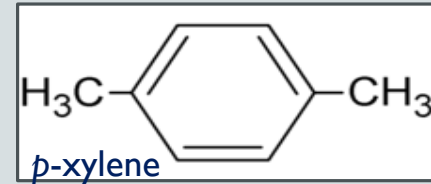


Figure 1. Phase diagram of the (EO)₁₉(PO)₄₄(EO)₁₉-H₂O("water")-p-xylene("oil") ternary system at 25 °C. The phase boundaries

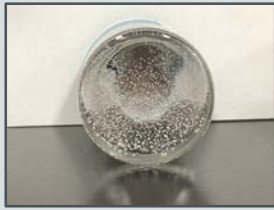
DETERMINATION OF PHASE BOUNDARIES

60 wt% P84 at 35°C

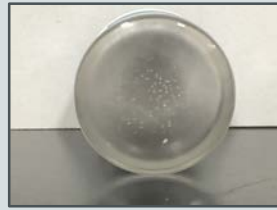
surfactant-oil-water

Increasing oil to water ratio →

60-0-40



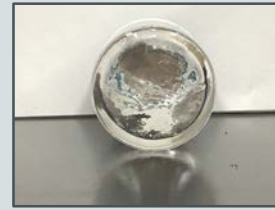
60-2-38



60-5-35



60-14-26



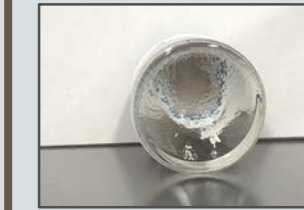
60-19-21



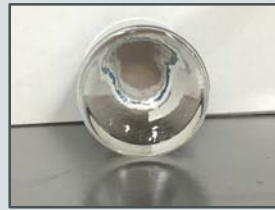
60-23-17



60-27-13



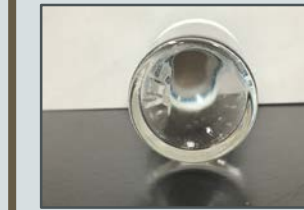
60-29-11



60-31-9



60-33-7

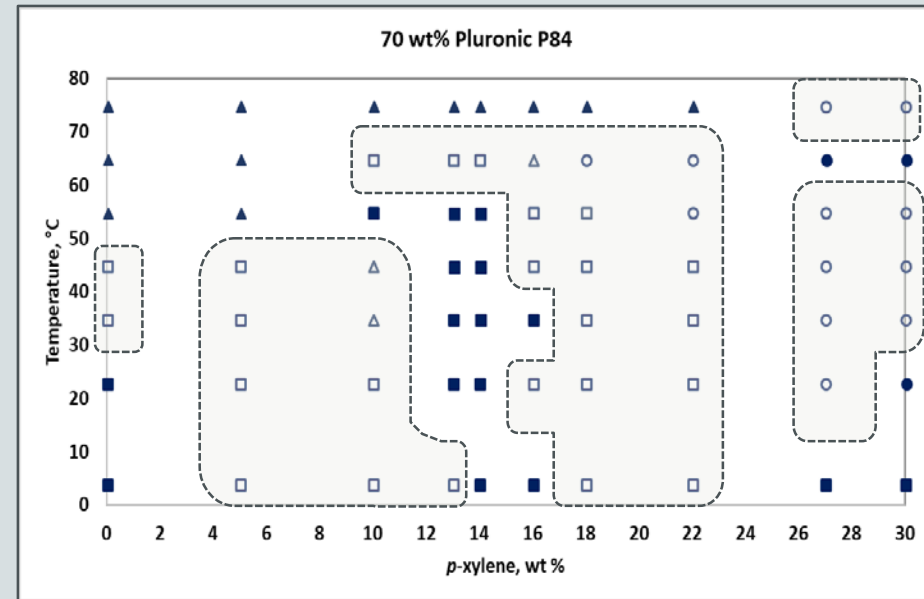
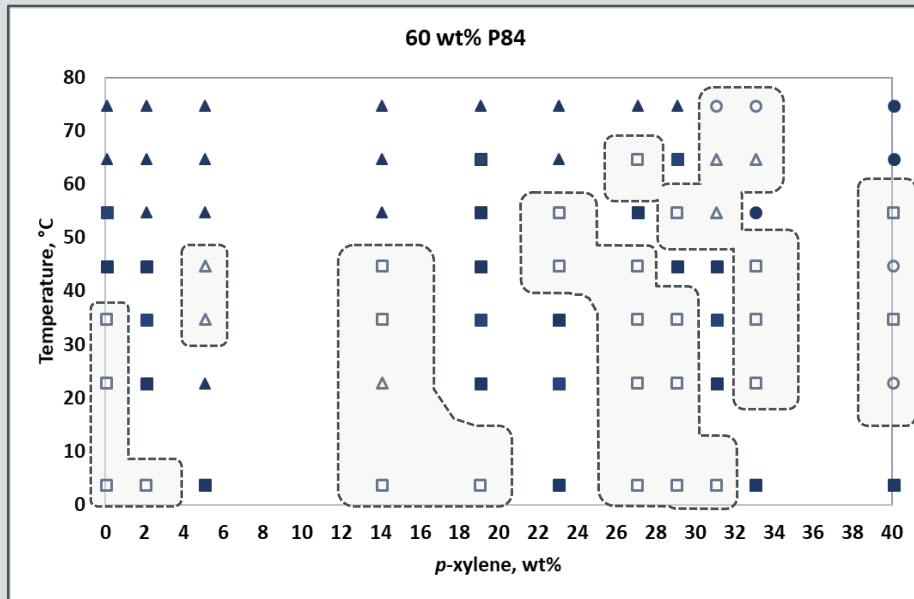


