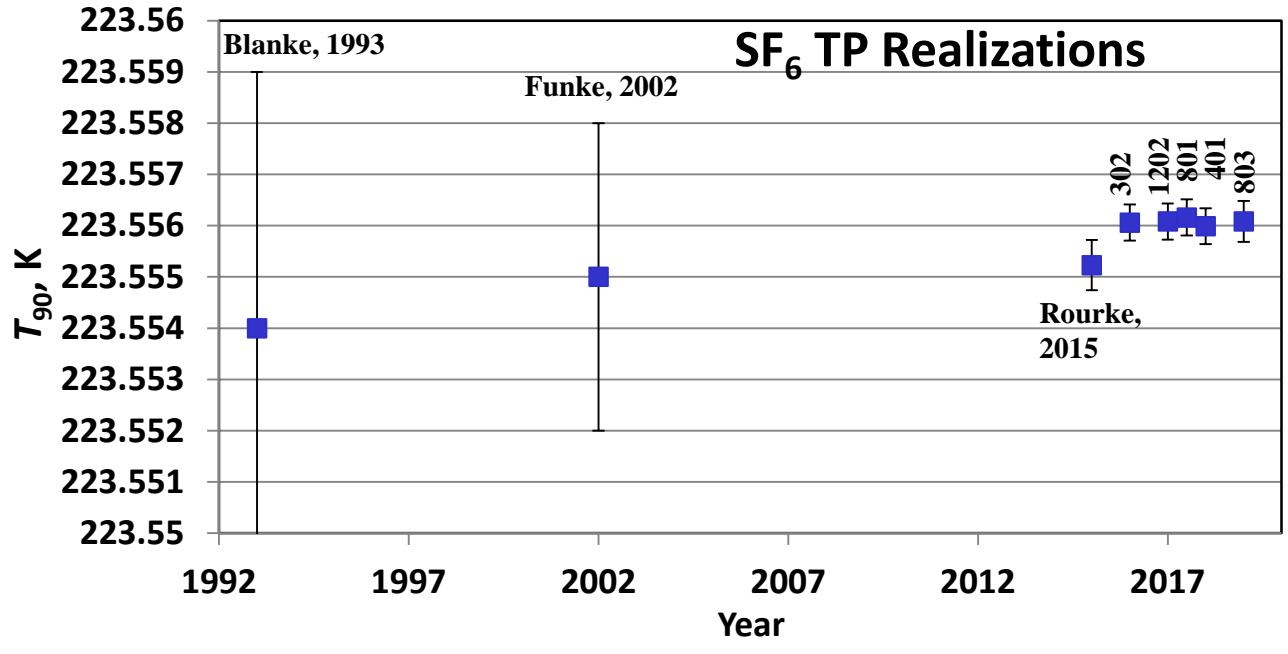
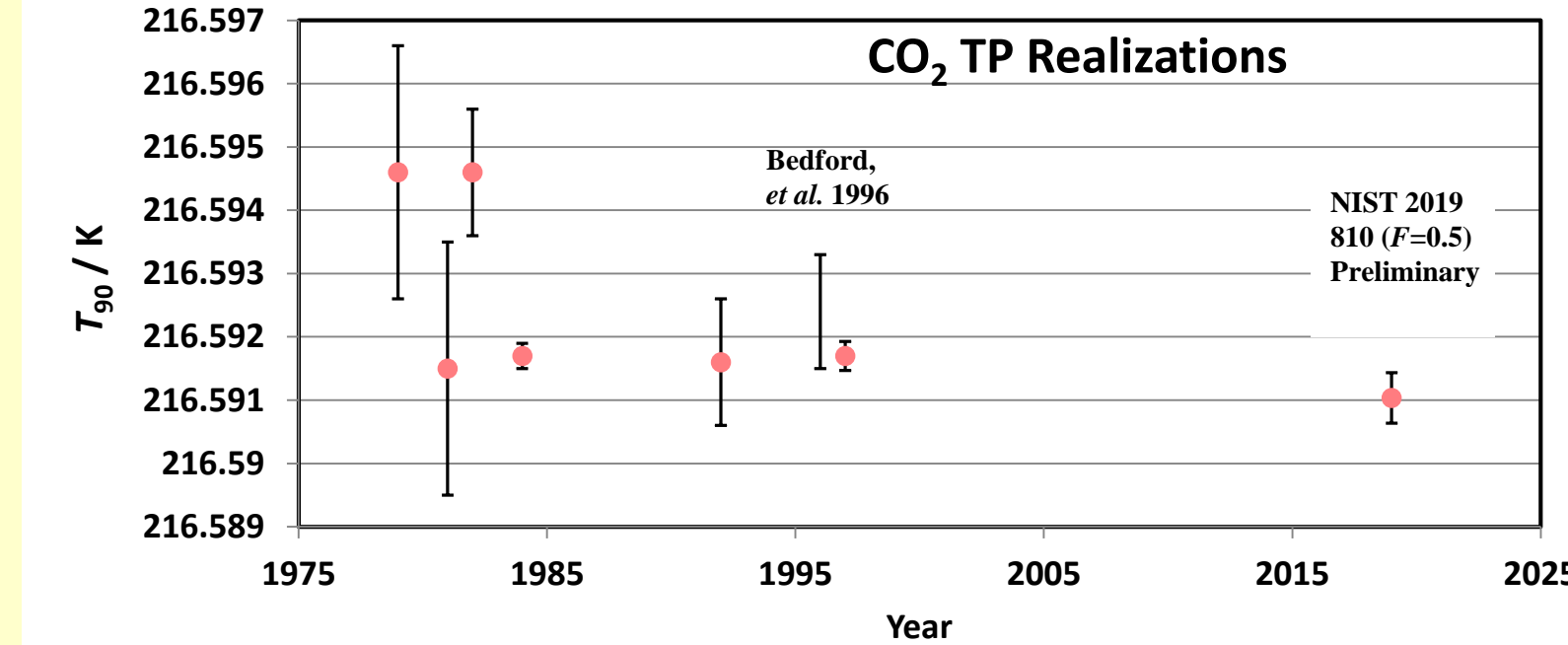