60-40-0



Increasing oil to water ratio →

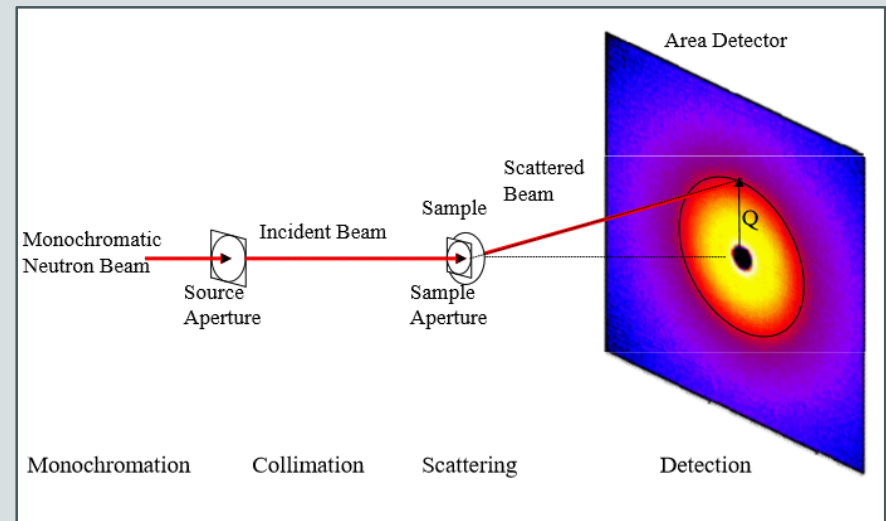
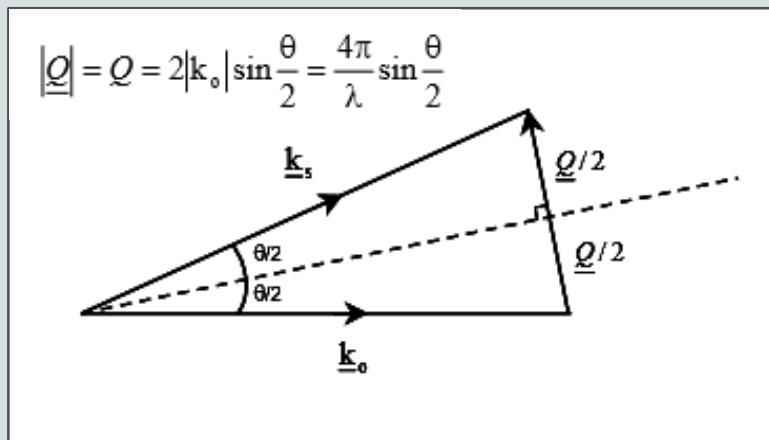
PHASE BOUNDARY TEMPERATURE DEPENDENCE



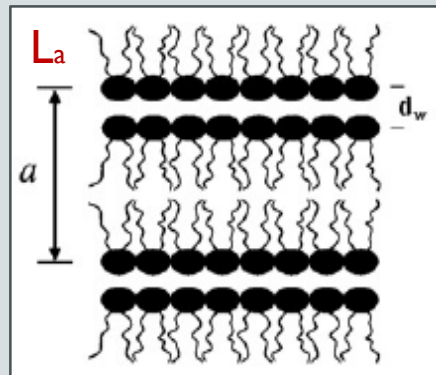
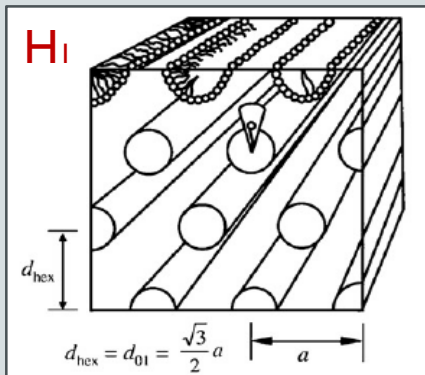
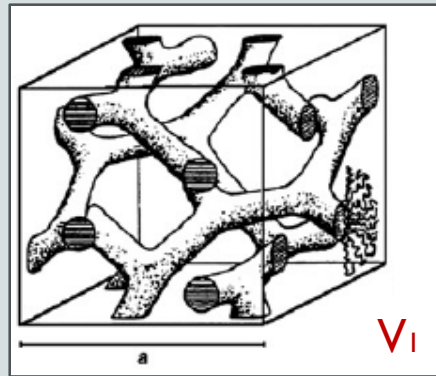
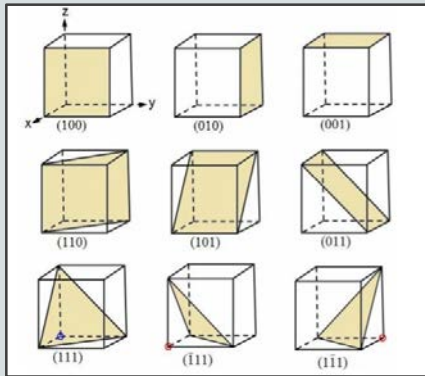
SMALL ANGLE NEUTRON SCATTERING (SANS)

- Beam of neutrons scatters off sample at an angle and onto detector
- Scattered distance (Q) is measured in reciprocal space
- Q reveals structures and dimensions within sample

For elastic scattering,

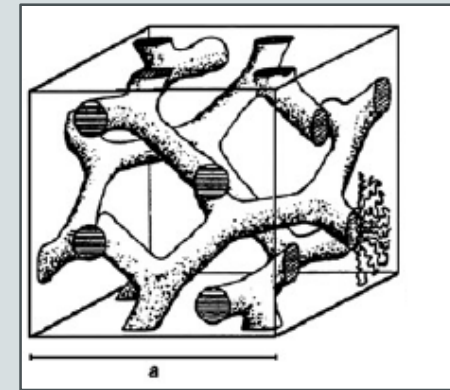
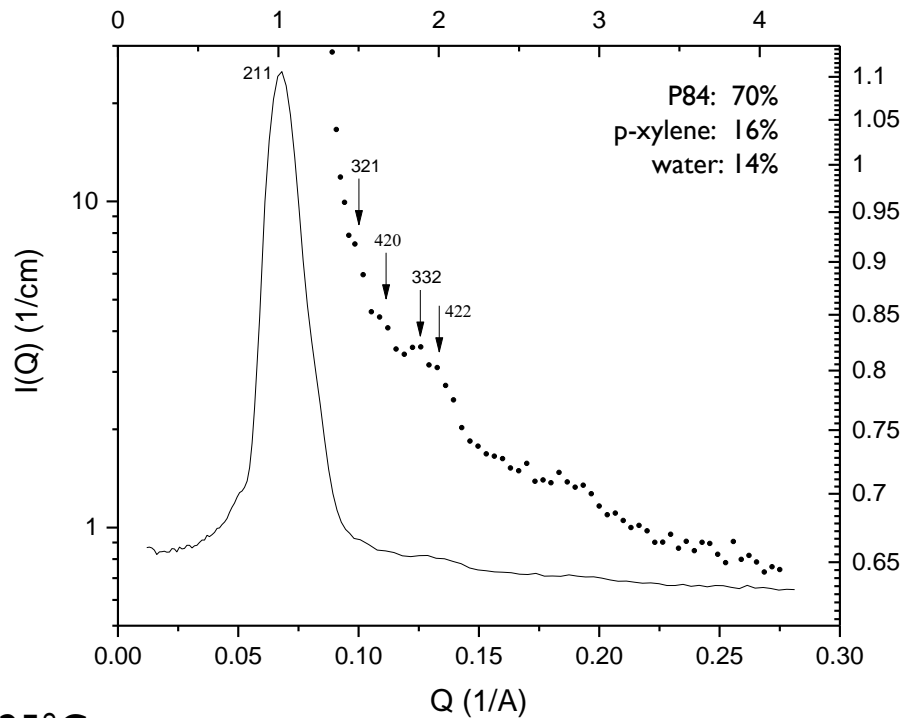