Abstract / Introduction

International restrictions on the trade of mercury-based products have created a near-term need to develop alternative materials for temperature standards. The triple points of SF₆ and CO₂ are good candidates for replacing the Hg triple point (TP) as a defined fixed point on the International Temperature Scale of 1990 (ITS-90). We have constructed a series of immersion cells for realizations of both the SF₆ TP and CO₂ TP that are suitable as 'drop-in-replacements' for Hg TP cells as currently used in conventional refrigerated baths. These immersion cells are pressurized, with no external manifold, and refillable via a conventional service valve. Realizations are performed in a quasi-adiabatic mode inside of a small vacuum chamber that is immersed in an ethanol bath. We report recent measurements using capsule-type standard platinum resistance thermometers (SPRTs) that have been calibrated on the subrange 4 of the ITS-90. Melting plateaus are presented versus both melted fraction F and inverse fraction F^{-1} scales. We compare the interpolated T_{90} temperatures as extrapolated to $F=1$ from these realizations with those reported in our prior work and from the published literature.



Recent History of SF₆ TP realizations.



History of CO₂ TP realizations since 1979.

Triple Point Cell Design and Construction

	T_{tp} (K)	p_{tp} (MPa)	ρ_l (kg/m ³)	ρ_s (kg/m ³)	Δh_f (J/cm ³)	dp_m/dT (MPa/K)	dT/dz (mK/m)	A^{-1} uK/ppm
Hg	234.3156	1.7×10^{-10}	13,690	14,184	156	18.5	8.6	199
SF ₆	223.556	0.231	1845	2281	66.7	1.56	11.6	79
CO ₂	216.59	0.55536	1178.5	1532	232	4.69	2.47	45