CRYSTALLOGRAPHY



- Mesophases can be liquid crystalline in structure
- Periodicity of liquid crystal results in Bragg reflections from different sets of crystallographic planes
- Reflections probed by SANS serve as structure IDs for mesophases

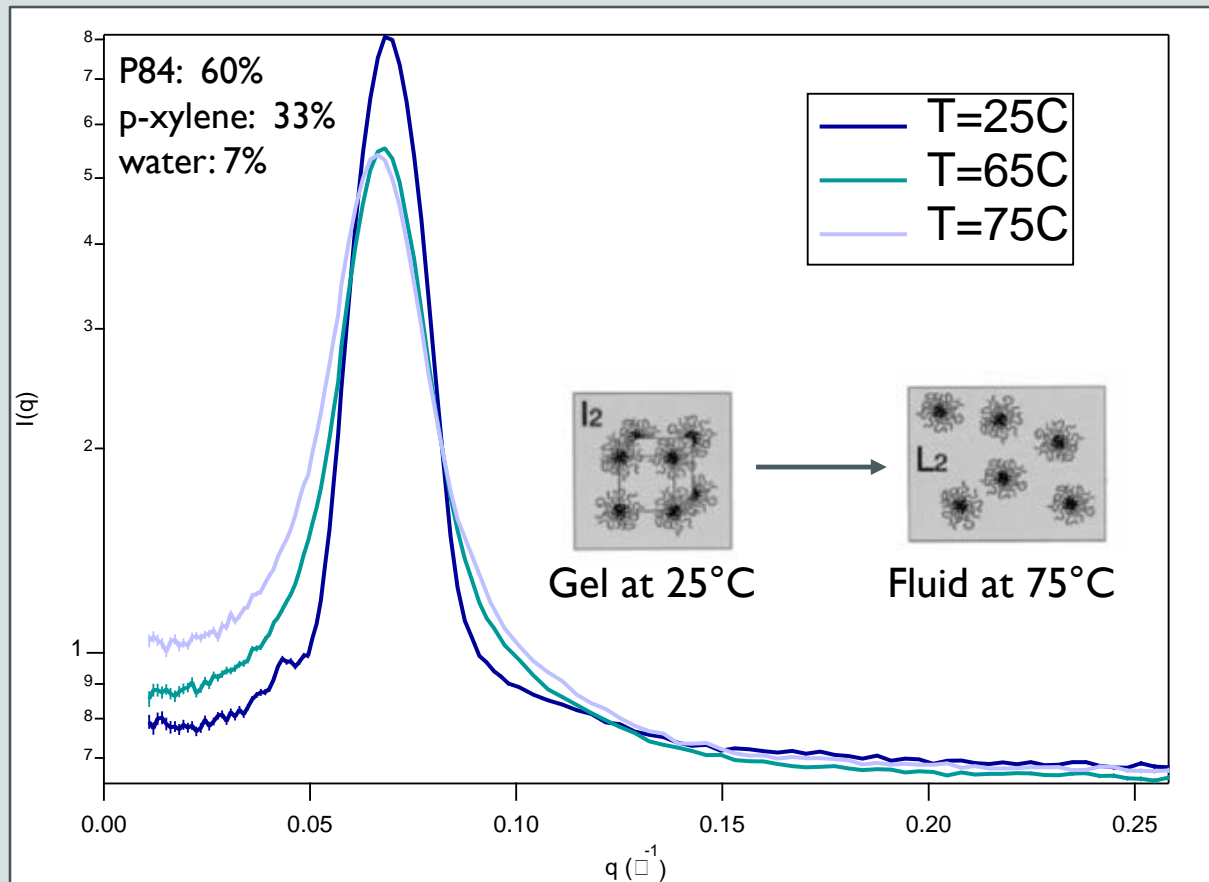
REVERSE BICONTINUOUS CUBIC MESOPHASE



- The relative peak positions identify the phase
- The q values of peaks are a function of the plane indices (hkl) and the unit cell dimension (a)

$$\{hkl\} \rightarrow q = \frac{2\pi\sqrt{h^2 + k^2 + l^2}}{a}$$

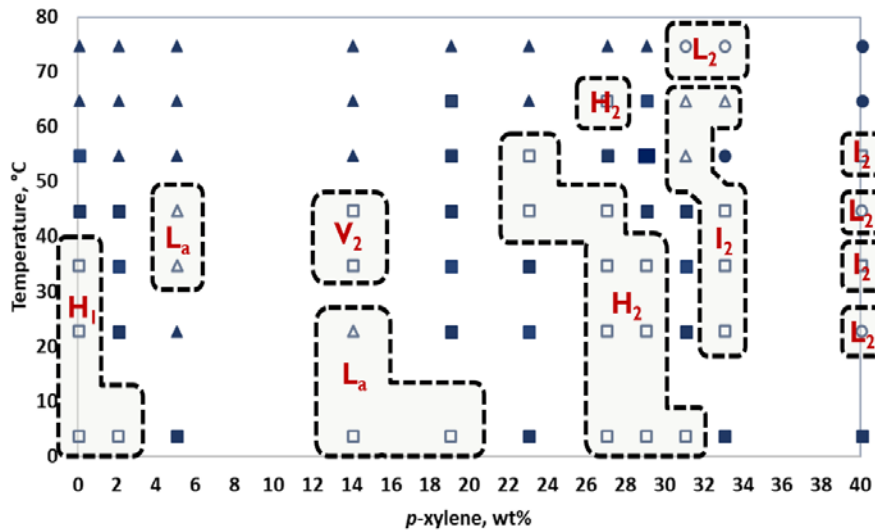
IDENTIFYING PHASE TRANSITIONS USING SANS



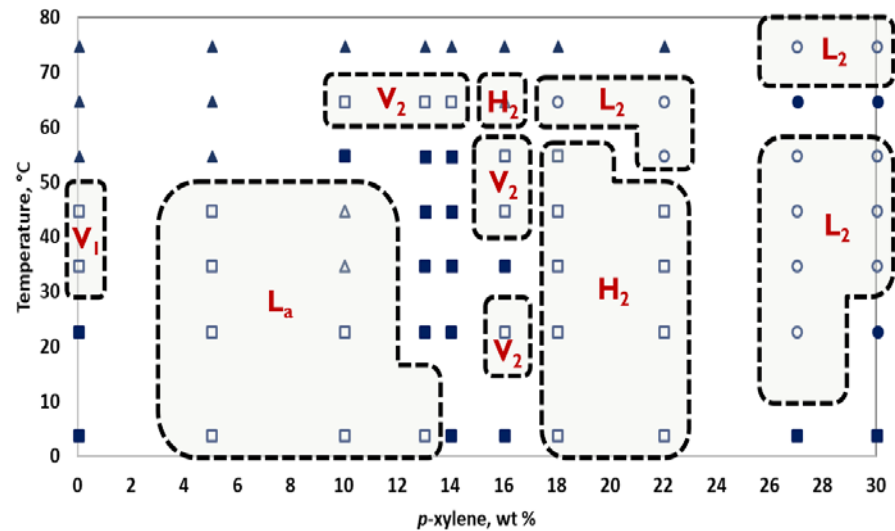
- Reverse micellar cubic at 25°C
- Phase transition around 65°C
- Reverse micellar solution at 75°C