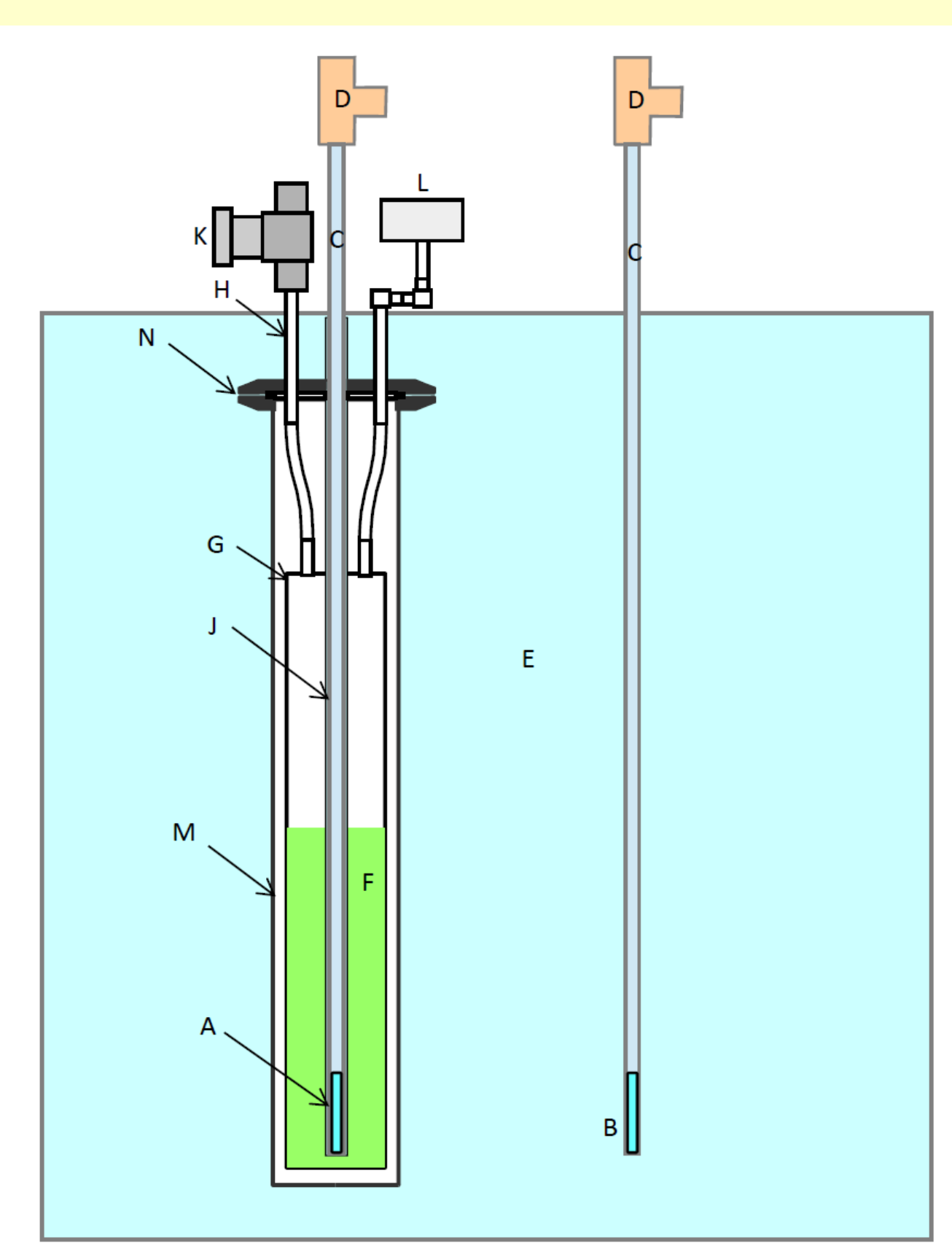
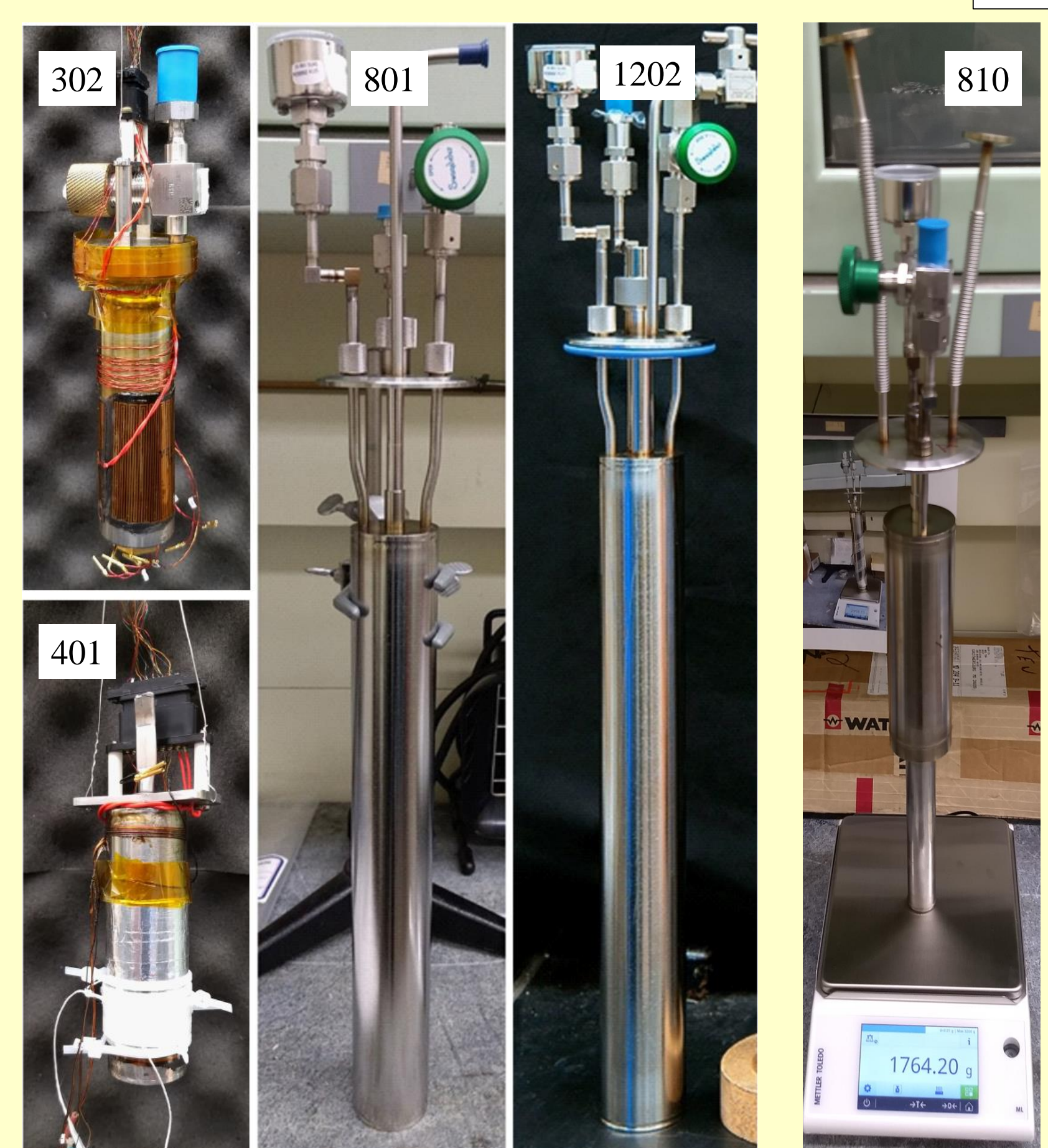
Five SF₆ TP cells and one CO₂ TP cell are in this study. Four cells (SF₆ serial numbers 1202, 801, 803 and CO₂ cell 810) are larger immersion types and two others (SF₆ Cell serial numbers 302 and 401) are smaller 'adiabatic' type cells. The cell design parameters are summarized in the below Table. The cells are shown in the photographs below.