PHASE BOUNDARY TEMPERATURE DEPENDENCE

60 wt% P84

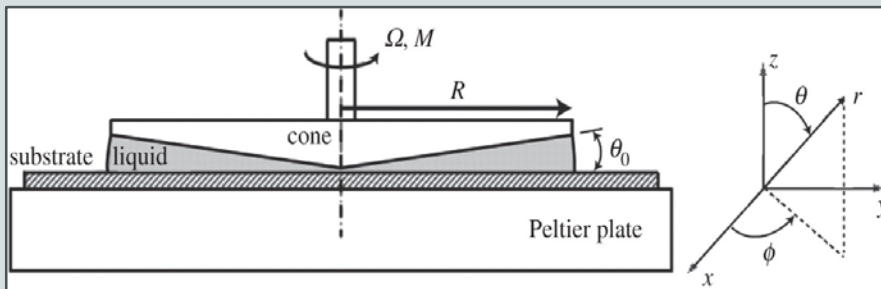
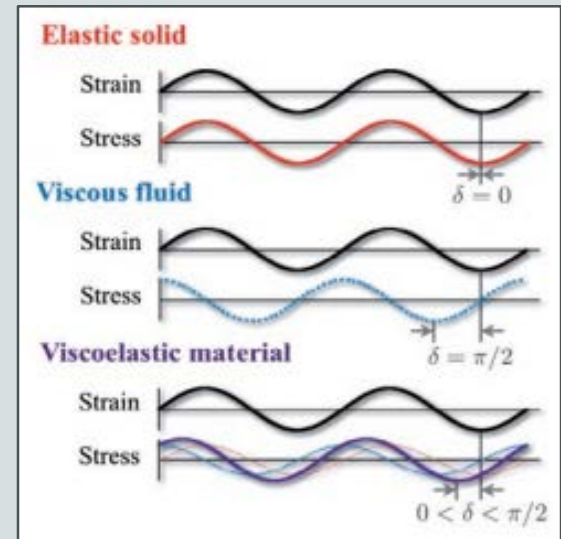


70 wt% Pluronic P84



RHEOLOGY

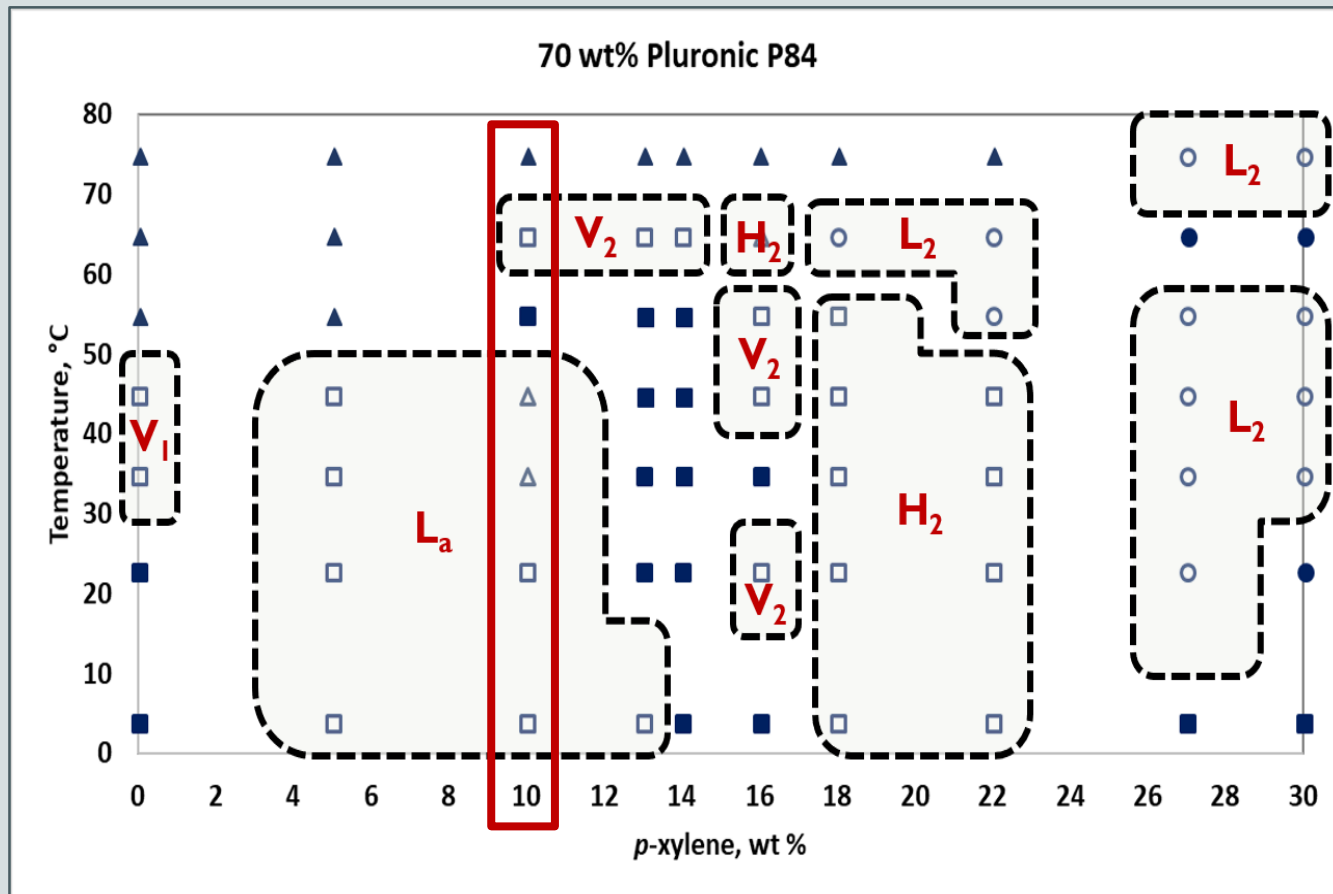
- Small-Amplitude Oscillatory Shear
- Elastic modulus G'
- Viscous modulus G''
- Moduli expected to change upon phase transition



$$G' = (\sigma_0 / \gamma_0) \cos \delta$$

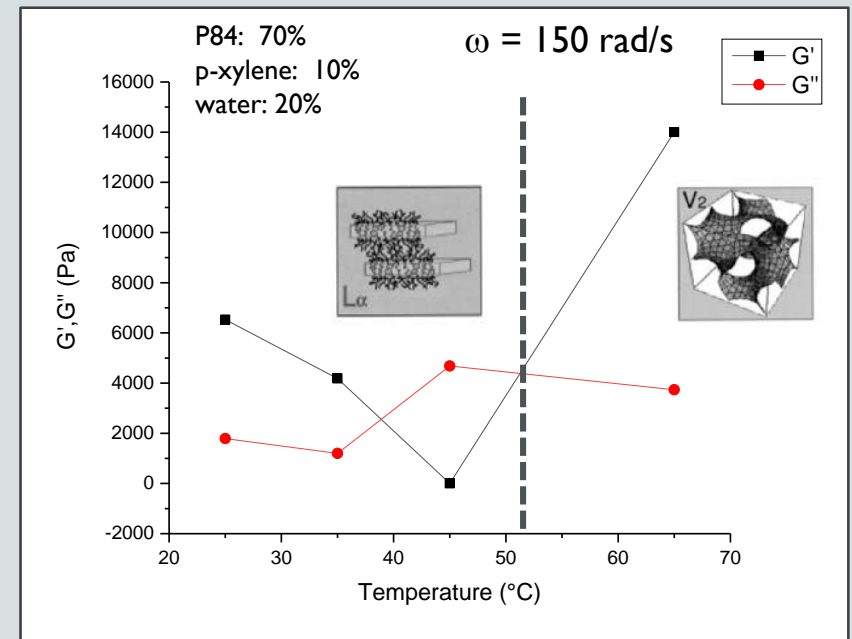
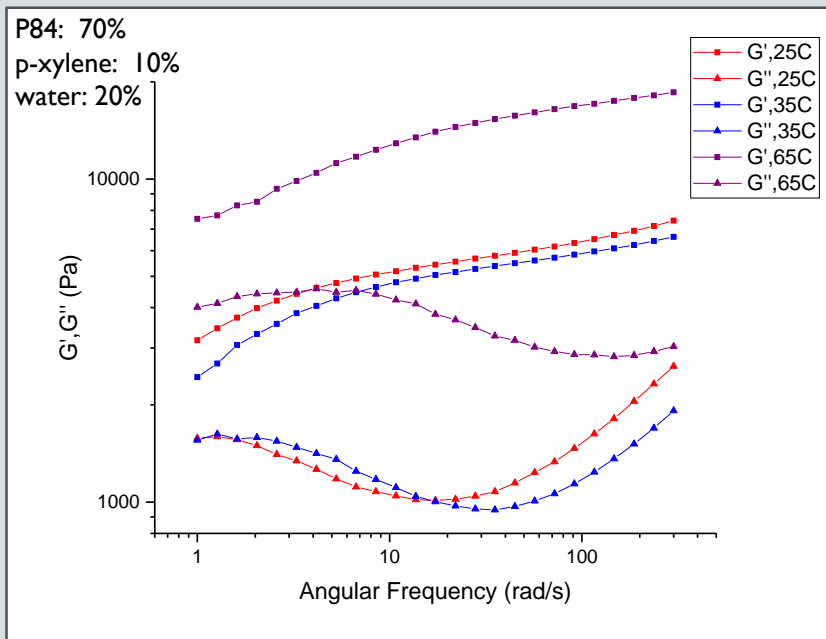
$$G'' = (\sigma_0 / \gamma_0) \sin \delta$$