Cell	Volume V_{cell} cm ³	SF ₆ /CO ₂ mass, m_{cell} g	Density ρ_{cell} g·cm ⁻³	CDR ρ_{cell}/ρ_{cr}	Effective Immersion cm	Enthapy ΔH_f kJ
s/n						
302	42	12.9	0.307	0.41	N/A	0.47
401	14	9.1	0.650	0.87	N/A	0.33
801	580	590	1.017	1.37	16	21.3
1202	550	467	0.849	1.14	13	17.0
803	537	573.3	1.068	1.44	18.3	20.7
810	352	68.55	0.195	0.42	16.8	13.5

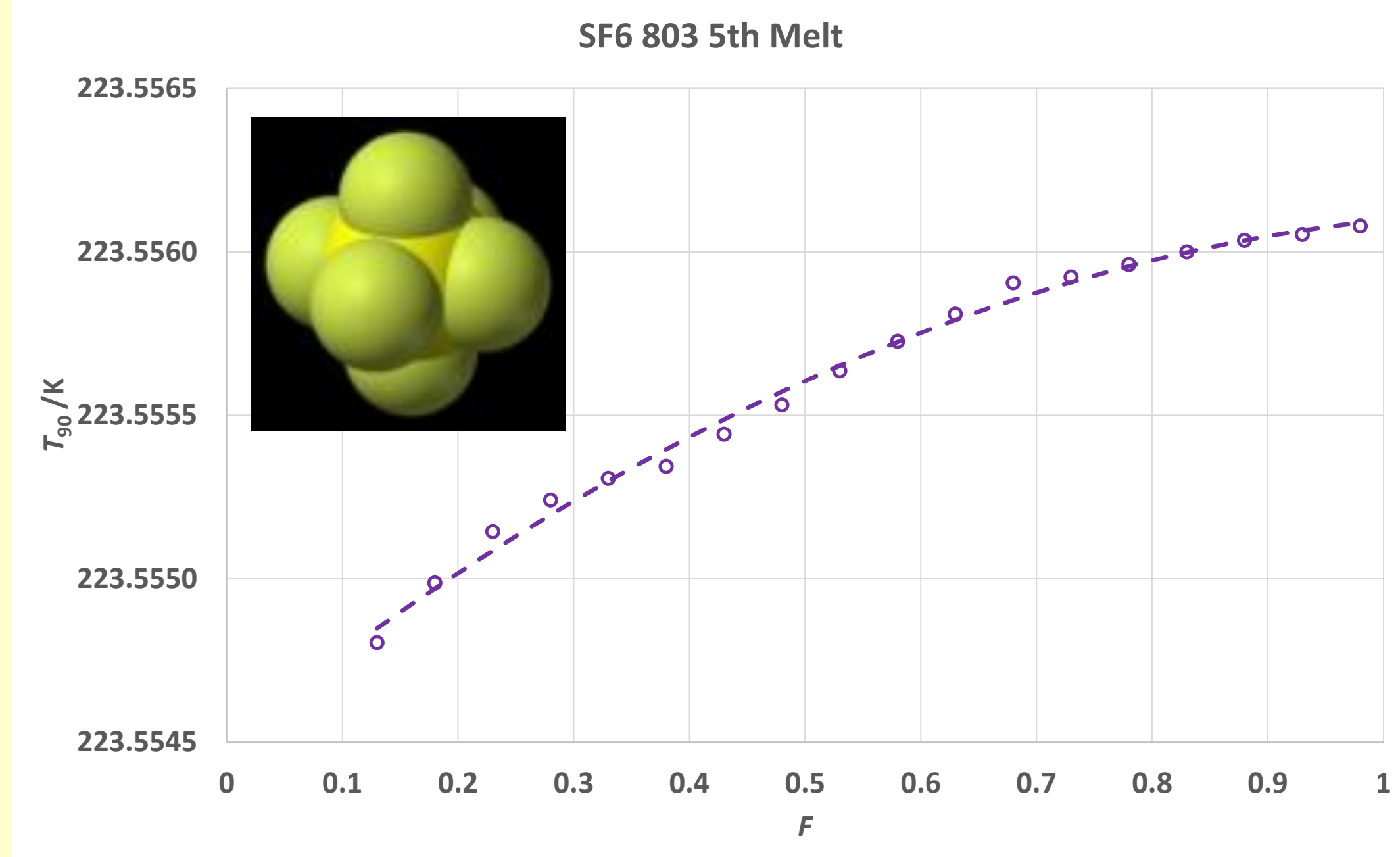
Cell 803 is similar to cell 801, but with the addition of an internal set of 30 copper baffles (shown below) to equalize radial gradients and support the solid SF₆ mantle.