PHASE BOUNDARY TEMPERATURE DEPENDENCE



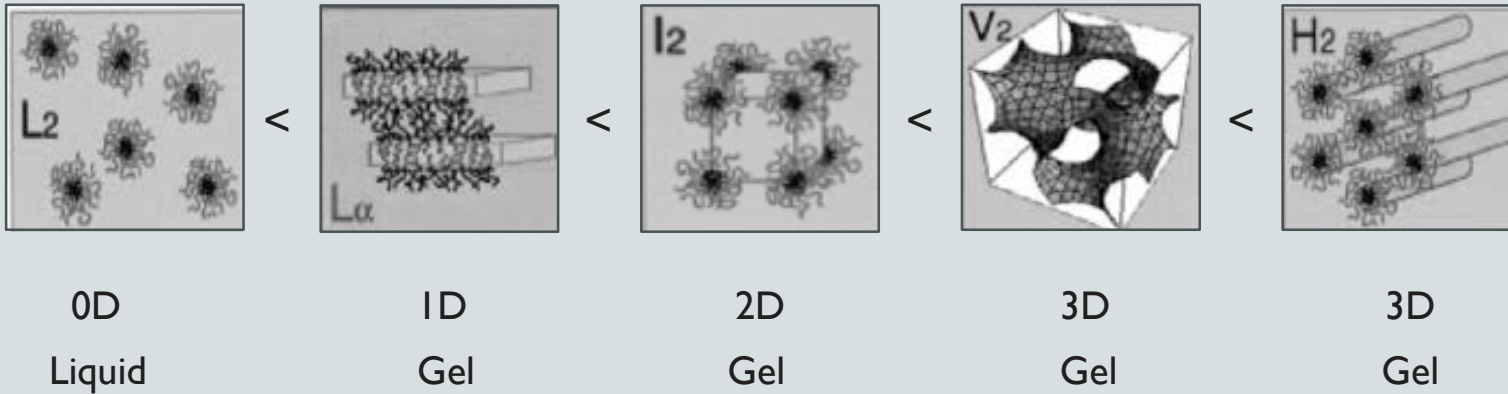
RHEOLOGY RESULTS

- Example of transition: 70 % P84, 10 % oil, 20 % water
 - Transition at ca. 50 °C
 - From lamellar gel to bicontinuous gel
 - Bicontinuous gel has higher connectivity and is therefore stronger



RHEOLOGY RESULTS

- The storage modulus generally increases with connectivity:



- Bulk rheological properties can be explained by mesophase nanostructures

CONCLUSION

- Temperature dependent phase behavior determined from SANS
- Connectivity of mesophase directly relates to rheological properties
- Future study: Relate connectivity and rheology to electrical conductivity

ACKNOWLEDGEMENTS

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- Joe Dura, Julie Borchers and the staff at the NIST Center for Neutron Research
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