Below: Experimental setup for immersion-type cells : capsule-type SPRTs to measure the cell (A) and bath (B) temperatures; (C) borosilicate glass adapter tubes; (D) nylon gas-tight fittings; (E) 50 l ethanol bath volume; (F) condensed (shown in green) portion of SF₆ (G) pressure wall of cell volume; (H) SF₆ fill line; (J) central thermowell; (K) bellows valve; (L) pressure gauge; (M) vacuum chamber; and (N) ISO-63 vacuum flange with aluminum gasket.



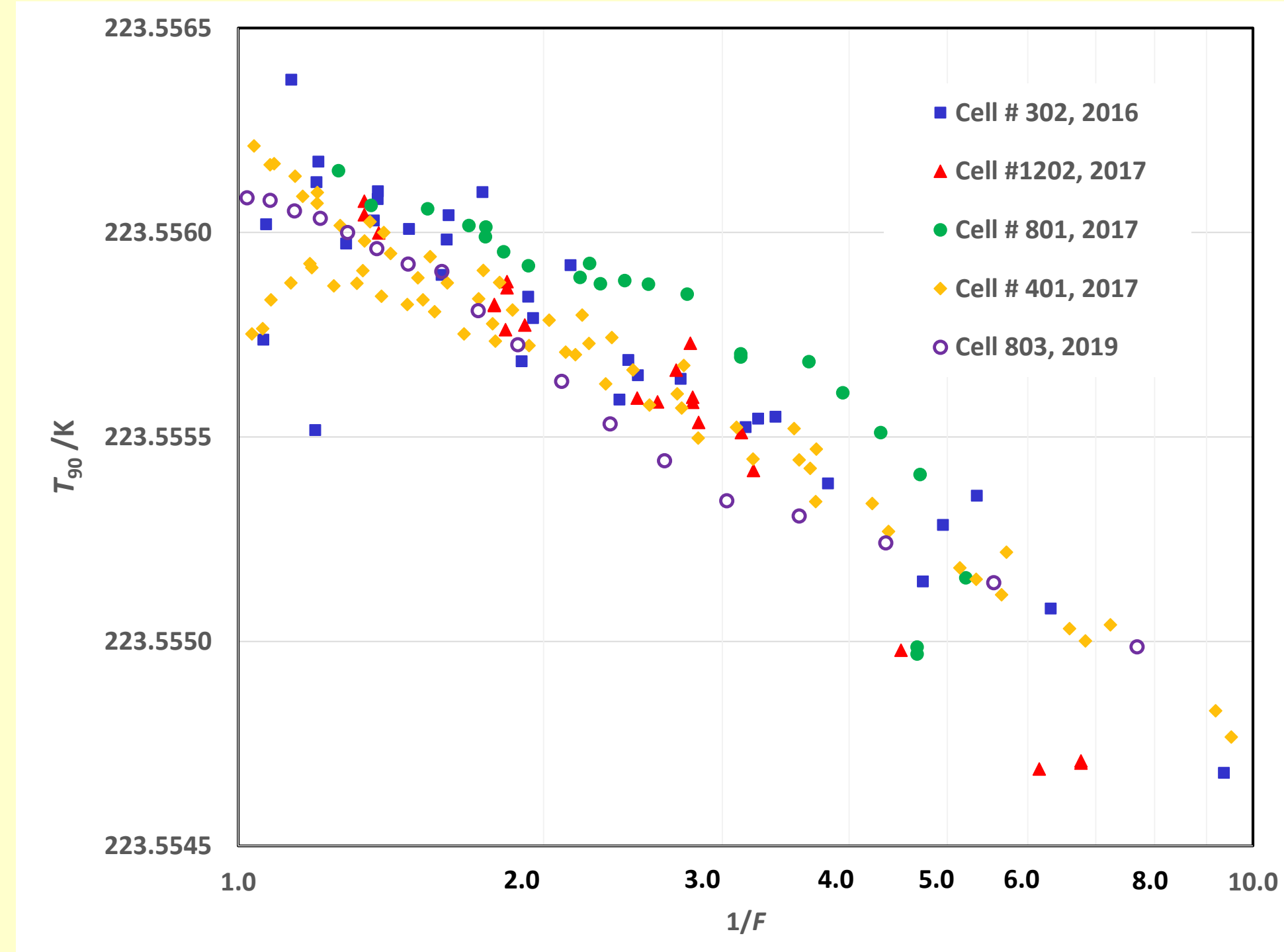
Results using SF₆



SF₆ Cell 803, 2019
Direct Heating Only
Pulse Melting
5 % ΔF / heat pulse
2 pulses / day
6 hrs. ON, 6 hrs. OFF
Stable over all F values

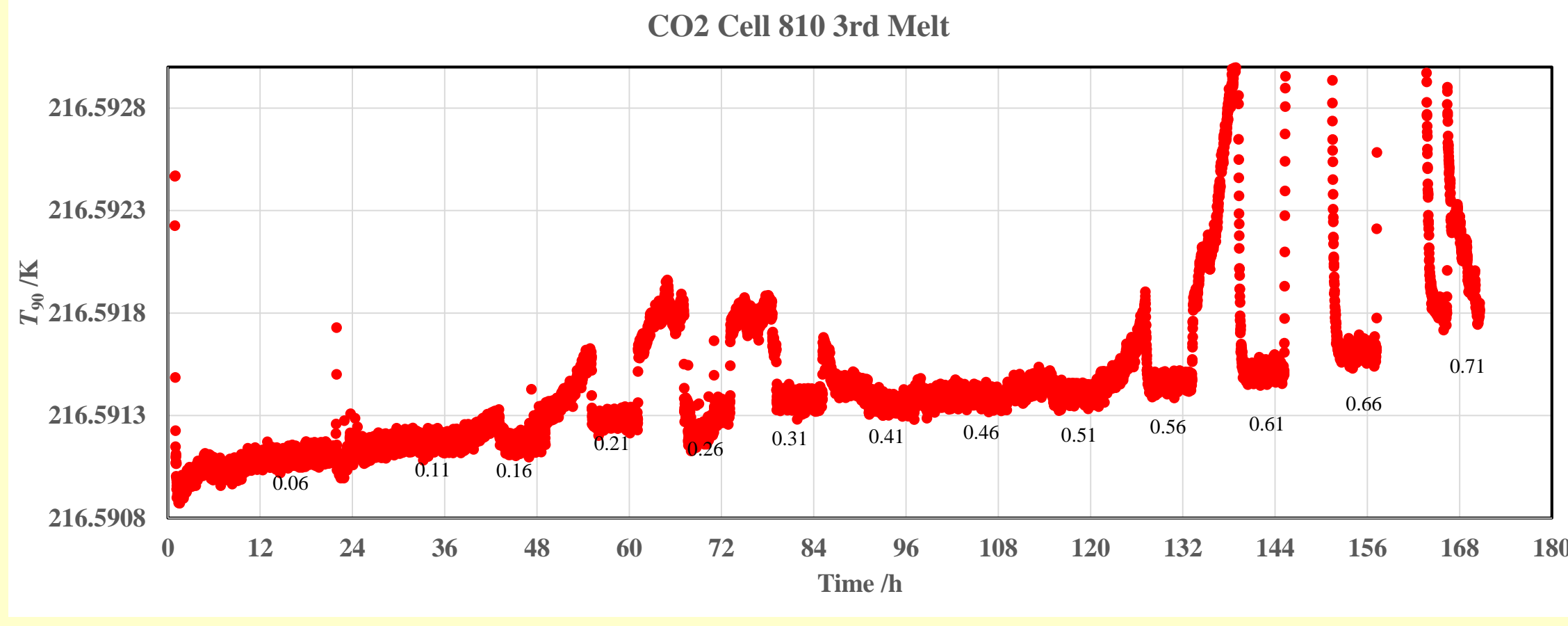
SF₆ Source Cylinder
Concorde Specialty Gas†
Report of Analysis:
CO₂ - ND*
SO₂ - ND*
N₂ + O₂* = 3.3 ppm
H₂O < 1 ppm*
CF₄ = 0.3 ppm
*Removed by in-line Getter

Complied SF₆ TP Realizations
Cells 302/401: Adiabatic 2016-17
Pulsed Direct Heating
Kapton Thermofoil Heaters
500 mW, 1000 mW
Cells 801/1202: 2016-17
Pulsed Indirect Heating
Gas-coupled to Bath Heat
<= 10 mW
Cell 803: 2019
Pulsed Direct Heating
Kapton Thermofoil Heaters
48 mW



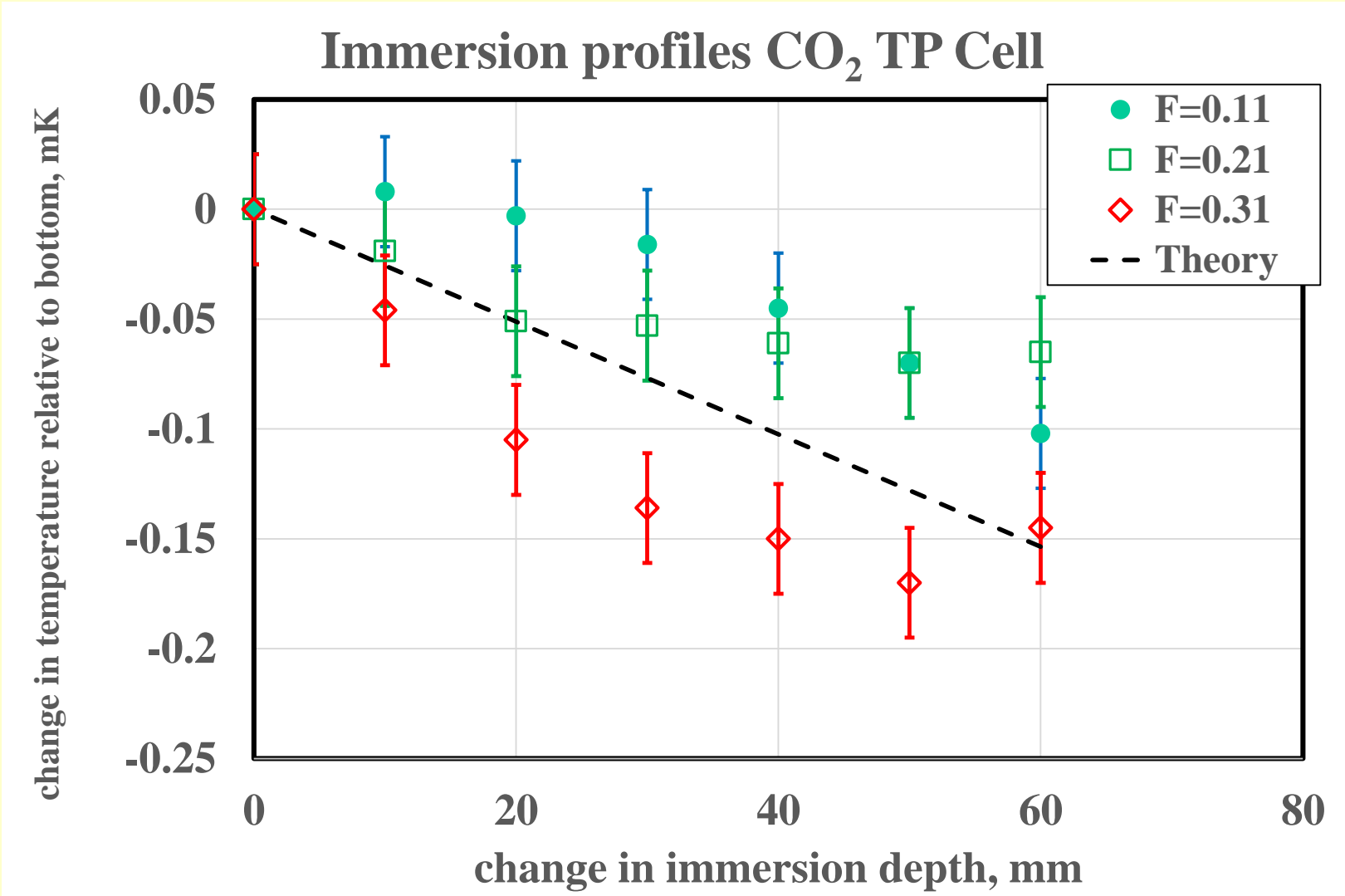
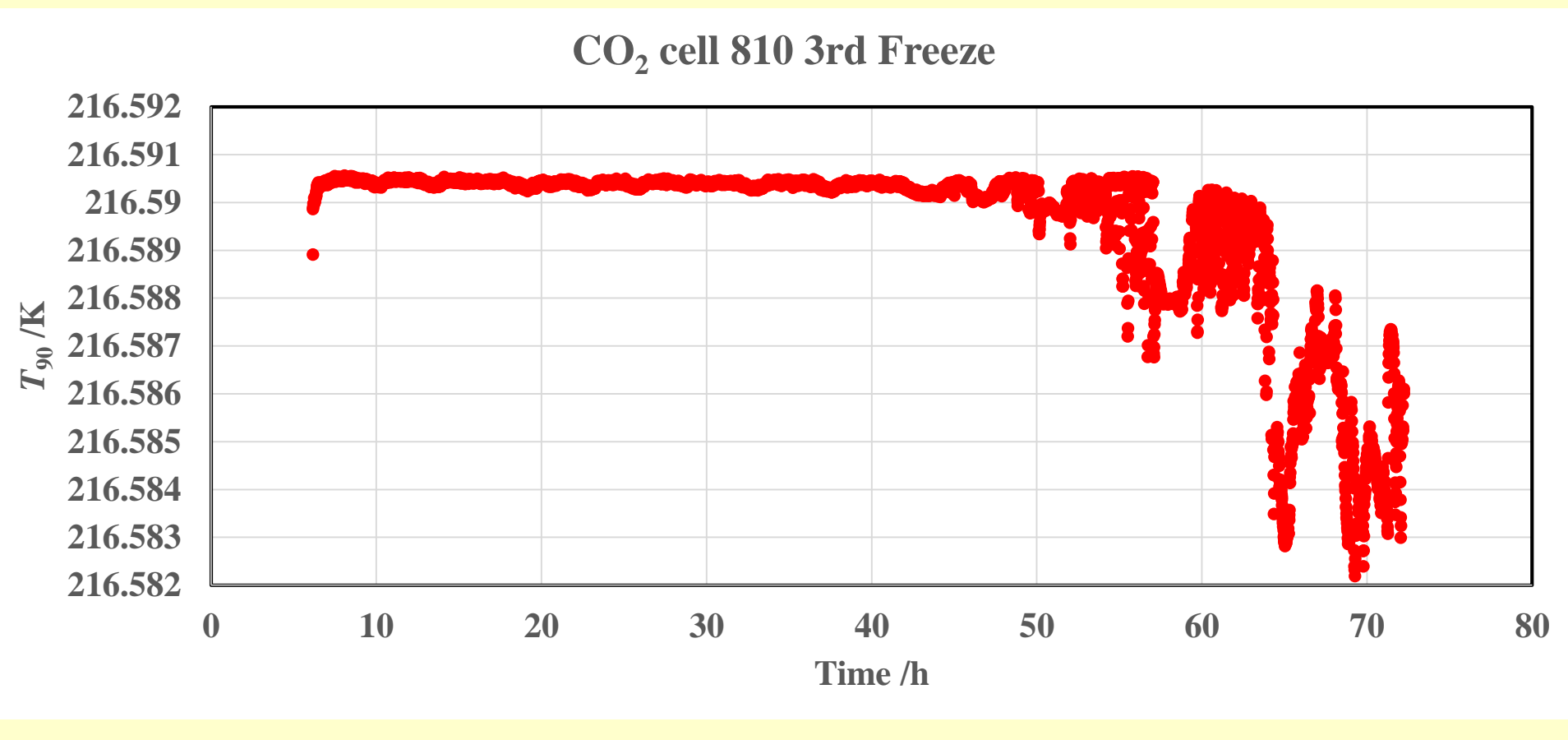
† Certain commercial equipment, instruments, or materials are identified in this poster in order to specify the experimental procedure adequately. Such identification is not intended to imply recommendation or endorsement by the National Institute of Standards and Technology, nor is it intended to imply that the materials or equipment identified are necessarily the best available for the purpose.

Results using CO₂



CO₂ Cell 810, 2019
Direct Heating
Pulse Melting
5 % ΔF / heat pulse
Stable Plateau $F < 0.56$

CO₂ Source Cylinder
NIST Gas Metrology Group
Measured:
Ar: 0.30 ± 0.03 ppm
N₂: 1.05 ± 0.03 ppm
Specification:
CO < 0.2 ppm*
H₂ < 1 ppm*
N₂ < 5 ppm
O₂ < 2 ppm*
H₂O < 2 ppm*
THC (as CH₄)* < 4 ppm
*Removed by in-line Getter



Immersion Profiles, CO₂ Cell 810, 2019

- Some distortions observed
- 10 mm steps downward
- Predicted from Clapeyron Equation :
 - 0.0247 mK/cm
 - assumes melting line: 4.69 MPa/K

$$\left. \frac{dT}{dz} \right|_{T=T_p} = \frac{\rho g}{\left. \frac{dp}{dT} \right|_{T=T_p}}$$

Summary

- Results using SF₆ :**
 - average of four cells (#302, 1202, 801, 401) 2017 [1]
 - 223.55587(33) K, $F=0.5$; 223.55607(33) K, $F=1$
 - cell #803 (immersion type), 2019, preliminary
 - 223.5556(5) K, $F=0.5$; 223.55608(50) K, $F=1$
 - Pressure head correction: 0.116(16) mK/cm [1]; Clapeyron Eqn. : 0.116 mK/cm
- Results using CO₂, Cell 810, 2019 preliminary:**
 - 216.5910(5) K, $F=0.5$
- References**
 - [1] W. Tew and K. Quelhas, *J Res NIST*, Vol 123, Art. No. 12013 (2018) <https://doi.org/10.6028/jres.123.013>